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(57) Abstract

Compounds of formula (I), wherein the substituents R1 to R6 and n are as defined in claim 1, and the salts and stereoisomers of the compounds of formula (I) have good pre- and post-emergence selective herbicidal properties. The preparation of those compounds and their use as herbicidal active ingredients are described.

$$R_{5} \xrightarrow{R_{4}} R_{1} \xrightarrow{R_{1}} S(O)_{n} R_{2}$$

$$R_{5} \xrightarrow{R_{6}} R_{3}$$

$$(I)$$

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Novel herbicides

The present invention relates to novel, herbicidally active phenylpyrazole derivatives, to processes for their preparation, to compositions comprising those compounds, and to their use in controlling weeds, especially in crops of useful plants or in the inhibition of plant growth.

Pyrazole compounds having herbicidal action are known and are disclosed, for example, in EP-A-0 361 114, JP-A-03 093 774, JP-A-02 300 173 and JP-A-03 163 063.

Novel phenylpyrazole derivatives having herbicidal and growth-inhibiting properties have now been found.

The present invention therefore relates to compounds of formula I

$$R_{4} \qquad R_{1} \qquad S(0) \qquad R_{2} \qquad (I),$$

$$R_{6} \qquad R_{3} \qquad (I),$$

wherein

R₁ is C₁-C₄alkyl;

R₂ is C_1 - C_4 alkyl, C_1 - C_4 haloalkyl, C_3 - or C_4 -alkenyl, C_3 - or C_4 -haloalkenyl or C_3 - or C_4 -alkynyl;

n is 0, 1 or 2;

R₃ is hydrogen, C₁-C₄alkyl, C₁-C₄haloalkyl, C₃- or C₄-alkenyl, C₃- or C₄-haloalkenyl, C₃- or C₄-alkynyl, -CH₂COOH, -CH₂COO-C₁-C₄alkyl or -CH₂CN;

R₄ is hydrogen, fluorine, chlorine or bromine;

R₅ is hydrogen, halogen, methyl, trifluoromethyl, cyano, nitro, amino or C₁-C₄halo-alkoxy;

R₆ is hydrogen, halogen, cyano, NHR₁₀, NR₁₀R₁₁ or SO₂Cl;

 R_{10} and R_{11} are each independently of the other C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_3 - C_6 cycloalkyl, C_1 - C_8 haloalkyl, C_3 - C_8 haloalkenyl, C_1 - C_4 alkylcarbonyl, C_1 - C_4 haloalkylcarbonyl, C_1 - C_4 alkylsulfonyl, C_1 - C_4 haloalkylsulfonyl, benzoyl, benzoyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen,

benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; or

 R_6 is OR_{20} ;

R₂₀ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl,

$$-CH_2-CH-CH_2$$
, C_3-C_8 haloalkenyl, C_3-C_6 cycloalkyl, $-CH_2-CH_2$,

 $\begin{array}{l} C_1\text{-}C_4\text{alkoxy-}C_1\text{-}C_4\text{alkyl}, C_1\text{-}C_4\text{alkylamino-}C_1\text{-}C_4\text{alkyl}, \text{di-}C_1\text{-}C_4\text{alkylamino-}\\ C_1\text{-}C_4\text{alkyl}, C_1\text{-}C_4\text{alkoxy-}C_1\text{-}C_4\text{alkoxy-}C_1\text{-}C_4\text{alkyl}, C_1\text{-}C_4\text{alkylthio-}C_1\text{-}C_4\text{alkyl},\\ \text{phenyl, benzyl, pyridyl, pyrimidinyl, pyrazinyl, pyridazinyl, those aromatic and heteroaromatic rings being unsubstituted or mono- to tri-substituted by $C_1\text{-}C_4\text{alkyl},$C_1\text{-}C_4\text{haloalkyl}$ or by halogen; or \\ \end{array}$

 R_{20} is C_1 - C_8 alkyl- $COXR_{21}$ or $CH(C_6H_5)COXR_{21}$;

X is oxygen, sulfur or NR₂₂;

 $R_{21} \quad \text{is hydrogen, C_1-C_8alkyl, C_3-C_8alkenyl, C_3-C_8alkynyl, C_1-C_8haloalkyl, C_3-C_6cycloalkyl, C_1-C_4alkoxy-C_1-C_4alkyl, C_1-C_4alkylthio-C_1-C_4alkyl, phenyl, phenyl mono- to tri-substituted by C_1-C_4alkyl, C_1-C_4alkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1-C_4alkyl or by halogen; and C_1-C_4alkyl or by halogen;$

R₂₂ is hydrogen, C₁-C₈alkyl or C₃-C₈alkenyl; or

 R_6 is $S(O)_m R_{30}$;

m is 0, 1 or 2;

 $R_{30} \quad \text{is hydrogen, C_1-C_8alkyl, C_3-C_8alkenyl, C_3-C_8alkynyl, C_1-C_8haloalkyl, C_3-C_8haloalkyl, C_1-C_4alkyl, C_1-C_4alkyl, C_1-C_4alkyl, C_1-C_4alkyl, C_1-C_4alkyl, C_1-C_4alkyl, C_1-C_4alkyl or by halogen, benzyl, benzyl mono- to tri-substituted by C_1-C_4alkyl, C_1-C_4haloalkyl or by halogen, or C_1-C_4alkyl-$COVR_{31}$;}$

V is oxygen, sulfur or NR₃₂;

R₃₁ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl monoto tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl monoto tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen; and

R₃₂ is hydrogen, C₁-C₈alkyl or C₃-C₈alkenyl; or

 R_6 is COR_{40} ;

R₄₀ is hydrogen, chlorine, C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_3 - C_8 haloalkenyl, C_3 - C_6 cycloalkyl, C_1 - C_4 alkoxy- C_1 - C_4 alkyl, C_1 - C_4 alkyl, phenyl, phenyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl

or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; or

 R_6 is COYR₅₀;

is oxygen, sulfur, NR₅₁ or NOR₅₄; Y

is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl,

$$-CH_2-CH-CH_2$$
, C_3-C_8 haloalkenyl, C_3-C_6 cycloalkyl, $-CH_2-CH_2$,

 C_1 - C_4 alkoxy- C_1 - C_4 alkyl, C_1 - C_4 alkylthio- C_1 - C_4 alkyl, phenyl, phenyl mono- to trisubstituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, C₁-C₄alkyl-COZR₅₂, C_3 - C_6 cycloalkyl- $COZR_{52}$, C_1 - C_4 alkyl-CO- C_1 - C_4 alkyl or C_1 - C_4 cyanoalkyl;

 \mathbf{Z} is oxygen, sulfur, NR₅₃ or NOR₅₅;

is hydrogen, C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, R_{52}

 $C_1-C_4\\ alkoxy-C_1-C_4\\ alkyl,\ C_1-C_4\\ alkoxy-C_1-C_4\\ alkoxy-C_1-C_4\\ alkyl,\ C_1-C_4\\ alkylthio-1\\ alkyl-C_4\\ alky$ C_1 - C_4 alkyl, phenyl, phenyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen;

 R_{51} and R_{53} are each independently of the other C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_1 - C_4 alkylcarbonyl, C_1 - C_4 haloalkylcarbonyl, C_1 - C_4 alkylsulfonyl, C_1 - C_4 haloalkylsulfonyl, benzoyl, benzoyl mono- to tri-substituted by C_1 - C_4 alkyl, C₁-C₄haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen;

 R_{54} and R_{55} are each independently of the other C_1 - C_4 alkyl; or

$$R_6$$
 is $-SO_2 N$ R_{56} ;

is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl or C₃-C₈alkynyl;

is hydrogen, C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl or C_1 - C_4 alkylcarbonyl; or

is C_1 - C_8 alkyl-B, C_1 - C_8 haloalkyl-B, C_2 - C_8 alkenyl-B, C_2 - C_8 alkynyl-B, C_2 - C_8 haloalkenyl-B, C_1 - C_4 alkoxy- C_1 - C_4 alkyl-B or C_1 - C_4 alkylthio- C_1 - C_4 alkyl-B; and

is hydrogen, -COZR₅₂, cyano or C₁-C₄alkyl-C(O)-,

and the salts and stereoisomers of the compound of formula I.

The present invention relates also to compounds of formula I wherein

 R_1 is C_1 - C_4 alkyl;

R₂ is C₁-C₄alkyl, C₁-C₄haloalkyl, C₃- or C₄-alkenyl or C₃- or C₄-alkynyl;

n is 0, 1 or 2:

R₃ is hydrogen, C₁-C₄alkyl, C₁-C₄haloalkyl, C₃- or C₄-alkenyl or C₃- or C₄-alkynyl;

R₄ is hydrogen, fluorine or chlorine;

 R_5 is hydrogen, halogen, methyl, trifluoromethyl, cyano, nitro, amino or C_1 - C_4 haloalkoxy;

R₆ is hydrogen, halogen, C₁-C₅alkyl, C₁-C₅haloalkyl, C₂-C₅alkenyl, C₂-C₅haloalkenyl, C₂-C₅haloalkynyl, cyano, NHR₁₀ or NR₁₀R₁₁;

 R_{10} and R_{11} are each independently of the other C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_1 - C_4 alkylcarbonyl, C_1 - C_4 haloalkylcarbonyl, C_1 - C_4 alkylsulfonyl, benzoyl mono- to tri-substituted by C_1 - C_4 alkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; or

 R_6 is OR_{20} ;

R₂₀ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, benzyl, pyridyl, pyrimidinyl, pyrazinyl, pyridazinyl, those aromatic and heteroaromatic rings being unsubstituted or mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, or C₁-C₈alkyl-COXR₂₁;

X is oxygen, sulfur or NR₂₂;

R₂₁ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl monoto tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, or benzyl monoto tri-substituted by C₁-C₄alkyl or by halogen; and

R₂₂ is hydrogen, C₁-C₈alkyl or C₃-C₈alkenyl; or

 R_6 is $S(O)_m R_{30}$;

m is 0, 1 or 2;

R₃₀ is hydrogen, chlorine, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, or C₁-C₄alkyl-COVR₃₁;

V is oxygen, sulfur or NR₃₂;

is hydrogen, C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_3 - C_6 cycloalkyl, C_1 - C_4 alkoxy- C_1 - C_4 alkyl, C_1 - C_4 alkylthio- C_1 - C_4 alkyl, phenyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen; and

is hydrogen, C₁-C₈alkyl or C₃-C₈alkenyl; or R_{32}

 R_6 is COR₄₀;

is hydrogen, C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_3 - C_6 cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen; or

is COYR₅₀; R_6

is oxygen, sulfur, NR₅₁ or NOR₅₄; Y

is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C_1 - C_4 alkoxy- C_1 - C_4 alkyl, C_1 - C_4 alkylthio- C_1 - C_4 alkyl, phenyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, benzyl monoto tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, or C_1 - C_4 alkyl-COZR₅₂;

Z is oxygen, sulfur, NR₅₃ or NOR₅₅;

is hydrogen, C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_3 - C_6 cycloalkyl, C_1 - C_4 alkoxy- C_1 - C_4 alkyl, C_1 - C_4 alkylthio- C_1 - C_4 alkyl, phenyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen;

 R_{51} and R_{53} are each independently of the other C_1 - C_8 alkyl, C_3 - C_8 alkynyl, $C_1-C_8 haloalkyl,\ C_1-C_4 alkylcarbonyl,\ C_1-C_4 haloalkylcarbonyl,\ C_1-C_4 alkylsulfonyl,$ C_1 - C_4 haloalkylsulfonyl, benzoyl, benzoyl mono- to tri-substituted by C_1 - C_4 alkyl, C₁-C₄haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen; and

 R_{54} and R_{55} are each independently of the other C_I - C_4 alkyl; or

is C_1 - C_4 alkyl $COZR_{52}$, C_1 - C_4 haloalkyl $COZR_{52}$, C_2 - C_4 alkenyl $COZR_{52}$, C₂-C₄alkynylCOZR₅₂ or C₂-C₄haloalkenylCOZR₅₂, and the salts and stereoisomers of the compound of formula I.

The present invention relates also to compounds of formula I wherein

is C₁-C₄alkyl; R_1

is C₁-C₄alkyl, C₁-C₄haloalkyl, C₃- or C₄alkenyl or C₃- or C₄alkynyl;

n is 0, 1 or 2;

- R₃ is C₁-C₄alkyl, C₁-C₄haloalkyl, C₃- or C₄-alkenyl or C₃- or C₄-alkynyl;
- R₄ is hydrogen, fluorine or chlorine;
- R₅ is hydrogen, halogen, methyl, trifluoromethyl, cyano, nitro, amino or C₁-C₄halo-alkoxy;
- R₆ is hydrogen, halogen, cyano, NHR₁₀ or NR₁₀R₁₁;
- R_{10} and R_{11} are each independently of the other C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_1 - C_4 alkylcarbonyl, C_1 - C_4 haloalkylcarbonyl, C_1 - C_4 alkylsulfonyl, benzoyl mono- to tri-substituted by C_1 - C_4 alkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; or
- R_6 is OR_{20} ;
- R₂₀ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, benzyl, pyridyl, pyrimidinyl, pyrazinyl, pyridazinyl, those aromatic and heteroaromatic rings being unsubstituted or mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, or C₁-C₈alkyl-COXR₂₁;
- X is oxygen, sulfur or NR₂₂;
- R₂₁ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl monoto tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, benzyl monoto tri-substituted by C₁-C₄alkyl or by halogen; and
- R₂₂ is hydrogen, C₁-C₈alkyl or C₃-C₈alkenyl; or
- R_6 is $S(O)_m R_{30}$;
- m is 0, 1 or 2;
- R₃₀ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl monotorisubstituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, benzyl monoto tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, or C₁-C₄alkyl-COVR₃₁;
- V is oxygen, sulfur or NR₃₂;
- is hydrogen, C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_3 - C_6 cycloalkyl, C_1 - C_4 alkoxy- C_1 - C_4 alkyl, C_1 - C_4 alkyl, phenyl, phenyl monoto tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, benzyl monoto tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; and
- R₃₂ is hydrogen, C₁-C₈alkyl or C₃-C₈alkenyl; or
- R_6 is COR_{40} ;

R₄₀ is hydrogen, C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_3 - C_6 cycloalkyl, C_1 - C_4 alkoxy- C_1 - C_4 alkyl, C_1 - C_4 alkylthio- C_1 - C_4 alkyl, phenyl, phenyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; or

 R_6 is $COYR_{50}$;

Y is oxygen, sulfur, NR₅₁ or NOR₅₄;

 $R_{50} \quad \text{is hydrogen, C_1-C_8alkyl, C_3-C_8alkenyl, C_3-C_8alkynyl, C_1-C_8haloalkyl, C_3-C_6cycloalkyl, C_1-C_4alkoxy-C_1-C_4alkyl, C_1-C_4alkylthio-C_1-C_4alkyl, phenyl, phenyl monoto tri-substituted by C_1-C_4alkyl, C_1-C_4haloalkyl or by halogen, benzyl, benzyl monoto tri-substituted by C_1-C_4alkyl, C_1-C_4haloalkyl or by halogen, or C_1-C_4alkyl-$COZR_{52}$;}$

Z is oxygen, sulfur, NR₅₃ or NOR₅₅;

R₅₂ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl monoto tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl monoto tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen;

 R_{51} and R_{53} are each independently of the other C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_1 - C_4 alkylcarbonyl, C_1 - C_4 haloalkylcarbonyl, C_1 - C_4 alkylsulfonyl, benzoyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; and

R₅₄ and R₅₅ are each independently of the other C₁-C₄alkyl; or

 R_6 is C_1 - C_4 alkylCOZR₅₂, C_1 - C_4 haloalkylCOZR₅₂, C_2 - C_4 alkenylCOZR₅₂, C_2 - C_4 alkynylCOZR₅₂ or C_2 - C_4 haloalkenylCOZR₅₂, and the salts and stereoisomers of the compound of formula I.

In the above definitions, unless otherwise indicated halogen is to be understood as being fluorine, chlorine, bromine and iodine, preferably fluorine, chlorine and bromine.

The alkyl, alkenyl and alkynyl groups may be straight-chained or branched, and this applies also to the alkyl, alkenyl and alkynyl moiety of alkylcarbonyl, haloalkyl, haloalkyl, haloalkylcarbonyl, haloalkylphenyl, alkoxyalkyl, alkoxyalkoxyalkyl, alkylthio, alkenylthio, alkynylthio, alkylsulfinyl, alkylsulfonyl, haloalkylsulfonyl, alkenylsulfonyl, alkylamino, dialkylamino, alkylaminocarbonyl-alkyl, haloalkylamino, di(haloalkyl)amino, alkoxyalkylamino, carboxyalkyl, alkylthio-alkyl, alkylthio-alkoxycarbonyl, alkylthiocarbonyl-alkyl, alkenylthiocarbonyl, alkynylthiocarbonyl,

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haloalkoxycarbonyl-alkyl, alkylcarbonyl, alkenyloxycarbonyl, alkynyloxycarbonyl and alkoxycarbonyl-alkyl groups.

Examples of alkyl groups that may be mentioned are methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, sec-butyl, tert-butyl, and the various isomeric pentyl, hexyl, heptyl and octyl radicals, preferably alkyl groups having from 1 to 4 carbon atoms.

Examples of alkenyls that may be mentioned are vinyl, allyl, methallyl, 1-methylvinyl, but-2-en-1-yl, pentenyl, 2-hexenyl, 3-heptenyl and 4-octenyl, preferably alkenyl radicals having a chain length of from 3 to 5 carbon atoms.

Example of alkynyls that may be mentioned are ethynyl, propargyl, 1-methylpropargyl, 3-butynyl, but-2-yn-1-yl, 2-methylbutyn-2-yl, but-3-yn-2-yl, 1-pentynyl, pent-4-yn-1-yl and 2-hexynyl, preferably alkynyl radicals having a chain length of from 2 to 4 carbon atoms.

Suitable as haloalkyl are alkyl groups mono- or poly-substituted, especially mono- to trisubstituted, by halogen, with halogen being in particular iodine and especially fluorine, chlorine and bromine, for example fluoromethyl, difluoromethyl, trifluoromethyl, chloromethyl, dichloromethyl, trichloromethyl, 2,2,2-trifluoroethyl, 2-fluoroethyl, 2-chloroethyl and 2,2,2-trichloroethyl.

Suitable as haloalkenyl are alkenyl groups mono- or poly-substituted by halogen, with halogen being in particular bromine, iodine and especially fluorine and chlorine, for example 3-fluoropropenyl, 3-chloropropenyl, 3-bromopropenyl, 2,3,3-trifluoropropenyl, 2,3,3-trichloropropenyl, 4,4,4-trifluoro-but-2-en-1-yl and 4,4,4-trichloro-but-2-en-1-yl. Of the C_2 - C_5 alkenyl radicals mono-, di- or tri-substituted by halogen, preference is given to those having a chain length of 3 or 4 carbon atoms. The alkenyl groups may be substituted with halogen at saturated or unsaturated carbon atoms.

Haloalkenyl in the definition of R_6 as haloalkenylCOZR₅₂ is, for example, 1,2-dichloroethenyl or 1,2-dibromoethenyl.

Suitable as haloalkynyl are, for example, alkynyl groups mono- or poly-substituted by halogen, with halogen being bromine, iodine and especially fluorine and chlorine, for example 3-fluoropropynyl, 3-chloropropynyl, 3-bromopropynyl, 3,3,3-trifluoropropynyl

and 4,4,4-trifluoro-but-2-yn-1-yl.

Carboxyalkyl is, for example, carboxymethyl, carboxyethyl, carboxyeth-1-yl and carboxy-propyl.

Alkoxyalkyl is, for example, methoxymethyl, ethoxymethyl, propoxymethyl, methoxyethyl, ethoxyethyl, propoxyethyl, butoxyethyl, methoxypropyl, ethoxypropyl or propoxypropyl.

Alkoxy is, for example, methoxy, ethoxy, n-propoxy, isopropoxy, n-butoxy, isobutoxy, sec-butoxy and tert-butoxy.

Alkenyloxy is, for example, allyloxy, methallyloxy and but-2-en-1-yloxy.

Alkynyloxy is, for example, propargyloxy and 1-methylpropargyloxy.

Alkoxycarbonyl is, for example, methoxycarbonyl, ethoxycarbonyl, n-propoxycarbonyl, isopropoxycarbonyl and n-butoxycarbonyl, preferably methoxycarbonyl and ethoxycarbonyl.

Alkenyloxycarbonyl is, for example, allyloxycarbonyl, methallyloxycarbonyl, but-2-en-1-yl-oxycarbonyl, pentenyloxycarbonyl, 2-hexenyloxycarbonyl and 3-heptenyloxy-carbonyl.

Alkynyloxycarbonyl is, for example, propargyloxycarbonyl, 3-butynyloxycarbonyl, but-2-yn-1-yl-oxycarbonyl and 2-methylbutyn-2-yl-oxycarbonyl.

Alkylamino is, for example, methylamino, ethylamino and the isomeric propylamino and butylamino.

Dialkylamino is, for example, dimethylamino, diethylamino and the isomeric dipropylamino and dibutylamino.

Alkenylamino is, for example, allylamino, methallylamino and but-2-en-1-yl-amino.

Alkynylamino is, for example, propargylamino and 1-methylpropargylamino.

Cycloalkyl radicals that come into consideration as substituents are, for example, cyclopropyl, cyclobutyl, cyclopentyl and cyclohexyl.

Alkoxyalkoxycarbonyl is, for example, methoxymethoxycarbonyl, ethoxymethoxycarbonyl, ethoxyethoxycarbonyl, propoxymethoxycarbonyl, propoxyethoxycarbonyl, propoxypropoxycarbonyl, butoxyethoxycarbonyl and butoxybutoxycarbonyl.

Haloalkoxy is, for example, fluoromethoxy, difluoromethoxy, trifluoromethoxy, 2,2,2-trifluoroethoxy, 1,1,2,2-tetrafluoroethoxy, 2-fluoroethoxy, 2-chloroethoxy and 2,2,2-trichloroethoxy.

Haloalkylamino is, for example, chloroethylamino, trifluoroethylamino and 3-chloropropylamino.

Di(haloalkyl)amino is, for example, di(chloroethyl)amino.

Alkylthioalkyl is, for example, methylthioethyl, ethylthioethyl, methylthiopropyl and ethylthiopropyl.

Alkenylthiocarbonyl is, for example, allylthiocarbonyl, methallylthiocarbonyl, but-2-en-1-yl-thiocarbonyl, pentenylthiocarbonyl and 2-hexenylthiocarbonyl.

Alkynylthiocarbonyl is, for example, propargylthiocarbonyl, 1-methylpropargylthiocarbonyl and but-2-yn-1-yl-thiocarbonyl.

Phenyl, benzyl or benzoyl as part of a substituent such as phenoxy, phenoxycarbonyl, phenoxycarbonylalkyl, benzoylamino or benzylamino is unsubstituted or substituted. The substituents may then be in the ortho-, meta- or para-position. Substituents are, for example, C_1 - C_4 alkyl, halogen, C_1 - C_4 haloalkyl, cyano, nitro, hydroxy, C_1 - C_4 alkoxy, C_1 - C_4 alkylamino, C_1 - C_4 alkylamino, di- C_1 - C_4 alkylamino, carboxyl, C_1 - C_4 alkoxy-carbonyl, carbamoyl, C_1 - C_4 alkylaminocarbonyl or di- C_1 - C_4 alkylaminocarbonyl.

Corresponding meanings can be assigned also to the substituents in combined definitions, for example cycloalkyl-oxy, cycloalkyl-thio, cycloalkylcarbonyl, cycloalkyl-oxycarbonyl-alkyl, phenylalkyl, phenylalkyl, alkoxycarbonylalkyl, alkenyloxycarbonylalkyl, alkynyl-

oxycarbonyl-alkyl, haloalkoxycarbonyl-alkyl, alkylaminocarbonyl-alkyl, alkenylaminocarbonyl-alkyl, alkynylaminocarbonyl-alkyl, dialkylaminocarbonyl-alkyl, alkoxyalkylamino, alkoxyalkylaminocarbonyl, alkoxyalkoxycarbonyl, alkylaminocarbonyl-alkyl, dialkylaminocarbonyl-alkyl, dialkylaminocarbonyl-alkyl, alkylaminocarbonyl, alkylaminocarbonyl-alkyl, alkylaminocarbonyl, alkylaminocarbon

Salts of the compounds of formula I with acidic hydrogen, especially the derivatives with carboxylic acid groups (for example carboxy-substituted alkyl and phenyl groups) are, for example, alkali metal salts, for example sodium and potassium salts; alkaline earth metal salts, for example calcium and magnesium salts; ammonium salts, that is to say unsubstituted ammonium salts and mono- or poly-substituted ammonium salts, for example triethylammonium and methylammonium salts; or salts with other organic bases.

Of the alkali metal and alkaline earth metal hydroxides as salt-forming agents, special mention should be made, for example, of the hydroxides of lithium, sodium, potassium, magnesium or calcium, but especially those of sodium or potassium.

Examples of amines suitable for ammonium salt formation are both ammonia and primary, secondary and tertiary C_1 - C_{18} alkylamines, C_1 - C_4 hydroxyalkylamines and C_2 - C_4 alkoxyalkylamines, for example methylamine, ethylamine, n-propylamine, isopropylamine, the four isomeric butylamines, n-amylamine, isoamylamine, hexylamine, heptylamine, octylamine, nonylamine, decylamine, pentadecylamine, hexadecylamine, heptadecylamine, octadecylamine, methyl-ethylamine, methyl-isopropylamine, methyl-hexylamine, methyl-nonylamine, methyl-pentadecylamine, methyl-octadecylamine, ethyl-butylamine, ethyl-heptylamine, ethyl-octylamine, hexyl-heptylamine, hexyl-octylamine, dimethylamine, diethylamine, di-n-propylamine, diisopropylamine, di-n-butylamine, di-n-amylamine, diisoamylamine, dihexylamine, diheptylamine, dioctylamine, ethanolamine, n-propanolamine, isopropanolamine, N,N-diethanolamine, N-ethylpropanolamine, N-butylethanolamine, allylamine, n-butenyl-2-amine, n-pentenyl-2-amine, 2,3-dimethylbutenyl-2-amine, dibutenyl-2-amine, n-hexenyl-2-amine, propylenediamine, trimethylamine, triethylamine, tri-n-propylamine, triisopropylamine, tri-n-butylamine, triisobutylamine, tri-sec-butylamine, tri-n-amylamine, methoxyethylamine and ethoxyethylamine; heterocyclic amines, for example pyridine, quinoline, isoquinoline, morpholine, thio-



morpholine, N-methylmorpholine, N-methyl-thiomorpholine, piperidine, pyrrolidine, indoline, quinuclidine and azepine; primary arylamines, for example anilines, methoxyanilines, ethoxyanilines, o,m,p-toluidines, phenylenediamines, benzidines, naphthylamines and o,m,p-chloroanilines; but especially triethylamine, isopropylamine and diisopropylamine.

Salts of the compounds of formula I with basic groups, especially the derivatives with amino groups, for example alkylamino, dialkylamino or alkenylamino, are, for example, salts with inorganic and organic acids, for example hydrohalic acids, such as hydrofluoric acid, hydrochloric acid, hydrobromic acid or hydroiodic acid, and also sulfuric acid, phosphoric acid, nitric acid and organic acids, such as acetic acid, trifluoroacetic acid, trichloroacetic acid, propionic acid, glycolic acid, thiocyanic acid, citric acid, benzoic acid, oxalic acid, formic acid, benzenesulfonic acid, p-toluenesulfonic acid, methanesulfonic acid and salicylic acid.

The possible presence of at least one asymmetric carbon or sulfur atom in the compounds of formula I, for example in the substituent R_6 = OR_{20} wherein R_{20} is a branched alkyl, alkenyl, haloalkyl or alkoxyalkyl group, or R_6 = $S(O)_mR_{30}$ wherein, for example, m=1 and/or R_{30} is a branched alkyl, alkenyl, haloalkyl or alkoxyalkyl group, means that the compounds can occur both as optically active individual isomers and in the form of racemic mixtures. In the present invention, the compounds of formula I are to be understood as including both the pure optical antipodes and the racemates or diastereoisomers.

If an aliphatic C=C double bond is present, geometric isomerism may occur. The present invention relates also to those isomers.

Preference is given to compounds of formula I wherein R_5 is chlorine, bromine, methyl, trifluoromethyl or cyano.

Preference is given also to compounds of formula I wherein R_6 is hydrogen, halogen, OR_{20} , $S(O)_mR_{30}$ or $COYR_{50}$.

Compounds of formula I wherein n is 0 or 2 are also preferred.

Compounds of formula I wherein R₁ is methyl are also preferred.

Preference is also given to compounds of formula I wherein R₂ is methyl.

Also preferred are compounds of formula I wherein R₃ is methyl or ethyl.

Compounds of formula I wherein R₃ is methyl are especially preferred.

In a group of very especially preferred compounds of formula I, R₄ is fluorine.

In a further very especially preferred group of compounds of formula I, R₄ is hydrogen.

In another very especially preferred group of compounds of formula I, R₄ is chlorine.

In a further group of very especially preferred compounds of formula I, R_4 is chlorine; and R_6 is OR_{20} wherein R_{20} is as defined for formula I.

A group of very especially preferred compounds of formula I comprises compounds wherein R_4 is fluorine; and R_6 is OR_{20} wherein R_{20} is as defined for formula I.

Further groups of very especially preferred compounds of formula I comprise compounds wherein R_4 is chlorine; and R_6 is $S(O)_m R_{30}$ wherein R_{30} and m are as defined for formula I.

In other groups of very especially preferred compounds of formula I, R_4 is fluorine; and R_6 is $S(O)_m R_{30}$ wherein R_{30} and m are as defined for formula I.

A further group of very especially preferred compounds of formula I comprises compounds wherein R_4 is chlorine; and R_6 is COR_{40} , $COYR_{50}$, C_1 - C_4 alkylCOZR₅₂, C_1 - C_4 haloalkylCOZR₅₂, C_2 - C_4 alkenylCOZR₅₂, C_2 - C_4 alkynylCOZR₅₂ or C_2 - C_4 haloalkenylCOZR₅₂ wherein R_{40} , R_{50} , R_{52} , Y and Z are as defined for formula I.

A further group of very especially preferred compounds of formula I comprises compounds wherein R_4 is fluorine; and R_6 is COR_{40} , $COYR_{50}$, C_1 - C_4 alkyl $COZR_{52}$, C_1 - C_4 haloalkyl $COZR_{52}$, C_2 - C_4 alkenyl $COZR_{52}$, C_2 - C_4 alkynyl $COZR_{52}$ or C_2 - C_4 haloalkyl $COZR_{52}$ wherein R_{40} , R_{50} , R_{52} , Y and Z are as defined for formula I.

In a further group of compounds of formula I that are likewise very especially preferred,

R₅ is chlorine; and R₆ is -COYR₅₀.

Another group of likewise very especially preferred compounds of formula I comprises compounds wherein R_5 is chlorine; and R_6 is C_1 - C_4 alkyl-B or C_1 - C_4 haloalkyl-B.

The compounds of formula I

wherein R_1 to R_6 and n are as defined for formula I can be prepared by means of processes known *per se*, for example by cyclising a compound of formula III

$$\begin{array}{c|c}
R_4 & R_1 & S-R_2 \\
\hline
C = C & S-R_2
\end{array}$$
(III),

wherein R_1 , R_2 and R_4 to R_6 are as defined,

a) with hydrazine optionally in the presence of a suitable solvent to form a compound of formula Ha

$$R_{4}$$
 R_{1}
 $S-R_{2}$
 R_{6}
(IIa)

and then reacting that compound in the presence of a compound of formula Xa containing a corresponding C_1 - C_4 alkyl, C_1 - C_4 haloalkyl, C_3 - or C_4 -alkenyl or C_3 - or C_4 -alkynyl group

$$R_3-L_1$$
 (Xa),

the radical R_3 in the compounds of formula Xa being as defined for formula I and L_1 being a leaving group, preferably chlorine, bromine, iodine, CH_3SO_2O - or

$$H_3C$$
 \longrightarrow SO_2O - , optionally in the presence of a suitable solvent to form a compound of formula I

$$R_{5} \xrightarrow{R_{4}} R_{1} \xrightarrow{S(0)_{n}} R_{2}$$

$$R_{5} \xrightarrow{R_{6}} R_{3}$$

$$(I)$$

wherein n is 0, and then oxidising that compound; or

b) with a compound of formula XI

$$NH_2$$
- NH - R_3 (XI),

wherein R_3 is as defined, optionally in the presence of a suitable solvent, to form a compound of formula I

$$R_{5} \xrightarrow{R_{4}} R_{1} \xrightarrow{S(O)_{n}} R_{2}$$

$$R_{5} \xrightarrow{R_{6}} R_{3}$$

$$(I),$$

wherein R_1 to R_6 are as defined, and n is 0, and then oxidising that compound.

The process according to the invention for the preparation of a compound of formula II

$$R_{5}$$
 R_{6}
 R_{1}
 R_{1}
 R_{1}
 R_{2}
 R_{5}
 R_{6}
 R_{1}
 R_{1}
 R_{2}
 R_{3}
 R_{4}
 R_{1}
 R_{1}
 R_{2}
 R_{3}
 R_{4}
 R_{1}
 R_{2}
 R_{3}
 R_{4}
 R_{5}
 R_{5}
 R_{6}
 R_{7}
 R_{1}
 R_{2}
 R_{3}
 R_{4}
 R_{1}
 R_{5}
 R_{5}
 R_{6}
 R_{7}
 R_{1}
 R_{2}
 R_{3}
 R_{4}
 R_{5}
 R_{5

wherein R_1 , R_2 , R_4 to R_6 and n are as defined for formula I, is carried out analogously to known processes and comprises halogenating a compound of formula IV

optionally in the presence of a solvent and a base, for example acetic acid and sodium acetate, to form a compound of formula XII

 R_1 and R_4 to R_6 in the compounds of formulae IV and XII being as defined and Hal being halogen, especially chlorine and bromine, and cyclising that compound of formula XII with a compound of formula XIII

$$H_2N-NH-C(S)S-R_2$$
 (XIII),

wherein R_2 is as defined, optionally in the presence of a solvent, for example an alcohol, for example ethanol, and a base, for example an alcoholate, for example an ethanolate, to form a compound of formula XIV

$$R_5$$
 R_6
 R_1
 S
 SR_2
 SR_2

which is not isolated, and then subjecting that compound to a ring contraction (extrusion reaction) (n=0) thermally or by acid catalysis, for example with 2N hydrochloric acid, and then oxidising the product (n=1 or 2).

The formation of the pyrazole rings of the compounds of formula I wherein n is 0, 1 or 2 is illustrated in more detail in Reaction Schemes 1, 2 and 10 below.

Reaction Scheme 1

Route a):

$$R_4$$
 R_4
 R_5
 R_5
 R_6
 R_5
 R_6
 R_6
 R_7
 R_8
 R_8

Route b):

$$\begin{array}{c} R_4 \\ R_5 \\ R_6 \end{array} \qquad \begin{array}{c} R_4 \\ C \\ C \end{array} \qquad \begin{array}{c} R_2 \\ S - R_2 \\ S - R_2 \end{array} \qquad \begin{array}{c} R_3 - N H N H_2 \\ N - N \\ N \end{array} \qquad \begin{array}{c} R_3 \\ R_3 \end{array} \qquad \begin{array}{c} R_4 \\ R_1 \\ N - N \\ R_3 \end{array} \qquad \begin{array}{c} R_3 \\ R_3 \end{array} \qquad \begin{array}{c} R_4 \\ R_1 \\ N - N \\ R_2 \\ N - N \\ R_3 \end{array} \qquad \begin{array}{c} R_4 \\ R_1 \\ N - N \\ R_3 \\ N - N \\ R_4 \\ N - N \\ N$$

Reaction Scheme 2

The formation of the pyrazole rings of formula IIa that are unsubstituted at the nitrogen atoms (Reaction Scheme 1, Route a)) is carried out by reaction of the compounds of formula III with hydrazine or hydrazine hydrate optionally in the presence of a suitable solvent at elevated temperature, preferably with hydrazine hydrate in alcoholic solution at elevated temperature.

For the formation of the pyrazole rings that are substituted at the nitrogen atom (Reaction Scheme 1, Route b)), the procedure is analogous to that indicated under Reaction Scheme 1, Route a), with a compound of formula XI, for example N-alkylhydrazine, preferably N-methylhydrazine, being used as reagent.

The formation of the pyrazole rings of formulae II and IIa that are unsubstituted at the nitrogen atoms (Reaction Scheme 2) can be carried out, for example, also by halogenation of the compounds of formula IV preferably with chlorine or bromine optionally in the presence of a suitable solvent and a base, for example acetic acid and sodium acetate, subsequent cyclisation with a compound of formula XIII optionally in a solvent, for example an alcohol, preferably ethanol, and in the presence of a base, for example an alcoholate, preferably an ethanolate, and ring contraction (extrusion reaction) analogously to known procedures, as described, for example, in Chem. Ber. 92, 2593 (1959) or Acta Chem. Scand. 16, 2395 (1962). That method, described in Reaction Scheme 2, is suitable for the preparation of derivatives of formulae IIa and II that are substituted by halogen, especially by fluorine and chlorine, at the phenyl ring.

In certain cases it is advantageous to prepare the N-alkyl-substituted pyrazole derivatives, especially the N-methyl-substituted pyrazole derivatives, via N-alkylation of the corresponding unsubstituted pyrazoles of formula II (or IIa). Reaction Scheme 3 illustrates this.

Reaction Scheme 3

In Reaction Schemes 1, Route b), and 3, the radical R₃ in the hydrazine derivative of

formula XI and in the alkylating agents of formulae Xa and Xb is as defined for formula I, and L_1 is a leaving group, for example chlorine, bromine, iodine, CH_3SO_2O - or

The N-alkylation of the pyrazole rings in the compounds of formulae II and IIa in Reaction Schemes 1 and 3 is carried out at room temperature or at slightly elevated temperatures in the presence of a solvent, for example acetone, methyl ethyl ketone, N,N-dimethylformamide, N-methylpyrrolidone or dimethyl sulfoxide, a base, for example potassium carbonate, sodium carbonate, sodium hydroxide or potassium hydroxide, and an alkylating agent of formula Xa or Xb, preferably methyl iodide or dimethyl sulfate.

The selection of the suitable preparation method and the corresponding reaction conditions is made in accordance with the properties (reactivities) of the substituents in the intermediates in question.

The subsequent oxidation of the compounds of formula I wherein n is 0 (Reaction Schemes 1 and 2) is carried out, for example, with peracids, for example m-chloroper-benzoic acid, or hydrogen peroxide in the presence of a suitable solvent, for example dichloromethane, chloroform or carbon tetrachloride, at temperatures of from -40°C to the reflux temperature of the solvent in question, preferably from 0°C to 35°C. The degree of oxidation at the sulfur atom can be controlled by the amount of oxidising agent: in the case of an equimolar amount of oxidising agent, compounds of formula I wherein n is 1 are obtained and in the case of an excess (at least 2 mol) of oxidising agent, compounds of formula I wherein n is 2 are obtained.

The starting compound of formula III in Reaction Scheme 1 can be prepared analogously to known procedures, for example in accordance with the method given in Reaction Scheme 4 below:

Reaction Scheme 4

The reaction in Reaction Scheme 4 is carried out, for example, analogously to WO 92/02509, according to which the phenylcarbonyl derivative of formula IV is allowed to react in the presence of a base, for example sodium hydride or potassium tert-butanolate, and an aprotic solvent, for example tetrahydrofuran, with carbon disulfide at temperatures of from 0° C to 80° C, and immediately afterwards an alkylating agent, for example R_2 -Hal or R_2 OSO $_2$ OR $_2$ wherein R_2 is as defined for formula I and Hal is halogen, especially chlorine, bromine or iodine, is added at temperatures of from 0° C to the reflux temperature of the solvent used.

The compounds of formula XIII in Reaction Scheme 2 can be prepared in accordance with known methods (for example Chem. Ber. 92, 2593 (1959) or Acta Chem. Scand. 16, 2395 (1962)), for example by reaction of hydrazine or hydrazine hydrate with carbon disulfide and subsequent alkylation with the reagent R₂-Hal or R₂OSO₂OR₂ wherein R₂ is as defined for formula I and Hal is halogen, especially chlorine or bromine, in the presence of a base. Suitable solvents are, for example, alcohols, for example ethanol, and suitable bases are, for example, alcoholates, for example sodium methanolate or sodium ethanolate, or potassium or sodium hydroxide.

The starting compound of formula IV in Reaction Scheme 4 can be prepared analogously to known procedures, for example in accordance with Methods a), b), c) and d) given in Reaction Scheme 5 below.



Reaction Scheme 5

Method a):

$$R_{5}$$
 R_{6}
 V_{I}
 R_{6}
 V_{I}
 R_{6}
 V_{I}
 R_{6}
 R_{6}
 R_{6}
 R_{6}
 R_{6}
 R_{7}
 R_{7}
 R_{8}
 R_{1}
 R_{2}
 R_{2}
 R_{3}
 R_{4}
 R_{4}
 R_{4}
 R_{4}
 R_{5}
 R_{6}
 R_{7}
 R_{8}
 R_{1}
 R_{2}
 R_{3}
 R_{4}
 R_{5}
 R_{6}
 R_{7}
 R_{8}
 R_{1}
 R_{1}
 R_{2}
 R_{3}
 R_{4}
 R_{5}
 R_{6}
 R_{7}
 R_{8}

Method b):

$$R_{5} \xrightarrow{R_{4}} COX_{1} \xrightarrow{e.g. R_{1}-CH_{2}Li \text{ or}} R_{5} \xrightarrow{R_{4}} CH_{2}-R_{1}$$

$$R_{6} \xrightarrow{Va} Va$$

$$(X_{1}=-OH \text{ or }-N(CH_{3})OCH_{3})$$

$$R_{6} \xrightarrow{Va} IV$$

Method c):

$$R_{5}$$
 R_{6}
 V_{C}
 R_{1}
 CH_{2}
 R_{1}
 CH_{2}
 R_{1}
 R_{2}
 CH_{2}
 R_{3}
 R_{4}
 CH_{2}
 R_{1}
 R_{4}
 CH_{2}
 R_{1}
 R_{5}
 R_{6}
 R_{6}
 R_{6}
 R_{7}
 R_{8}
 R_{1}
 R_{2}
 R_{3}
 R_{4}
 R_{5}
 R_{4}
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 R_{8}
 R_{9}
 R_{1}
 R_{1}
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 R_{9}
 R_{1}
 R_{1}
 R_{2}
 R_{3}
 R_{4}
 R_{5}
 R_{5}
 R_{6}
 R_{7}
 R_{8}
 R_{8}
 R_{9}
 R_{9}
 R_{1}

Method d):

In Reaction Scheme 5, the radicals R_1 , R_4 , R_5 and R_6 are as defined for formula I, it being necessary to note that not all substituent definitions are compatible with all the procedures indicated. The selection of the suitable preparation method is made in accordance with the properties (reactivities) of the substituents in the intermediates in question.

The reaction according to Method a) in Reaction Scheme 5 is carried out analogously to 'Vogel's Textbook of Practical Organic Chemistry', Longman 1989, page 1006 ff. In that

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reaction the aromatic compound of formula V is allowed to react in the presence of a carboxylic acid derivative, for example a carboxylic acid chloride of formula VI and an acid, for example a Lewis acid such as aluminium chloride, with or without a solvent at temperatures of from 0°C to 150°C.

The reaction according to Method b) in Reaction Scheme 5 is carried out, for example, starting from the carboxylic acid derivatives of formula Va wherein X₁ is -OH or -N(CH₃)OCH₃ with an alkyllithium compound of formula VII or a Grignard compound (alkylmagnesium chloride or bromide) of formula VIII in an inert solvent, preferably diethyl ether at temperatures of from -100°C to 50°C, analogously to Organic Reactions 18, 1 (1970), Organic Synthesis 49, 81 (1969) and 'Comprehensive Organic Transformations', Editor R.C. Larock, VCH 1989, page 685.

The reaction according to Method c) in Reaction Scheme 5 is carried out analogously to 'Advanced Organic Chemistry', Editor J. March, McGraw-Hill Book Company, New York, 1985, pages 816 ff. and 1057 ff., starting from an aldehyde of formula Vc by means of a Grignard reagent of formula VIII, for example alkylmagnesium chloride or bromide, or by means of alkyllithium in an inert solvent, preferably diethyl ether, at temperatures of from -80°C to 25°C and subsequent oxidation of the alcohol to the ketone. Suitable oxidising agents are, for example, potassium permanganate, pyridinium dichromate and sodium dichromate.

The reaction according to Method d) in Reaction Scheme 5 is carried out analogously to J. Chem. Soc. 1954, 1297. The amines of formula Vb are accordingly first diazotised to form the corresponding diazonium salts and allowed to react with an aldehydeoxime of formula IX. Subsequent hydrolysis, for example with aqueous sodium acetate and copper sulfate, yields the corresponding methyl ketone of formula IV.

The starting compounds of formulae V, Va, Vb, Vc, VI, VII, VIII, IX and XI in Reaction Schemes 1, Route b), and 5 are known or can be prepared in accordance with known procedures.

The phenylpyrazole derivatives of formula II are novel and have been developed especially for the synthesis of the compounds of formula I. They are therefore also a subject of the present invention.

A large number of known standard procedures is available for the preparation of the phenylpyrazoles of formula I substituted in the 5-position of the phenyl ring (R_6) , the selection of the suitable preparation processes being made in accordance with the properties (reactivities) of the substituents in the intermediates in question. Some examples are given in Reaction Schemes 6 to 9.

The preparation of the phenylpyrazole derivatives of formula I that are O-substituted in the 5-position of the phenyl ring, wherein $R_6 = OR_{20}$, starting from the methoxy- or benzyloxy-substituted derivatives of formula I_1 or I_2 , respectively, is illustrated in Reaction Scheme 6.

Reaction Scheme 6

The phenolpyrazole derivatives of formula I_3 in Reaction Scheme 6 can be obtained, for example, a) from the compounds of formula I_1 via ether cleavage by means of lithium chloride in N,N-dimethylformamide (DMF) at elevated temperature, as described, for example, in Synthesis 1989, 287, or by means of boron tribromide in dichloromethane at temperatures of from -80°C to 20°C, as described, for example, in Org. Synth., Collect. Vol. V, 412, 1973; or b) from the compounds of formula I_2 via hydrogenolysis by means of hydrogen in the presence of a catalyst, for example palladium on carbon, as described, for example, in J. Am. Chem. Soc. 93, 746 (1971).

The derivatisation of the phenolpyrazoles of formula I_3 in Reaction Scheme 6 to form the compounds of formula I can be carried out in accordance with standard procedures, for example via alkylation with R_{20} -Hal wherein R_{20} is as defined for formula I and Hal is halogen, especially chlorine, bromine or iodine.

The preparation of the phenylpyrazole derivatives of formula I that are S-substituted in the 5-position of the phenyl ring, wherein $R_6=S(O)_mR_{30}$, starting from the derivatives of formula I_4 that are unsubstituted in the 5-position, is illustrated in Reaction Scheme 7.

Reaction Scheme 7

The preparation of the thiophenolpyrazoles of formula I_6 in Reaction Scheme 7 can be carried out analogously to known procedures, as described, for example, in J. Org. Chem. 54, 6096 (1989), EP-A-0 259 265 or in "Sulfonation and Related Reactions", Editor Gilbert, Interscience Publishers, New York 1965. The phenylpyrazole of formula I_4 is then chlorosulfonylated with chlorosulfonic acid or sulfur trioxide in sulfuric acid to form the compound of formula I_5 and then reduced with tin chloride or zinc chloride to the thiophenol derivative of formula I_6 . The derivatisation of the thiophenolpyrazoles of formula I_6 to form the compounds of formula I in Reaction Scheme 7 can be carried out in accordance with standard procedures, for example via alkylation with R_{30} -Hal wherein R_{30} is as defined for formula I and Hal is halogen, especially chlorine, bromine or iodine (m=0). The subsequent oxidation to the sulfine or sulfone derivatives of formula I (m= 1 or 2, respectively) can likewise be carried out in accordance with standard procedures, for example with peracids, for example m-chloroperbenzoic acid.

The preparation of the phenylpyrazole derivatives of formula I that are carboxy-

substituted in the 5-position of the phenyl ring, wherein R_6 is halogen, cyano, nitro, amino, NHR₁₀, NR₁₀R₁₁, COR₄₀ or COYR₅₀, starting from the derivatives of formula I_4 or I_{11} that are unsubstituted in the 5-position or triflate-substituted in the 5-position, respectively, is illustrated in Reaction Scheme 8.

Reaction Scheme 8

R₅

HNO₃

R₅

NO₂

R₄

H₂/cat or Bechamps

R₅

NH₂ I₈

R₄

I₇

R₄

R₅

NO₂

NH₂ I₈

R₄

PdCl₂(TPP)₂

CO
EtOH, p,
$$\Delta$$
T

R₅

HOOC

R₄

HOOC

R₄

R₅

W

derivatisation

standard methods
e.g. alkylation,
acylation

R₄

PdCl₂(TPP)₂

CO
EtOH, p, Δ T

R₅

HOOC

R₄

HOOC

R₄

R₅

R₄

PdCl₂(TPP)₂

CO
EtOH, p, Δ T

R₅

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In Reaction Scheme 8, W is the radical $N \cdot N = S(0) \cdot R_2$ wherein R_1 to R_3 and R_3 and R_4 are as

defined for formula I. In accordance with Reaction Scheme 8, the phenylpyrazole of formula I_4 can be converted into an aniline derivative of formula I_8 in accordance with standard procedures, for example nitration in a nitric acid and sulfuric acid mixture and subsequent reduction of the resulting nitro compound of formula I_7 with hydrogen in the presence of a catalyst or according to Bechamps. The aniline derivative of formula I_8 can then be either derivatised directly to form the corresponding compounds of formula I according to standard procedures, for example alkylation or acylation, or converted into the halogen compound of formula I_9 by means of diazotisation and Sandmeyer reaction. The benzoic acid ester of formula I_{10} in Reaction Scheme 8 can be obtained, for example, analogously to J. Org. Chem. 39, 3318 (1974) or ibid. 40, 532 (1975) from the compound

of formula I_9 by means of carbon monoxide and a catalyst, for example palladium chloride-triphenylphosphine $(PdCl_2(TPP)_2)$ in the presence of a solvent, for example ethanol, optionally under pressure at elevated temperature. A further possible method of preparing the intermediate of formula I_{10} is carried out analogously to Tetrahedron Letters 25, 2271 (1984) and ibid. 27, 3931 (1986). According to that method, the compound of formula I_{11} is carbonylated in the presence of a catalyst, for example palladium. The subsequent hydrolysis of the benzoic acid ester of formula I_{10} yields the benzoic acid derivative of formula I_{12} , which can be converted into the corresponding compounds of formula I in accordance with standard procedures, for example esterification or amidation.

The preparation of the phenylpyrazole derivatives of formula I that are substituted in the 5-position of the phenyl ring, wherein R_6 is C_1 - C_4 alkylCOZR $_{52}$, C_1 - C_4 haloalkylCOZR $_{52}$, C_2 - C_4 alkenylCOZR $_{52}$, C_2 - C_4 alkynylCOZR $_{52}$ or C_2 - C_4 haloalkenylCOZR $_{52}$, starting from the derivatives of formula I_9 that are substituted in the 5-position of the phenyl ring by halogen, especially by chlorine, bromine or iodine, via Heck reaction (Route a)), or starting from the derivatives of formula I_8 that are substituted in the 5-position of the phenyl ring by amino via diazotisation and subsequent Meerwein reaction, is illustrated in Reaction Scheme 9.

Reaction Scheme 9

Route a):

In Reaction Scheme 9, W is the radical (N, N) wherein R_1 to R_3 and R_3 and R_3 wherein R_1 to R_3 and R_3 are as

defined for formula I. In accordance with Reaction Scheme 9, Route a), the alkynyl ester derivatives of formula I_{13} can be prepared, for example, via Heck reaction analogously to

R. F. Heck in W. G. Dauben (Edit.), Organic Reactions 27, 345 (1982). It is possible to obtain therefrom by means of standard procedures, for example by means of partial or complete hydrogenation, the corresponding alkenyl- or alkyl-COZR₅₂ derivatives, respectively, or *via* halogenation the corresponding haloalkenyl- or haloalkyl-COZR₅₂ derivatives of formula I.

In accordance with Reaction Scheme 9, Route b), the haloalkylCOZR $_{52}$ derivatives of formula I_{14} can be produced from the aniline derivatives of formula I_{8} analogously to Organic Reactions 11, 189-260 (1960) via diazotisation and Meerwein reaction. Known standard procedures, for example hydrogenolysis or halogen removal, yield therefrom the corresponding alkyl- or alkenyl-COZR $_{52}$ derivatives of formula I.

Reaction scheme 10

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The formation of the pyrazole rings of the compounds of formula I according to Reaction Scheme 10 is carried out

a) starting from the ketone derivative of formula IV wherein R_1 and R_4 to R_6 are as defined for formula I, by reaction with a carbonate of formula XVI wherein R_7 is C_1 - C_4 alkyl, phenyl or benzyl, in the presence of a base, especially the corresponding sodium alcoholate R_7 O⁻ +Na, in a solvent, for example the corresponding alcohol R_7 OH

together with a second solvent, for example an ether or hydrocarbon, at temperatures of from 0°C to the boiling point of the solvent in question, or

b) starting from the α -haloketone of formula XII wherein R_1 and R_4 to R_6 are as defined and Hal is halogen, especially chlorine or bromine, by carbonylation in the presence of the palladium(II) chloride•bis-triphenylphosphine complex $PdCl_2[(PH)_3P]_2$ as catalyst, carbon monoxide and a phase transfer catalyst in a suitable solvent, for example the alcohol R_7OH wherein R_7 is as defined, together with a second solvent, for example N,N-dimethylformamide, analogously to Indian J. Chem. B 31, 363 (1982).

In Reaction Scheme 10 the preparation of the pyrazolinone rings of formula XVII that are unsubstituted at the nitrogen atom is carried out by reaction of the above-prepared keto ester of formula XVI with hydrazine or hydrazine hydrate optionally in the presence of a suitable solvent at elevated temperature, preferably hydrazine hydrate in alcoholic solution at elevated temperature.

For the preparation of the pyrazolinone rings of formula XVIII that are substituted at the nitrogen atom, the reagent used is the compound of formula XI, for example N-alkyl-hydrazine, preferably N-methylhydrazine.

If desired, the substituted pyrazolinone derivative of formula XVIII can be prepared also via N-alkylation of the corresponding unsubstituted pyrazolinones of formula XVII, in a manner analogous to that described in Reaction Scheme 3.

The conversion of the pyrazolinone derivatives of formula XVIII into the corresponding mercapto analogues of formula XV is carried out in accordance with standard methods, for example with the aid of Lawesson reagent in a suitable solvent at elevated temperatures. The subsequent S-alkylation yields compounds of formula I (n=0) and is carried out in accordance with standard methods with the aid of an alkylating agent, for example R_2 -Hal wherein R_2 is as defined for formula I and Hal is halogen, especially chlorine or bromine, optionally in the presence of a solvent and a base.

The compounds of formula XV are novel and have been developed especially for the synthesis of the compounds of formula I. They are therefore also a subject of the present invention.

All further compounds originating from the scope of formula I can easily be prepared from the described compounds of formula I in manner analogous to that described above, or in accordance with methods as described, for example, in "Methoden der Organischen Chemie" (Houben-Weyl), Volume E 8b, Georg Thieme Verlag Stuttgart, 1994, page 399 ff. or in "Pyrazoles, Pyrazolines, Pyrazolidines, Indazoles and Condensed Rings", Editor R. H. Wiley, Interscience Publishers, New York, 1967, page 1 ff., or by derivatisation in accordance with known standard methods, for example alkylation, acylation and amidation.

The end products of formula I can be isolated in customary manner by concentration and/or evaporation of the solvent and purified by recrystallisation or trituration of the solid residue in solvents in which they are not readily soluble, such as ethers, aromatic hydrocarbons or chlorinated hydrocarbons, by distillation or by means of column chromatography or flash column chromatography and a suitable eluant.

For the use according to the invention of the compounds of formula I, or compositions comprising them, there come into consideration all the methods of application customary in agriculture, for example pre-emergence application, post-emergence application and seed dressing, and also various methods and techniques, for example the controlled release of active ingredient. For that purpose a solution of the active ingredient is applied to mineral granule carriers or polymerised granules (urea/formaldehyde) and dried. If required, it is also possible to apply a coating (coated granules) which allows the active ingredient to be released in metered amounts over a specific period of time.

The compounds of formula I may be used in unmodified form, that is to say as obtained in the synthesis, but they are preferably formulated in customary manner together with the adjuvants customarily employed in formulation technology e.g. into emulsifiable concentrates, directly sprayable or dilutable solutions, dilute emulsions, wettable powders, soluble powders, dusts, granules or microcapsules. As with the nature of the compositions, the methods of application, such as spraying, atomising, dusting, wetting, scattering or pouring, are chosen in accordance with the intended objectives and the prevailing circumstances.

The formulations, i.e. the compositions, preparations or mixtures comprising the compound (active ingredient) of formula I or at least one compound of formula I and generally one or more solid or liquid formulation adjuvants, are prepared in known

manner, e.g. by homogeneously mixing and/or grinding the active ingredients with the formulation adjuvants, e.g. solvents or solid carriers. It is also possible to use surface-active compounds (surfactants) in the preparation of the formulations.

Suitable solvents are: aromatic hydrocarbons, preferably the fractions containing 8 to 12 carbon atoms, such as mixtures of alkylbenzenes, e.g. xylene mixtures or alkylated naphthalenes; aliphatic and cycloaliphatic hydrocarbons, such as paraffins, cyclohexane or tetrahydronaphthalene; alcohols, such as ethanol, propanol or butanol; glycols and their ethers and esters, such as propylene glycol or dipropylene glycol ether; ketones, such as cyclohexanone, isophorone or diacetone alcohol; strongly polar solvents, such as N-methyl-2-pyrrolidone, dimethyl sulfoxide or water; vegetable oils and esters thereof, such as rape oil, castor oil or soybean oil; and, where appropriate, also silicone oils.

The solid carriers used, e.g. for dusts and dispersible powders, are normally natural mineral fillers, such as calcite, talcum, kaolin, montmorillonite or attapulgite. In order to improve the physical properties it is also possible to add highly dispersed silicic acid or highly dispersed absorbent polymers. Suitable granulated adsorptive carriers are porous types, for example pumice, broken brick, sepiolite or bentonite, and suitable nonsorbent carriers are, for example, calcite or sand. In addition, a great number of pregranulated materials of inorganic or organic nature can be used, e.g. especially dolomite or pulverised plant residues.

Depending upon the nature of the compound of formula I to be formulated, suitable surface-active compounds are non-ionic, cationic and/or anionic surfactants having good emulsifying, dispersing and wetting properties. The term "surfactants" will also be understood as comprising mixtures of surfactants.

Both water-soluble soaps and water-soluble synthetic surface-active compounds are suitable anionic surfactants.

Suitable soaps are the alkali metal salts, alkaline earth metal salts or unsubstituted or substituted ammonium salts of higher fatty acids (C_{10} - C_{22}), e.g. the sodium or potassium salts of oleic or stearic acid, or of natural fatty acid mixtures which can be obtained e.g. from coconut oil or tallow oil. Mention may also be made of fatty acid methyltaurin salts.

More frequently, however, so-called synthetic surfactants are used, especially fatty alcohol

sulfonates, fatty alcohol sulfates, sulfonated benzimidazole derivatives or alkylaryl-sulfonates.

The fatty alcohol sulfonates or sulfates are usually in the form of alkali metal salts, alkaline earth metal salts or unsubstituted or substituted ammonium salts and contain a C_8 - C_{22} alkyl radical, which also includes the alkyl moiety of acyl radicals, e.g. the sodium or calcium salt of lignosulfonic acid, of dodecyl sulfate or of a mixture of fatty alcohol sulfates obtained from natural fatty acids. These compounds also comprise the salts of sulfated and sulfonated fatty alcohol/ethylene oxide adducts. The sulfonated benzimidazole derivatives preferably contain two sulfonic acid groups and one fatty acid radical containing 8 to 22 carbon atoms. Examples of alkylarylsulfonates are the sodium, calcium or triethanolamine salts of dodecylbenzenesulfonic acid, dibutylnaphthalenesulfonic acid, or of a condensate of naphthalenesulfonic acid and formaldehyde.

Also suitable are corresponding phosphates, e.g. salts of the phosphoric acid ester of an adduct of p-nonylphenol with 4 to 14 mol of ethylene oxide, or phospholipids.

Non-ionic surfactants are preferably polyglycol ether derivatives of aliphatic or cyclo-aliphatic alcohols, saturated or unsaturated fatty acids and alkylphenols, said derivatives containing 3 to 30 glycol ether groups and 8 to 20 carbon atoms in the (aliphatic) hydrocarbon moiety and 6 to 18 carbon atoms in the alkyl moiety of the alkylphenols.

Further suitable non-ionic surfactants are the water-soluble adducts of polyethylene oxide with polypropylene glycol, ethylenediaminopolypropylene glycol and alkylpolypropylene glycol containing 1 to 10 carbon atoms in the alkyl chain, which adducts contain 20 to 250 ethylene glycol ether groups and 10 to 100 propylene glycol ether groups. These compounds usually contain 1 to 5 ethylene glycol units per propylene glycol unit.

Examples of non-ionic surfactants are nonylphenol polyethoxyethanols, castor oil polyglycol ethers, polypropylene/polyethylene oxide adducts, tributylphenoxypolyethoxyethanol, polyethylene glycol and octylphenoxypolyethoxyethanol.

Fatty acid esters of polyoxyethylene sorbitan, e.g. polyoxyethylene sorbitan trioleate, are also suitable non-ionic surfactants.

Cationic surfactants are preferably quaternary ammonium salts which contain, as

N-substituent, at least one C_8 - C_{22} alkyl radical and, as further substituents, unsubstituted or halogenated lower alkyl, benzyl or hydroxy-lower alkyl radicals. The salts are preferably in the form of halides, methyl sulfates or ethyl sulfates, e.g. stearyltrimethylammonium chloride or benzyldi(2-chloroethyl)ethylammonium bromide.

The surfactants customarily employed in formulation technology, which may also be used in the compositions according to the invention, are described <u>inter alia</u> in "Mc Cutcheon's Detergents and Emulsifiers Annual", MC Publishing Corp., Ridgewood, New Jersey, 1981; Stache, H., "Tensid-Taschenbuch" (Surfactant Handbook), Carl Hanser Verlag, Munich/Vienna 1981; and M. and J. Ash, "Encyclopedia of Surfactants", Vol. I-III, Chemical Publishing Co., New York, 1980-1981.

The herbicidal compositions usually comprise 0.1 to 99%, preferably 0.1 to 95%, of a compound of formula I, 1 to 99.9% of a solid or liquid formulation adjuvant, and 0 to 25%, preferably 0.1 to 25%, of a surfactant.

Whereas commercial products will preferably be formulated as concentrates, the end user will normally employ dilute formulations.

The compositions may also comprise further auxiliaries, such as stabilisers, e.g. vegetable oils or epoxidised vegetable oils (epoxidised coconut oil, rape oil or soybean oil), antifoams, e.g. silicone oil, preservatives, viscosity regulators, binders and tackifiers, as well as fertilisers or other active ingredients.

Preferred formulations have especially the following composition (throughout, percentages are by weight):

Emulsifiable concentrates:

active ingredient:

1 to 90 %, preferably 5 to 50 %

surface-active agent:

5 to 30 %, preferably 10 to 20 %

solvent:

15 to 94 %, preferably 70 to 85 %

Dusts:

active ingredient:

0.1 to 50 %, preferably 0.1 to 1 %

solid carrier:

99.9 to 90 %, preferably 99.9 to 99 %

Suspension concentrates:

active ingredient:

water:

surface-active agent:

5 to 75 %, preferably 10 to 50 %

94 to 24 %, preferably 88 to 30 %

1 to 40 %, preferably 2 to 30 %

Wettable powders:

active ingredient:

surface-active agent:

solid carrier:

0.5 to 90 %, preferably 1 to 80 %

0.5 to 20 %, preferably 1 to 15 %

5 to 95 %, preferably 15 to 90 %

Granules:

active ingredient:

solid carrier:

0.1 to 30 %, preferably 0.1 to 15 %

99.5 to 70 %, preferably 97 to 85 %

The compounds of formula I are generally used successfully when applied to the plant or to the locus thereof at rates of application of from 0.001 to 2 kg/ha, especially from 0.005 to 1 kg/ha. The concentration required to achieve the desired effect can be determined by experiment. It is dependent upon the type of action, the stage of development of the cultivated plant and of the weed, and also upon the application (place, time, method) and, in dependence on those parameters, can vary within wide limits.

The compounds of formula I are distinguished by herbicidal and growth-inhibiting properties which render them suitable for use in crops of useful plants, especially in cereals, cotton, soybeans, sugar beet, sugar cane, plantation crops, rape, maize and rice, and also for non-selective weed control.

Crops are also to be understood as being those which have been rendered tolerant to herbicides or classes of herbicide by conventional methods of breeding or genetic engineering.

The following Examples further illustrate, but do not limit, the invention.

Preparation examples

Example P1: 1-(2,4-Dichlorophenyl)-3,3-bis(methylthio)-2-methyl-2-propen-1-one

1.07 g (0.005 mol) of 1-(2,4-dichlorophenyl)-1-propanone (95 %) are added to an emulsion of 0.24 g (0.01 mol) of sodium hydride in 8 ml of dry tetrahydrofuran and the mixture is stirred at 40°C for one hour. The reaction mixture is cooled to 0°C and 0.381 g (0.05 mol) of carbon disulfide is added at 0-5°C. Immediately after the addition of carbon disulfide is complete, 1.42 g (0.01 mol) of methyl iodide are added dropwise at 0-5°C and the reaction mixture is stirred for one hour, then poured into 25 ml of a mixture of ice/water and stirred for a further 30 minutes. The crude product is extracted with ether and the organic phase is separated off, dried over sodium sulfate and concentrated by evaporation. The resulting residue is purified by column chromatography using 10 % ethyl acetate in hexane as eluant. The desired product is obtained in a yield of 1.38 g (89.8 %) in the form of a yellow oil.

Example P2: 3-(2,4-Dichlorophenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole

0.138 g (0.003 mol) of methyl hydrazine is added to a solution of 0.614 g (0.002 mol) of 1-(2,4-dichlorophenyl)-3,3-bis(methylthio)-2-methyl-2-propen-1-one in 10 ml of acetonitrile and the reaction mixture is heated at 80°C for 8 hours. The mixture is then concentrated *in vacuo* and the residue is taken up in ether, washed with water and dried over sodium sulfate. After the ether phase has been concentrated by evaporation, the residue is purified by column chromatography over silica gel with hexane/ethyl acetate 2/1 as eluant. The desired product is obtained in a yield of 0.425 g (74.0 %) in the form of a white solid having a melting point of 65-66°C (recrystallised from hexane).

Example P3: 3-(2,4-Dichlorophenyl)-4-methyl-5-(methylsulfonyl)-1-methyl-[1H]-pyrazole

0.515 g (0.0015 mol) of m-chloroperbenzoic acid (50-60 %) is added to a solution of 0.214 g (0.00075 mol) of 3-(2,4-dichlorophenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole in 10 ml of dichlormethane and the reaction mixture is stirred at 22°C overnight. The mixture is then washed with a saturated sodium hydrogen carbonate solution that contains 5 % sodium thiosulfate and then with water, dried over sodium sulfate and concentrated by evaporation. The resulting residue is purified by column chromatography over silica gel with hexane/ethyl acetate 1/1. The desired product is obtained in a yield of 0.206 g (86.2 %) in the form of a white solid having a melting point of 110-112°C.

Example P4: 3-(4-Chloro-2-fluorophenyl)-4-methyl-5-(methylthio)-[1H]-pyrazole

0.6 ml (0.01 mol) of carbon disulfide in 2 ml of ethanol is added dropwise at 20-28°C to 3.35 ml (0.01 mol) of a 21 % ethanolic sodium ethanolate solution and 0.49 ml (0.01 mol) of hydrazine hydrate in 10 ml of ethanol. After 0.5 hour's stirring at 22°C, 0.62 ml (0.01 mol) of methyl iodide is added dropwise at 20-22°C to the suspension that has formed. After 1 hour's stirring at 22°C, 3.35 ml (0.01 mol) of 21 % ethanolic sodium ethanolate solution are added to the resulting suspension and stirring is continued for a further 10 minutes. The resulting solution is then cooled to 5-10°C and 2.7 g (0.01 mol) of α-bromo-4-chloro-2-fluoropropiophenone in 5 ml of ethanol are added dropwise to the solution and stirring is continued for a further 10 minutes at 40-45°C. 10 ml of 2N hydro-chloric acid are added dropwise at 22°C to the solution that has formed and stirring is continued for 1 hour at 22°C and then for 15 minutes at 40-45°C. The resulting suspension

is dissolved in tert-butyl methyl ether, washed three times with water and concentrated by evaporation using a rotary evaporator. As residue there is obtained 2.5 g of an oil which is chromatographed over 50 g of silica gel with ethyl acetate/hexane 1/2 as eluant. 1.4 g (54.6%) of the desired compound are obtained in the form of colourless crystals having a melting point of 88-90°C.

Example P5: α-Bromo-4-chloro-2-fluoropropiophenone

2.2 ml (0.42 mol) of bromine are added dropwise at 20-25°C to a solution of 8.5 g (0.04 mol) of 4-chloro-2-fluoropropiophenone and 0.1 ml of 33 % hydrobromic acid in 20 ml of acetic acid. After being stirred for 1 hour at 22°C, the mixture is poured into icewater, extracted with tert-butyl methyl ether and washed neutral with dilute sodium hydrogen carbonate solution. The reaction mixture is then concentrated by evaporation using a rotary evaporator and then dried at 22°C for 1 hour under a high vacuum. 10.4 g of 95 % α -bromo-4-chloro-2-fluoropropiophenone are obtained in the form of a light-yellow oil.

Example P6: α-Bromo-4-chloro-2-fluoro-5-methoxypropiophenone

54 ml (1.05 mol) of bromine are added dropwise at 20-25°C to a suspension of 217 g (1 mol) of 4-chloro-2-fluoro-5-methoxypropiophenone and 10 ml of 33 % hydrobromic acid in glacial acetic acid in 0.5 litre of acetic acid. After being stirred for 1 hour to complete the reaction, the solution that has formed is poured into 2 litres of ice-water, extracted with tert-butyl methyl ether (MTBE), washed four times with water, dried over magnesium sulfate and concentrated by evaporation. 284 g of the desired compound are obtained in the form of a 95 % oil.

Example P7: 3-(4-Chloro-2-fluoro-5-methoxyphenyl)-4-methyl-5-(methylthio)-[1H]-pyrazole

At 0-5°C, 37 g (0.3 mol) of methyl dithiocarbazate are added in portions to a solution of 120 ml (0.32 mol) of 21 % sodium ethanolate solution in ethanol. 95 g (0.3 mol) of α-bromo-4-chloro-2-fluoro-5-methoxypropiophenone (90 %) are added dropwise at -5-0°C to the resulting solution. After being stirred for 30 minutes at 0-5°C, the suspension that has formed is diluted with 250 ml of ethanol and stirred at 0-5°C for a further 2 hours. Then, after 1 hour's stirring at 22°C, 25 ml of 37 % hydrochloric acid are added dropwise at 25-30°C and stirring is continued for a further 3 hours. Then 30 ml of 30 % sodium hydroxide solution are added dropwise and the resulting mixture is concentrated by evaporation. After the addition of about 1 litre of tert-butyl methyl ether, the reaction mixture is washed with water and the organic phase is dried and concentrated by evaporation. The crude product is purified by silica gel chromatography. 60 g (70 % of the theoretical yield) of the desired compound are obtained in the form of brown crystals having a melting point of 95-100°C.

Example P8: 3-(4-Chloro-2-fluorophenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole

1.4 ml (0.022 mol) of methyl iodide are added to a mixture of 5.4 g (0.02 mol) of 3-(4-chloro-2-fluorophenyl)-4-methyl-5-(methylthio)-[1H]-pyrazole (Example P4) and 4.2 g (0.030 mol) of potassium carbonate in 25 ml of 1-methyl-2-pyrrolidone (NMP).

After stirring overnight at 22°C, approximately 200 ml of tert-butyl methyl ether (MTBE) are added, and the mixture is washed three times with water and concentrated by evaporation. The crude product is purified by means of silica gel chromatography. 3.2 g (59 % of the theoretical yield) of the desired compound are obtained in the form of an oil.

Example P9: 3-(4-Chloro-2-fluoro-5-hydroxyphenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole

A mixture of 15 g (0.05 mol) of 3-(4-chloro-2-fluoro-5-methoxyphenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole (Example P7) and 10.6 g (0.25 mol) of lithium chloride in 100 ml of N,N-dimethylformamide is stirred at 22°C in an argon atmosphere for 2.5 days. The solution is cooled and poured into 0.5 litre of ice-water and 15 ml of 37 % hydrochloric acid and extracted with tert-butyl methyl ether (MTBE). The organic phase is extracted with a dilute sodium hydroxide solution, and the aqueous phase is separated off and again acidified and extracted with MTBE. After concentration by evaporation and recrystallisation in diethyl ether, 4.8 g (33.5 % of the theoretical yield) of the desired product are obtained in the form of crystals having a melting point of 136-138°C and 7.6 g of crude product of the desired compound in the form of an oil.

Example P10: 3-(4-Chloro-2-fluoro-5-trifluoromethylsulfonyloxyphenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole

65 ml (0.8 mol) of pyridine and then 80 ml (0.48 mol) of trifluoromethanesulfonic acid anhydride are added dropwise at 0-5°C to a solution of 129 g (0.3 mol) of 3-(4-chloro-2-

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fluoro-5-hydroxyphenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole (approx. 70 %) (Example P9) in 1.2 litres of 1,4-dioxane. After being stirred for 30 minutes at 0-5°C and then at 22°C overnight, the reaction mixture is concentrated to approximately 1/3 of its original volume by evaporation, poured into a mixture of ice, water and hydrochloric acid, extracted with MTBE, washed with water and concentrated by evaporation. After purification of the crude product by silica gel chromatography, 87.7 g (70% of the theoretical yield) of the desired compound are isolated in the form of crystals having a melting point of 54-56°C.

Example P11: 3-(4-Chloro-2-fluoro-5-methoxycarbonylphenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole

A mixture of 40.7 g (0.1 mol) of 3-(4-chloro-2-fluoro-5-trifluoromethylsulfonyloxyphenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole (Example P10), 31 ml (0.22 mol) of triethylamine, 1.12 g (0.005 mol) of palladium(II) acetate and 2.06 g (0.005 mol) of 1,3-bis(diphenylphosphino)propane (Ph₂P(CH₂)₃PPh₂) in 300 ml of N,N-dimethylformamide and 215 ml of methanol is stirred at 70°C at a pressure of 5 bar of carbon monoxide for 2 hours. The solution is then concentrated by evaporation, the resulting residue is dissolved in tert-butyl methyl ether (MTBE), washed with 0.2N hydrochloric acid and water, concentrated by evaporation and purified by means of silica gel column chromatography. 23.4 g (71.3 % of the theoretical yield) of the desired compound are obtained in the form of crystals having a melting point of 82-83°C.

Example P12: 3-(4-Chloro-2-fluoro-5-propargyloxyphenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole

1 ml (0.0132 mol) of propargyl bromide are added dropwise at 20-25°C to a mixture of 3.15 g (0.011 mol) of 3-(4-chloro-2-fluoro-5-hydroxyphenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole (Example P9) and 2.5 g (0.018 mol) of potassium carbonate in 30 ml of N,N-dimethylformamide. After being stirred for 18 hours at 22°C, the mixture is poured into water, extracted with tert-butyl methyl ether (MTBE), washed with water and concentrated by evaporation. After recrystallisation in petroleum ether, 2.84 g (81.1% of the theoretical yield) of the desired compound are obtained in the form of crystals having a melting point of 76-78°C.

Example P13: 3-(4-Chloro-2-fluoro-5-propargyloxyphenyl)-4-methyl-5-(methylsulfinyl)-1-methyl-[1H]-pyrazole

A solution of 1.57 g (0.005 mol) of 50-60 % 3-chloroperbenzoic acid in 30 ml of dichloromethane is added dropwise at 0-5°C to a solution of 1.7 g (0.005 mol) of 3-(4-chloro-2-fluoro-5-propargyloxyphenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole (Example P12) in 20 ml of dichloromethane. After being stirred overnight at 22°C, the reaction mixture is washed with a dilute sodium hydrogen carbonate solution, then with water and concentrated by evaporation, and the resulting residue is recrystallised in petroleum ether/diethyl ether. 1.5 g (88.2 % of the theoretical yield) of the desired compound are isolated in the form of crystals having a melting point of 93-96°C.

Example P14: 3-(4-Chloro-2-fluoro-5-propargyloxyphenyl)-4-methyl-5-(methylsulfonyl)-1-methyl-[1H]-pyrazole

$$\begin{array}{c|c} & & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ &$$

A solution of 3.45 g (0.011 mol) of 50-60 % 3-chloroperbenzoic acid in 60 ml of dichloromethane is added dropwise at 0-5°C to a solution of 1.7 g (0.005 mol) of 3-(4-chloro-2-fluoro-5-propargyloxyphenyl)-4-methyl-5-(methylthio)-1-methyl-[1H]-pyrazole (Example P12) in 20 ml of dichloromethane. After being stirred overnight at 22°C, the reaction mixture is washed with a dilute sodium hydrogen carbonate solution, then with water and concentrated by evaporation, and the resulting residue is recrystallised in diethyl ether. 1.4 g (78.6 % of the theoretical yield) of the desired compound are obtained in the form of crystals having a melting point of 123-124°C.

In an analogous manner or by means of known methods it is also possible to prepare the compounds listed in the Tables which follow.

Table 1: Compounds of formula la

$$CI \xrightarrow{F} CH_3 SCH_3$$

$$CH_3$$

$$CH_3$$

$$CH_3$$

Comp.No.	R ₆	M.p.
1.1	Н	Oil
12	NH ₂	55-60
1.3	NO ₂	
1.4	Br	
1.5	I	
1.6	CN	
1.7	OCH₃	64-70
1.8	N(SO ₂ CH ₃) ₂	
1.9	NHSO₂CH₃	142-144
1.10	OC ₃ H ₇ (iso)	oil
1.11	O-propargyl	72-75
1.12	осн(сн₃)с≡сн	oil
1.13	O-phenyl	
1.14	O-2-pyridyl	
1.15	O-2-pyrimidinyl	
1.16	OCH ₂ COOCH ₂ CH ₃	95-97
1.17	OCH ₂ CH ₂ OCH ₃	
1.18	OCH ₂ CH ₂ SCH ₂ CH ₃	
1.19	OCH₂COOCH₃	
1.20	OCH ₂ COOC ₅ H ₁₁ (n)	

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Comp.No.	R ₆	M.p.
1.21	OCH₂COO-benzyl	
1.22	OCH(CH ₃)COObenzyl (S)	
1.23	OCH(CH ₃)COObenzyl (R)	
1.24	OCH(CH ₃)COObenzyl (R,S)	
1.25	SC ₃ H ₇ (iso)	
1.26	SH	solid
1.27	SCH₂COOCH₃	oil
1.28	SCH ₂ COOC ₂ H ₅	
1.29	SCH(CH ₃)COObenzyl (S)	
1.30	SCH(CH ₃)COObenzyl (R)	
1.31	SCH(CH₃)COObenzyl (R,S)	- I.
1.32	SCH₂COObenzyl	
1.33	SO ₂ CI	solid
1.34	SO₂CH₃	Solid
1.35	SO₂NHCH₃	
1.36	СООН	177-179
1.37	COOCH ₃	82-83
1.38	COOC ₃ H ₇ (iso)	oil
1.39	COOC(CH ₃) ₂ COOH	
1.40	COOC(CH₃)₂COO-allyl	amorphous
1.41	COOC(CH ₃) ₂ COOCH ₃	oil
1.42	COOC(CH ₃)₂COOethyl	70.77
1.43	COOC(CH ₃) ₂ CONH-allyl	72-75
1.44	CH ₂ CHCICOOethyl	
1.45	CH ₂ CH=CH ₂	oil
1.46	CH ₂ CH ₂ CH ₃	oil
1.47	CH ₂ CH ₂ CF ₃	
1.48		
	OCH(CH₃)COOC₂H₅(R)	

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Сотр. No.	R ₆	M.p.
1.49	OCH(CH ₃)COOC₂H₅(S)	
1.50	OCH(CH₃)COOC₂H₅(R,S)	oil
1.51	CH₂CHCICOOH	170-172
1.52	CH₂CHCICOOCH₃	
1.53	CH₂CHCICOOC₃H₂(iso)	
1.54	CH₂CHCICONHallyl	
1.55	CH₂C(CH₃)ClCOOH	
1.56	CH ₂ C(CH ₃)ClCOOC ₂ H ₅	oil
1.57	CH₂C(CH₃)CICOOEt	
1.58	CH₂C(CH₃)CICONHEt	
1.59	CH₂CH₂COOH	
1.60	CH₂CH₂COOCH₃	
1.61	CH₂CH₂COOEt	
1.62	CHCICHCICOOH	
1.63	CHCICHCICOOCH₃	
1.64	CHCICHCICOOEt	
1.65	CH₂CH(OCH₃)COOH	
1.66	CH₂CH(OCH₃)COOCH₃	
1.67	CH₂CH(OCH₃)COOEt	
1.68	CH₂CH(SCH₃)COOH	
1.69	CH₂CH(SCH₃)COOCH₃	
1.70	CH₂CH(SCH₃)COOEt	oil
1.71	CH=CHCOOH	
1.72	CH=CHCOOCH₃	
1.73	CH=CHCOOEt	
1.74	CH=CCICOOH	·
1.75	CH=CCICOOCH ₃	
1.76	COOEt	

Comp.No.	R ₆	M.p.
1.77	CONH₂	
1.78	-C(O)O-CH ₂	55-57
1.79	CONHSO₂CH₃	
1.80	COOCH₂COOH	
1.81	COOCH₂COOCH₃	
1.82	COOCH(CH ₃)COOH	
1.83	COOCH(CH ₃)COOCH ₃	
1.84	COOCH(CH₃)CH₂COOH	
1.85	COOCH(CH ₃)CH ₂ COOCH ₃	
1.86	COOC(CH ₃)₂CN	
1.87	COOCH₂CH₂OCH₃	
1.88	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
1.89	COOC(CH ₃) ₂ -C(O)O-CH ₂	59-62
1.90	COOC(CH ₃) ₂ COOCH ₂ PHENYL	oil
1.91	соосн₂с≡сн	
1.92	COOC(CH ₃) ₂ COOCH ₂ C==CH	
1.93	соосн(сн³)с≡сн	
1.94	COOC(CH₃)₂COCH₃	
1.95	NHallyl	
1.96	N(COCH₃)allyl	
1.97	N(Et)SO ₂ CH ₃	
1.98	N(allyl)SO₂CH₃	83-85
1.99	N(allyl)SO₂Et	65-68





Comp.No.	R ₆	M.p.
1.100	SO ₂ N(CH ₃) ₂	<u> </u>
1.101	SO ₂ NH ₂	
1.102	SO₂NHCOCH ₃	
1.103	ОН	oil
1.104	OEt	
1.105	Oallyl	
1.106	OCH ₂ C≡CCH ₃	
1.107	OCH(CH ₃)CH=CH ₂	
1.108	OCH₂CH₂OCH₂CH₃	
1.109	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	oil
1.110	OCH ₂	86-89
1.111	OCH ₂ CH ₂ NHCH ₃	
1.112	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
1.113	OCH ₂ CH ₂ COOH	
1.114	OC(CH ₃) ₂ COOH	
1.115	OC(CH ₃)₂COOCH ₃	
1.116	OC(CH ₃) ₂ COOEt	
1.117	OCH₂COOH	
1.118	OSO₂CH₃	
1.119	OSO ₂ CF ₃	54-56
1.120	CH₂CHCICOOC₂H₅	
1.121	CH ₂ CHCICON(C ₂ H ₅) ₂	
1.122	CH₂CHCICONHOH	
1.123	CH₂CHCICOOCH₂C6H5	
1.124	CH ₂ CH(CH ₃)COOH	
1.125	CH₂CH(CH₃)COOC₂H₅	

Comp.No.	R ₆	M.p.
1.126	-COOCH ₂	
1.127	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
1.128	-COOC(CH ₃) ₂ COOCH ₂	
1.129	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
1.130	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
1.131	OCH ₂	91-93
1.132	NHSO₂C₂H₅	121-124
1.133	NHCH₂C≡CH	103-105





Table 2: Compounds of formula lb

$$CI \xrightarrow{F} CH_3 \xrightarrow{SOCH_3} CH_3$$

$$CH_3 \xrightarrow{CH_3} CH_3$$

Comp.No.	R ₆	M.p.
2.1	Н	113-115
2.2	NH ₂	solid
2.3	NO ₂	128-130
2.4	Br	
2.5	ı	
2.6	CN	
2.7	OCH₃	89-92
2.8	N(SO₂CH₃)₂	
2.9	NHSO₂CH₃	
2.10	OC ₃ H ₇ (iso)	oil
2.11	O-propargyl	93-96
2.12	осн(сн₃)с≡сн	resin
2.13	O-phenyl	
2.14	O-2-pyridyl	
2.15	O-2-pyrimidinyl	
2.16	OCH ₂ COOC ₂ H ₅	95-98
2.17	OCH ₂ COOC ₅ H ₁₁ (n)	
2.18	OCH₂COO-benzyl	
2.19	OCH(CH ₃)COObenzyl (S)	
2.20	OCH(CH ₃)COObernzyl (R)	

Comp.No.	R ₆	M.p.
2.21	OCH(CH ₃)COObenzyl (R,S)	
2.22	SC ₃ H ₇ (iso)	
2.23	SH	
2.24	SCH₂COOCH₃	
2.25	SCH₂COOC₂H₅	
2.26	SCH(CH ₃)COObenzyl (S)	
2.27	SCH(CH ₃)COObenzyl (R)	
2.28	SCH(CH ₃)COObenzyl (R,S)	
2.29	SCH₂COObenzyl	
2.30	SO₂CI	
2.31	SO₂CH₃	
2.32	SO₂NHCH₃	
2.33	СООН	138-164
2.34	COOCH ₃	oil
2.35	COOC ₃ H ₇ (iso)	oil
2.36	COOC(CH ₃) ₂ COOH	153-166
2.37	COOC(CH ₃) ₂ COO-allyl	. oil
2.38	COOC(CH₃)₂COOCH₃	
2.39	COOC(CH₃)₂COOethyl	oil
2.40	COOC(CH₃)₂CONH-allyl	
2.41	CH₂CHClCOOethyl	resin
2.42	CH₂CH=CH₂	99-101
2.43	CH₂CH₂CH₃	
2.44	CH₂CH₂CF₃	
2.45	OCH(CH ₃)COOC ₂ H ₅ (R)	
2.46	OCH(CH ₃)COOC ₂ H ₅ (S)	
2.47	OCH(CH ₃)COOC ₂ H ₅ (R,S)	oil
2.48	CH₂CHCICOOH	





Comp.No.	R ₆	M.p.
2.49	CH₂CHCICOOCH₃	m.p.
2.50	CH₂CHCICOOC₃H₂(iso)	
2.51	CH₂CHCICONHallyl	
2.52	CH₂C(CH₃)CICOOH	
2.53	CH₂C(CH₃)CICOOCH₃	
2.54	CH₂C(CH₃)CICOOEt	oil
2.55	CH₂C(CH₃)CICONHEt	Oil
2.56	CH₂CH₂COOH	
2.57	CH₂CH₂COOCH₃	
2.58	CH₂CH₂COOEt	
2.59	CHCICHCICOOH	
2.60	CHCICHCICOOCH ₃	
2.61	CHCICHCICOOEt	·
2.62		
2.63	CH₂CH(OCH₃)COOH	
	CH ₂ CH(OCH ₃)COOCH ₃	
2.64	CH₂CH(OCH₃)COOEt	
2.65	CH₂CH(SCH₃)COOH	·
2.66	CH ₂ CH(SCH ₃)COOCH ₃	
2.67	CH₂CH(SCH₃)COOEt	
2.68	CH=CHCOOH	
2.69	CH=CHCOOCH₃	
2.70	CH=CHCOOEt	
2.71	CH=CCICOOH	
2.72	CH=CCICOOCH₃	
2.73	COOEt	
2.74	CONH ₂	
2.75	-C(O)O-CH ₂	68-71

Comp.No.	R ₆	M.p.
2.76	CONHSO₂CH₃	
2.77	COOCH₂COOH	
2.78	COOCH₂COOCH₃	
2.79	COOCH(CH₃)COOH	
2.80	COOCH(CH ₃)COOCH ₃	
2.81	COOCH(CH₃)CH₂COOH	
2.82	COOCH(CH₃)CH₂COOCH₃	
2.83	COOC(CH ₃)₂CN	
2.84	COOCH₂CH₂OCH₃	· .
2.85	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
2.86	COOC(CH ₃) ₂ -C(O)O-CH ₂	· ·
2.87	COOC(CH ₃) ₂ COOCH ₂ PHENYL	91-93
2.88	СООСН₂С≡СН	
2.89	COOC(CH3)2COOCH2C=CH	
2.90	соосн(сн₃)с≡сн	
2.91	COOC(CH ₃) ₂ COCH ₃	
2.92	NHallyl	
2.93	N(COCH ₃)allyl	
2.94	N(Et)SO ₂ CH ₃	
2.95	N(allyl)SO₂CH₃	133-136
2.96	N(allyl)SO₂Et	solid
2.97	SO ₂ N(CH ₃) ₂	
2.98	SO ₂ NH ₂	
2.99	SO₂NHCOCH₃	

Comp.No.	R ₆	M.p.
2.100	ОН	
2.101	OEt	
2.102	Oallyl	
2.103	OCH₂C≡CCH₃	
2.104	OCH(CH ₃)CH=CH ₂	
2.105	OCH ₂ CH ₂ OCH ₂ CH ₃	
2.106	OCH₂CH₂OCH₂CH₂OCH₃	oil
2.107	OCH ₂	96-99
2.108	OCH₂CH₂NHCH₃	
2.109	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
2.110	OCH₂CH₂COOH	
2.111	OC(CH₃)₂COOH	
2.112	OC(CH ₃) ₂ COOCH ₃	
2.113	OC(CH ₃) ₂ COOEt	
2.114	OCH₂COOH	
2.115	OSO ₂ CH ₃	
2.116	OSO ₂ CF ₃	
2.117	CH₂CHCICOOC₂H₅	
2.118	CH ₂ CHCICON(C ₂ H ₅) ₂	
2.119	CH₂CHCICONHOH	
2.120	CH₂CHCICOOCH₂C ₆ H ₅	
2.121	CH₂CH(CH₃)COOH	
2.122	CH ₂ CH(CH ₃)COOC ₂ H ₅	
2.123	-COOCH ₂ —	
2.124	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	(5)

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Comp.No.	R ₆	M.p.
2.125	-соос(сн ₃) ₂ соосн ₂ —	
2.126	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
2.127	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
2.128	OCH ₂	96-98

Table 3: Compounds of formula Ic

$$CI \longrightarrow \begin{array}{c} F & CH_3 \\ \hline N-N & CH_3 \end{array}$$
 (Ic)

Comp.No.	R ₆	M.p.
3.1	Н	113-115
3.2	NH ₂	
3.3	NO₂	
3.4	Br	
3.5		
3.6	CN	
3.7	OCH₃	146-149
3.8	N(SO ₂ CH ₃) ₂	
3.9	NHSO₂CH₃	199-201
3.10	OC ₃ H ₇ (iso)	100-103
3.11	O-propargyl	123-124
3.12	осн(сн₃)с≡сн	oil .
3.13	O-phenyl	
3.14	O-2-pyridyl	
3.15	O-2-pyrimidinyl	
3.16	OCH ₂ COOC ₂ H ₅	123-126
3.17	OCH₂COOCH₃	
3.18	OCH₂COO-benzyl	
3.19	OCH(CH₃)COObenzyl (S)	
3.20	OCH(CH₃)COObenzyl (R)	
3.21	OCH(CH₃)COObenzyl (R,S)	

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3.22 SC ₃ H ₇ (iso) 3.23 SH 3.24 SCH ₂ COOCH ₃ 3.25 SCH ₂ COOC ₂ H ₅ 3.26 SCH(CH ₃)COObenzyl (S) 3.27 SCH(CH ₃)COObenzyl (R) 3.28 SCH(CH ₃)COObenzyl (R,S) 3.29 SCH ₂ COObenzyl 3.30 SO ₂ Cl 3.31 SO ₂ CH ₃ 3.32 SO ₂ NHCH ₃ 3.33 COOH 237-240 3.34 COOCH ₃ 3.35 COOC ₃ H ₇ (iso) 92-94	
3.24 SCH2COOCH3 3.25 SCH2COOC2H5 3.26 SCH(CH3)COObenzyl (S) 3.27 SCH(CH3)COObenzyl (R) 3.28 SCH(CH3)COObenzyl (R,S) 3.29 SCH2COObenzyl 3.30 SO2Cl 3.31 SO2CH3 3.32 SO2NHCH3 3.33 COOH 237-240 3.34 COOCH3 3.35 COOC3H7(iso)	
3.25 SCH ₂ COOC ₂ H ₅ 3.26 SCH(CH ₃)COObenzyl (S) 3.27 SCH(CH ₃)COObenzyl (R) 3.28 SCH(CH ₃)COObenzyl (R,S) 3.29 SCH ₂ COObenzyl 3.30 SO ₂ Cl 3.31 SO ₂ CH ₃ 3.32 SO ₂ NHCH ₃ 3.33 COOH 3.34 COOCH ₃ 94-96 3.35 COOC ₃ H ₇ (iso) 92-94	
3.26 SCH(CH ₃)COObenzyl (S) 3.27 SCH(CH ₃)COObenzyl (R) 3.28 SCH(CH ₃)COObenzyl (R,S) 3.29 SCH ₂ COObenzyl 3.30 SO ₂ Cl 3.31 SO ₂ CH ₃ 3.32 SO ₂ NHCH ₃ 3.33 COOH 237-240 3.34 COOCH ₃ 94-96 3.35 COOC ₃ H ₇ (iso)	
3.27 SCH(CH ₃)COObenzyl (R) 3.28 SCH(CH ₃)COObenzyl (R,S) 3.29 SCH ₂ COObenzyl 3.30 SO ₂ Cl 3.31 SO ₂ CH ₃ 3.32 SO ₂ NHCH ₃ 3.33 COOH 237-240 3.34 COOCH ₃ 3.35 COOC ₃ H ₇ (iso)	
3.28 SCH(CH ₃)COObenzyl (R,S) 3.29 SCH ₂ COObenzyl 3.30 SO ₂ Cl 3.31 SO ₂ CH ₃ 3.32 SO ₂ NHCH ₃ 3.33 COOH 237-240 3.34 COOCH ₃ 94-96 3.35 COOC ₃ H ₇ (iso)	
3.29 SCH ₂ COObenzyl 3.30 SO ₂ Cl 3.31 SO ₂ CH ₃ 3.32 SO ₂ NHCH ₃ 3.33 COOH 237-240 3.34 COOCH ₃ 94-96 3.35 COOC ₃ H ₇ (iso) 92-94	
3.30 SO ₂ Cl 3.31 SO ₂ CH ₃ 3.32 SO ₂ NHCH ₃ 3.33 COOH 237-240 3.34 COOCH ₃ 94-96 3.35 COOC ₃ H ₇ (iso) 92-94	
3.31 SO ₂ CH ₃ 3.32 SO ₂ NHCH ₃ 3.33 COOH 237-240 3.34 COOCH ₃ 94-96 3.35 COOC ₃ H ₇ (iso) 92-94	
3.32 SO₂NHCH₃ 3.33 COOH 237-240 3.34 COOCH₃ 94-96 3.35 COOC₃H₂(iso) 92-94	
3.33 COOH 237-240 3.34 COOCH ₃ 94-96 3.35 COOC ₃ H ₇ (iso) 92-94	
3.34 COOCH ₃ 94-96 3.35 COOC ₃ H ₇ (iso) 92-94	
3.35 COOC ₃ H ₇ (iso) 92-94	
92-94	
3.36 COOC(CH ₃) ₂ COOH 167-169	
3.37 COOC(CH ₃)₂COO-allyl oil	
3.38 COOC(CH ₃)₂COOCH ₃	
3.39 COOC(CH₃)₂COOethyl oil	
3.40 COOC(CH₃)₂CONH-allyl	
3.41 CH₂CHClCOOethyl resin	
3.42 CH ₂ CH=CH ₂ 98-100	
3.43 CH ₂ CH ₂ CH ₃	
3.44 CH₂CH₂CF₃	
3.45 OCH(CH ₃)COOC ₂ H ₅ (R)	
3.46 OCH(CH ₃)COOC ₂ H ₅ (S)	
3.47 OCH(CH ₃)COOC ₂ H ₅ (R,S) oil	
3.48 CH₂CHCICOOH	
3.49 CH₂CHCICOOCH₃	
3.50 CH₂CHCICOOC₃H₂(iso)	Į.





2 51	CHOLOGONIA	
3.51	CH₂CHClCONHallyl	
3.52	CH₂C(CH₃)CICOOH	
3.53	CH ₂ C(CH ₃)CICOOCH ₃	
3.54	CH₂C(CH₃)CICOOEt	oil
3.55	CH₂C(CH₃)CICONHEt	
3.56	CH₂CH₂COOH	
3.57	CH₂CH₂COOCH₃	
3.58	CH₂CH₂COOEt	
3.59	CHCICHCICOOH	
3.60	CHCICHCICOOCH3	
3.61	CHCICHCICOOEt	
3.62	CH₂CH(OCH₃)COOH	
3.63	CH₂CH(OCH₃)COOCH₃	
3.64	CH₂CH(OCH₃)COOEt	
3.65	CH₂CH(SCH₃)COOH	
3.66	CH₂CH(SCH₃)COOCH₃	
3.67	CH₂CH(SCH₃)COOEt	
3.68	СН=СНСООН	
3.69	CH=CHCOOCH₃	
3.70	CH=CHCOOEt	
3.71	CH=CCICOOH	
3.72	CH=CCICOOCH ₃	
3.73	COOEt	·
3.74	CONH₂	
3.75	-C(O)O-CH ₂	110-112
3.76	CONHSO₂CH ₃	
3.77	COOCH₂COOH	
3.78	COOCH₂COOCH₃	

3.79	Loopourous	· •
	COOCH(CH3)COOH	
3.80	COOCH(CH ₃)COOCH ₃	
3.81	COOCH(CH₃)CH₂COOH	
3.82	COOCH(CH ₃)CH ₂ COOCH ₃	
3.83	COOC(CH₃)₂CN	
3.84	COOCH₂CH₂OCH₃	
3.85	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
3.86	COOC(CH ₃) ₂ -C(O)O-CH ₂	oil
3.87	COOC(CH ₃) ₂ COOCH ₂ PHENYL	83-86
3.88	COOCH₂C=CH	
3.89	COOC(CH ₃) ₂ COOCH ₂ C _E CH	
3.90	COOCH(CH₃)C⊫CH	
3.91	COOC(CH ₃) ₂ COCH ₃	
3.92	NHallyl	
3.93	N(COCH₃)allyl	
3.94	N(Et)SO ₂ CH ₃	
3.95	N(allyl)SO₂CH₃	120-122
3.96	N(allyl)SO₂Et	105-107
3.97	SO₂N(CH ₃)₂	
3.98	SO ₂ NH ₂	
3.99	SO₂NHCOCH₃	
3.100	ОН	
3.101	OEt	
3.102	Oallyl	
3.103	OCH ₂ C≡CCH ₃	
3.104	OCH(CH₃)C=CH₂	
3.105	OCH ₂ CH ₂ OCH ₂ CH ₃	
3.106	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	51-55

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3.107	0	125-127
	OCH ₂	123-127
3.108	OCH₂CH₂NHCH₃	
3.109	OCH₂CH₂N(CH₃)COCH₃	
3.110	OCH ₂ CH ₂ COOH	
3.111	OC(CH ₃)₂COOH	
3.112	OC(CH ₃)₂COOCH ₃	
3.113	OC(CH ₃)₂COOEt	
3.114	OCH₂COOH	
3.115	OSO₂CH ₃	
3.116	OSO ₂ CF ₃	
3.117	CH₂CHCICOOC₂H₅	
3.118	CH ₂ CHCICON(C ₂ H ₅) ₂	
3.119	CH₂CHCICONHOH	
3.120	CH₂CHCICOOCH₂C ₆ H ₅	
3.121	CH₂CH(CH₃)COOH	
3.122	CH ₂ CH(CH ₃)COOC ₂ H ₅	· ·
3.123	-COOCH ₂	
3.124	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
3.125	-cooc(ch ₃) ₂ соосн ₂ —	
3.126	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
3.127	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
3.128	OCH ₂ —	108-109

Table 4: Compounds of formula Id

$$CI \longrightarrow N \longrightarrow CH_3$$
 (Id)

Comp.No.	R ₆	M.p.
4.1	Н	69-71
4.2	NH₂	oil
4.3	NO ₂	
4.4	Br	
4.5	1	74-77
4.6	CN	
4.7	OCH ₃	
4.8	N(SO ₂ CH ₃) ₂	
4.9	NHSO₂CH₃	
4.10	OC ₃ H ₇ (iso)	
4.11	O-propargyl	
4.12	OCH(CH₃)C≡CH	
4.13	OCH₂COOCH₂CH₃	
4.14	OCH ₂ CH ₂ OCH ₃	
4.15	OCH ₂ CH ₂ SCH ₂ CH ₃	
4.16	OCH₂COOCH₃	
4.17	OCH₂COOC₅H₁₁ (n)	
4.18	OCH₂COO-benzyl	
4.19	OCH(CH₃)COObenzyl (S)	
4.20	OCH(CH₃)COObenzyl (R)	
4.21	OCH(CH₃)COObenzyl (R,S)	

(3)

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Comp.No.	R_6	M.p.
4.22	SC ₃ H ₇ (iso)	
4.23	SH	
4.24	SCH₂COOCH₃	
4.25	SCH ₂ COOC ₂ H ₅	
4.26	SCH(CH₃)COObenzyl (S)	
4.27	SCH(CH₃)COObenzyl (R)	
4.28	SCH(CH₃)COObenzyl (R,S)	
4.29	SCH ₂ COObenzyl	
4.30	SO₂CI	
4.31	SO₂CH₃	
4.32	SO₂NHCH₃	
4.33	СООН	150-151
4.34	COOCH₃	60-61
4.35	COOC ₃ H ₇ (iso)	oil
4.36	COOC(CH ₃) ₂ COOH	184-188
4.37	COOC(CH ₃) ₂ COO-allyl	
4.38	COOC(CH ₃) ₂ COOCH ₃	
4.39	COOC(CH₃)₂COOethyl	52-55
4.40	COOC(CH ₃) ₂ CONH-allyl	
4.41	CH₂CHClCOOethyl	
4.42	CH ₂ CH=CH ₂	
4.43	CH ₂ CH ₂ CH ₃	
4.44	CH ₂ CH ₂ CF ₃	
4.45	OCH(CH ₃)COOC ₂ H ₅ (R)	
4.46	OCH(CH ₃)COOC ₂ H ₅ (S)	
4.47	OCH(CH₃)COOC₂H₅(R,S)	
4.48	CH₂CHCICOOH	
4.49	CH₂CHCICOOCH3	





Comp.No.	R ₆	
4.50	CH ₂ CHCICOOC ₃ H ₇ (iso)	M.p.
4.51		
	CH₂CHCICONHaliyl	
4.52	CH₂C(CH₃)CICOOH	
4.53	CH ₂ C(CH ₃)CICOOCH ₃	
4.54	CH₂C(CH₃)CICOOEt	
4.55	CH₂C(CH₃)CICONHEt	
4.56	CH₂CH₂COOH	
4.57	CH₂CH₂COOCH₃	
4.58	CH₂CH₂COOEt	
4.59	СНСІСНСІСООН	
4.60	CHCICHCICOOCH ₃	
4.61	CHCICHCICOOEt	
4.62	CH ₂ CH(OCH ₃)COOH	
4.63	CH ₂ CH(OCH ₃)COOCH ₃	
4.64	CH₂CH(OCH₃)COOEt	
4.65	CH₂CH(SCH₃)COOH	
4.66	CH ₂ CH(SCH ₃)COOCH ₃	
4.67	CH ₂ CH(SCH ₃)COOEt	
4.68	CH=CHCOOH	
4.69		
	CH=CHCOOCH ₃	
4.70	CH=CHCOOEt	
4.71	CH=CCICOOH	
4.72	CH=CCICOOCH3	
4.73	COOEt	· ·
1.74	CONH ₂	
1.75	-C(O)O-CH ₂	
.76	CONHSO₂CH₃	





Comp.No.	R ₆	M.p.
4.77	COOCH₂COOH	
4.78	COOCH₂COOCH₃	
4.79	COOCH(CH ₃)COOH	
4.80	COOCH(CH ₃)COOCH ₃	
4.81	COOCH(CH₃)CH₂COOH	
4.82	COOCH(CH ₃)CH ₂ COOCH ₃	
4.83	COOC(CH ₃) ₂ CN	
4.84	COOCH ₂ CH ₂ OCH ₃	
4.85	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
4.86	COOC(CH ₃) ₂ -C(O)O-CH ₂	
4.87	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
4.88	COOCH₂C≡CH	
4.89	COOC(CH ₃) ₂ COOCH ₂ C≡CH	
4.90	COOCH(CH₃)C≡CH	
4.91	COOC(CH ₃)₂COCH ₃	
4.92	NHallyl	
4.93	N(COCH ₃)allyl	
4.94	N(Et)SO₂CH₃	
4.95	N(allyl)SO ₂ CH ₃	
4.96	N(allyI)SO ₂ Et	
4.97	SO₂N(CH₃)₂	
1.98	SO₂NH₂	
1.99	SO₂NHCOCH₃	
1.100	ОН	
1.101	OEt .	
.102	Oallyl	
.103	OCH₂C≡CCH₃	

Comp.No.	R ₆	M.p.
4.104	OCH(CH ₃)CH=CH ₂	
4.105	OCH ₂ CH ₂ OCH ₂ CH ₃	·
4.106	OCH₂CH₂OCH₂CH₃OCH₃	
4.107	OCH ₂	·
4.108	OCH₂CH₂NHCH₃	
4.109	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
4.110	OCH ₂ CH ₂ COOH	
4.111	OC(CH ₃)₂COOH	
4.112	OC(CH ₃) ₂ COOCH ₃	
4.113	OC(CH ₃) ₂ COOEt	
4.114	OCH₂COOH	
4.115	OSO ₂ CH ₃	
4.116	OSO ₂ CF ₃	
4.117	CH ₂ CHClCOOC ₂ H ₅	
4.118	CH₂CHCICON(C₂H₅)₂	
4.119	CH₂CHCICONHOH	
4.120	CH₂CHCICOOCH₂C ₆ H ₅	
4.121	CH₂CH(CH₃)COOH	
4.122	CH ₂ CH(CH ₃)COOC ₂ H ₅	
4.123	-COOCH ₂	
4.124	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
4.125	-COOC(CH ₃)₂COOCH₂	
4.126	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
4.127	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	





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Comp.No.	R ₆	M.p.
4.128	OCH ₂	

Table 5: Compounds of formula le

$$CI \longrightarrow N \longrightarrow N \longrightarrow CH_3$$
 (Ie)

Comp.No.	R_6	M.p.
5.1	Н	120-121
5.2	NH ₂	136-139
5.3	NO ₂	151-153
5.4	Br	
5.5		83-86
5.6	CN	
5.7	OCH ₃	
5.8	N(SO ₂ CH ₃) ₂	
5.9	NHSO₂CH₃	
5.10	OC ₃ H ₇ (iso)	
5.11	O-propargyl	
5.12	OCH(CH ₃)C≡CH	
5.13	OCH ₂ COOCH ₂ CH ₃	
5.14	OCH ₂ CH ₂ OCH ₃	
5.15	OCH ₂ CH ₂ SCH ₂ CH ₃	
5.16	OCH₂COOCH₃	
5.17	OCH₂COOC₅H₁₁ (n)	· ·
5.18	OCH₂COO-benzyl	
5.19	OCH(CH ₃)COObenzyl (S)	
5.20	OCH(CH₃)COObenzyl (R)	
5.21	OCH(CH ₃)COObenzyi (R,S)	



Comp.No.	R ₆	M.p.
5.22	SC ₃ H ₇ (iso)	
5.23	SH	
5.24	SCH₂COOCH₃	
5.25	SCH₂COOC₂H₅	
5.26	SCH(CH₃)COObenzyl (S)	
5.27	SCH(CH₃)COObenzyl (R)	
5.28	SCH(CH ₃)COObenzyl (R,S)	
5.29	SCH₂COObenzyl	
5.30	SO₂CI	· ·
5.31	SO₂CH₃	
5.32	SO₂NHCH₃	
5.33	соон	161-165
5.34	соосн₃	95-97
5.35	COOC ₃ H ₇ (iso)	117-120
5.36	COOC(CH ₃)₂COOH	
5.37	COOC(CH ₃) ₂ COO-allyl	
5.38	COOC(CH ₃)₂COOCH ₃	
5.39	COOC(CH ₃) ₂ COOethyl	oil
5.40	COOC(CH₃)₂CONH-allyl	
5.41	CH₂CHClCOOethyl	
5.42	CH₂CH=CH₂	
5.43	CH₂CH₂CH₃	
5.44	CH ₂ CH ₂ CF ₃	
5.45	OCH(CH ₃)COOC ₂ H ₅ (R)	
5.46	OCH(CH ₃)COOC ₂ H ₅ (S)	
5.47	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
5.48	CH₂CHCICOOH	
5.49	CH₂CHCICOOCH3	

Comp.No.	R ₆	M.p.
5.50	CH₂CHCICOOC₃H₂(iso)	·
5.51	CH₂CHCICONHallyl	
5.52	CH₂C(CH₃)CICOOH	
5.53	CH₂C(CH₃)CICOOCH₃	
5.54	CH₂C(CH₃)CICOOEt	
5.55	CH₂C(CH₃)CICONHEt	
5.56	CH₂CH₂COOH	
5.57	CH₂CH₂COOCH₃	
5.58	CH₂CH₂COOEt	
5.59	СНСІСНСІСООН	· · · · · · · · · · · · · · · · · · ·
5.60	CHCICHCICOOCH ₃	
5.61	CHCICHCICOOEt	
5.62	CH₂CH(OCH₃)COOH	
5.63	CH₂CH(OCH₃)COOCH₃	
5.64	CH₂CH(OCH₃)COOEt	
5.65	CH₂CH(SCH₃)COOH	
5.66	CH₂CH(SCH₃)COOCH₃	
5.67	CH₂CH(SCH₃)COOEt	
5.68	СН=СНСООН	
5.69	CH=CHCOOCH₃	
5.70	CH=CHCOOEt	
5.71	CH=CCICOOH	
5.72	CH=CCICOOCH ₃	
5.73	COOEt	
5.74	CONH₂	
5.75	-C(O)OCH ₂	
5.76	CONHSO₂CH₃	

Comp.No.	R ₆	M.p.
5.77	COOCH₂COOH	
5.78	COOCH₂COOCH₃	
5.79	COOCH(CH3)COOH	
5.80	COOCH(CH ₃)COOCH ₃	
5.81	COOCH(CH₃)CH₂COOH	
5.82	COOCH(CH₃)CH₂COOCH₃	
5.83	COOC(CH ₃) ₂ CN	
5.84	COOCH ₂ CH ₂ OCH ₃	
5.85	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
5.86	COOC(CH ₃) ₂ -C(O)O-CH ₂	
5.87	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
5.88	COOCH₂C≡CH	
5.89	COOC(CH ₃) ₂ COOCH ₂ C≡CH	
5.90	COOCH(CH₃)C≡CH	
5.91	COOC(CH ₃)₂COCH ₃	
5.92	NHallyl	
5.93	N(COCH₃)allyl	
5.94	N(Et)SO₂CH ₃	
5.95	N(allyl)SO ₂ CH ₃	
5.96	N(allyl)SO₂Et	
5.97	SO ₂ N(CH ₃) ₂	
5.98	SO ₂ NH ₂	
5.99	SO₂NHCOCH ₃	
5.100	OH .	
5.101	OEt	
5.102	Oallyl	
5.103	OCH ₂ C≡CCH ₃	

Comp.No.	R ₆	M.p.
5.104	OCH(CH ₃)CH=CH ₂	
5.105	OCH ₂ CH ₂ OCH ₂ CH ₃	
5.106	OCH₂CH₂OCH₂CH₂OCH₃	
5.107	OCH ₂	
5.108	OCH₂CH₂NHCH₃	
5.109	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
5.110	OCH₂CH₂COOH	
5.111	OC(CH₃)₂COOH	
5.112	OC(CH ₃) ₂ COOCH ₃	
5.113	OC(CH ₃) ₂ COOEt	
5.114	OCH₂COOH	
5.115	OSO ₂ CH ₃	
5.116	OSO ₂ CF ₃	
5.117	CH2CHCICOOC2H5	·
5.118	CH ₂ CHCICON(C ₂ H ₅) ₂	· ·
5.119	CH ₂ CHCICONHOH	
5.120	CH ₂ CHCICOOCH ₂ C ₆ H ₅	
5.121	CH₂CH(CH₃)COOH	
5.122	CH ₂ CH(CH ₃)COOC ₂ H ₅	
5.123	-COOCH ₂ —	
.124	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
.125	-COOC(CH ₃) ₂ COOCH ₂	
.126	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
.127	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	





Comp.No.	R_6	M.p.
5.128	OCH ₂	

Table 6: Compounds of formula If

$$CI \xrightarrow{CH_3} SO_2CH_3$$

$$CH_3$$

$$CH_3$$

$$CH_3$$

$$CH_3$$

Comp.No.	R ₆	M.p.
6.1	Н	122-124
6.2	NH₂	140-144
6.3	NO ₂	175-176
6.4	Br	
6.5	I	155-159
6.6	CN	· ·
6.7	OCH₃	
6.8	N(SO ₂ CH ₃) ₂	
6.9	NHSO₂CH₃	
6.10	OC ₃ H ₇ (iso)	
6.11	O-propargyl	
6.12	OCH(CH ₃)C≡CH	
6.13	OCH ₂ COOCH ₂ CH ₃	
6.14	OCH ₂ CH ₂ OCH ₃	
6.15	OCH ₂ CH ₂ SCH ₂ CH ₃	
6.16	OCH ₂ COOCH ₃	
6.17	OCH ₂ COOC ₅ H ₁₁ (n)	
6.18	OCH₂COO-benzyl	
6.19	OCH(CH ₃)COObenzyl (S)	
6.20	OCH(CH₃)COObenzyl (R)	
6.21	OCH(CH ₃)COObenzyl (R,S)	

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Comp.No.	R ₆	M.p.
6.22	SC ₃ H ₇ (iso)	
6.23	SH	
6.24	SCH₂COOCH₃	
6.25	SCH₂COOC₂H₅	
6.26	SCH(CH ₃)COObenzyl (S)	
6.27	SCH(CH ₃)COObenzyl (R)	
6.28	SCH(CH₃)COObenzyl (R,S)	
6.29	SCH₂COObenzyl	
6.30	SO₂CI	
6.31	SO₂CH₃	
6.32	SO₂NHCH₃	
5.33	СООН	225-227
6.34	COOCH3	104-106
6.35	COOC ₃ H ₇ (iso)	98-99
3.36	COOC(CH₃)₂COOH	173-177
5.37	COOC(CH₃)₂COO-allyl	173-177
5.38	COOC(CH ₃) ₂ COOCH ₃	
3.39	COOC(CH₃)₂COOethyl	90-92
.40	COOC(CH₃)₂CONH-allyl	90-92
.41	CH₂CHClCOOethyl	
.42	CH ₂ CH=CH ₂	
.43	CH ₂ CH ₂ CH ₃	
.44	CH ₂ CH ₂ CF ₃	
.45	OCH(CH ₃)COOC ₂ H ₅ (R)	
.46	OCH(CH ₃)COOC ₂ H ₅ (S)	
47	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
48	CH₂CHCICOOH	
49	CH₂CHCICOOCH₃	

Comp.No.	R ₆	M.p.
6.50	CH₂CHCICOOC₃H₂(iso)	·
6.51	CH₂CHCICONHallyl	
6.52	CH₂C(CH₃)CICOOH	
6.53	CH₂C(CH₃)CICOOCH₃	
6.54	CH₂C(CH₃)CICOOEt	
6.55	CH₂C(CH₃)CICONHEt	
6.56	CH₂CH₂COOH	
6.57	CH ₂ CH ₂ COOCH ₃	
6.58	CH ₂ CH ₂ COOEt	
6.59	CHCICHCICOOH	
6.60	CHCICHCICOOCH ₃	
6.61	CHCICHCICOOEt	
6.62	CH₂CH(OCH₃)COOH	
6.63	CH₂CH(OCH₃)COOCH₃	
6.64	CH ₂ CH(OCH ₃)COOEt	
6.65	CH₂CH(SCH₃)COOH	
6.66	CH₂CH(SCH₃)COOCH₃	
6.67	CH ₂ CH(SCH ₃)COOEt	
6.68	СН=СНСООН	
6.69	CH=CHCOOCH ₃	· ·
6.70	CH=CHCOOEt	
6.71	CH=CCICOOH	
6.72	CH=CCICOOCH ₃	
6.73	COOEt	
6.74	CONH ₂	
5.75	-C(O)OCH ₂	
6.76	CONHSO₂CH ₃	





Comp.No.	R ₆	M.p.
6.77	COOCH₂COOH	<u> </u>
6.78	COOCH₂COOCH₃	
6.79	COOCH(CH ₃)COOH	
6.80	COOCH(CH ₃)COOCH ₃	
6.81	COOCH(CH₃)CH₂COOH	
6.82	COOCH(CH3)CH2COOCH3	
6.83	COOC(CH ₃) ₂ CN	
6.84	COOCH ₂ CH ₂ OCH ₃	
6.85	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
6.86	COOC(CH ₃) ₂ -C(O)O-CH ₂	
6.87	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
6.88	COOCH₂C≡CH	
5.89	COOC(CH ₃) ₂ COOCH ₂ C≡CH	
6.90	COOCH(CH₃)C≡CH	
5.91	COOC(CH ₃) ₂ COCH ₃	
5.92	NHallyl	
5.93	N(COCH₃)allyl	
5.94	N(Et)SO ₂ CH ₃	
5.95	N(allyl)SO₂CH₃	
5.96	N(allyl)SO₂Et	
.97	SO₂N(CH ₃)₂	
.98	SO ₂ NH ₂	
.99	SO₂NHCOCH ₃	
.100	ОН	
.101	OEt	
.102	Oallyl	
.103	OCH₂C≡CCH₃	

Comp.No.	R ₆	M.p.
6.104	OCH(CH ₃)CH=CH ₂	·
6.105	OCH₂CH₂OCH₂CH₃	
6.106	OCH₂CH₂OCH₂CH₂OCH₃	
6.107	OCH ₂	
6.108	OCH₂CH₂NHCH₃	
6.109	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
6.110	OCH ₂ CH ₂ COOH	
6.111	OC(CH ₃) ₂ COOH	
6.112	OC(CH ₃) ₂ COOCH ₃	
6.113	OC(CH ₃) ₂ COOEt	
6.114	OCH₂COOH	
6.115	OSO ₂ CH ₃	
6.116	OSO ₂ CF ₃	
6.117	CH₂CHCICOOC₂H₅	
6.118	CH ₂ CHCICON(C ₂ H ₅) ₂	
6.119	CH₂CHCICONHOH	
6.120	CH ₂ CHCICOOCH ₂ C ₆ H ₅	
6.121	CH₂CH(CH₃)COOH	
6.122	CH ₂ CH(CH ₃)COOC ₂ H ₅	·
6.123	-COOCH ₂	
5.124	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
5.125	-соос(сн ₃) ₂ соосн ₂ —	
6.126	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
5.127	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	





Comp.No.	R ₆	M.p.
6.128	OCH ₂	

Table 7: Compounds of formula Ig

$$CI \longrightarrow CH_3$$
 SCH_3 CH_3 CH_3

Comp.No.	R ₆	M.p.
7.1	Н	oil
7.2	NO ₂	
7.3	Br	
7.4		90.00
7.5	CN	88-90
7.6	OCH ₃	88-90
7.7	N(SO ₂ CH ₃) ₂	196-198
7.8	NHSO₂CH₃	120-122
7.9	OC ₃ H ₇ (iso)	
7.10	O-propargyl	88-89
7.11	OCH(CH₃)C≡CH	
7.12	O-phenyl	
7.13	O-2-pyridyl	
7.14	O-2-pyrimidinyl	
7.15	OCH ₂ COOC ₂ H ₅	115-116
7.16	OCH ₂ COOC ₅ H ₁₁ (n)	74-76
7.17	OCH₂COO-benzyl	
'.18	OCH(CH ₃)COObenzyl (S)	
'.19	OCH(CH ₃)COObenzyl (R)	
.20	OCH(CH ₃)COObenzyl (R,S)	
.21	SC ₃ H ₇ (iso)	oil





Comp.No.	R ₆	M.p.
7.22	SH	78-80
7.23	SCH₂COOCH₃	
7.24	SCH₂COOC₂H₅	106-108
7.25	SCH(CH₃)COObenzyl (S)	
7.26	SCH(CH₃)COObenzyl (R)	
7.27	SCH(CH₃)COObenzyl (R,S)	
7.28	SCH₂COObenzyl	
7.29	SO ₂ CI	121-123
7.30	SO ₂ CH ₃	
7.31	SO₂NHCH₃	152-153
7.32	СООН	210-217
7.33	COOCH ₃	92-93
7.34	COOC₃H ₇ (iso)	oil
7.35	COOC(CH ₃) ₂ COOH	157-162
7.36	COOC(CH ₃) ₂ COO-allyl	
2.37	COOC(CH ₃) ₂ COOCH ₃	
.38	COOC(CH ₃) ₂ COOethyl	79.5-81.5
.39	COOC(CH ₃) ₂ CONH-allyl	resin
.40	CH₂CHClCOOethyl	Oil
.41	CH ₂ CH=CH ₂	Oil
.42	CH ₂ CH ₂ CH ₃	oil
.43	CH ₂ CH ₂ CF ₃	74-76
.44	OCH(CH ₃)COOC ₂ H ₅ (R)	
.45	OCH(CH ₃)COOC ₂ H ₅ (S)	
46	OCH(CH ₃)COOC₂H ₅ (R,S)	· · · · · ·
47	CH₂CHCICOOH	98-99
48	CH₂CHCICOOCH₃	oil
49	CH ₂ CHClCOOC ₃ H ₇ (iso)	

Comp.No.	R ₆	M.p.
7.50	CH₂CHClCONHallyl	
7.51	CH₂C(CH₃)CICOOH	
7.52	CH₂C(CH₃)CICOOCH₃	oil
7.53	CH₂C(CH₃)CICOOEt	
7.54	CH₂C(CH₃)CICONHEt	
7.55	CH₂CH₂COOH	
7.56	CH₂CH₂COOCH₃	110-111
7.57	CH₂CH₂COOEt	
7.58	СНСІСНСІСООН	
7.59	CHCICHCICOOCH3	
7.60	CHCICHCICOOEt	
7.61	CH₂CH(OCH₃)COOH	
7.62	CH₂CH(OCH₃)COOCH₃	
7.63	CH₂CH(OCH₃)COOEt	
7.64	CH₂CH(SCH₃)COOH	
7.65	CH₂CH(SCH₃)COOCH₃	
7.66	CH₂CH(SCH₃)COOEt	
7.67	CH=CHCOOH	
7.68	CH=CHCOOCH ₃	148-149
7.69	CH=CHCOOEt	
7.70	CH=CCICOOH	
7.71	CH=CCICOOCH3	
7.72	COOEt	
7.73	CONH ₂	
7.74	-C(O)OCH ₂	
7.75	CONHSO₂CH₃	
7.76	COOCH ₂ COOH	





Comp.No.	R ₆	M.p.
7.77	COOCH₂COOCH₃	
7.78	COOCH(CH3)COOH	
7.79	COOCH(CH ₃)COOCH ₃	
7.80	COOCH(CH₃)CH₂COOH	
7.81	COOCH(CH₃)CH₂COOCH₃	
7.82	COOC(CH₃)₂CN	
7.83	COOCH₂CH₂OCH₃	
7.84	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
7.85	COOC(CH ₃) ₂ -C(O)O-CH ₂	
7.86	COOC(CH ₃) ₂ COOCH ₂ PHENYL	oil
7.87	COOCH₂C=CH	
7.88	COOC(CH ₃) ₂ COOCH ₂ C=CH	
7.89	COOCH(CH₃)C≡CH	
7.90	COOC(CH ₃) ₂ COCH ₃	
7.91	NHallyl	
7.92	N(COCH₃)allyl	
7.93	N(Et)SO₂CH₃	
7.94	N(allyl)SO₂CH₃	
7.95	N(allyl)SO₂Et	
7.96	SO ₂ N(CH ₃) ₂	oil
7.97	SO ₂ NH ₂	181-182
7.98	SO ₂ NHCOCH ₃	
7.99	ОН	164-166
7.100	OEt .	
7.101	Oallyl	
7.102	OCH ₂ C≡CCH ₃	
7.103	OCH(CH ₃)CH=CH ₂	

Comp.No.	R ₆	M.p.
7.104	OCH₂CH₂OCH₂CH₃	
7.105	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
7.106	OCH ₂	
7.107	OCH₂CH₂NHCH₃	
7.108	OCH₂CH₂N(CH₃)COCH₃	
7.109	OCH₂CH₂COOH	
7.110	OC(CH₃)₂COOH	
7.111	OC(CH ₃) ₂ COOCH ₃	
7.112	OC(CH ₃) ₂ COOEt	
7.113	ОСН₂СООН	
7.114	OSO ₂ CH ₃	
7.115	OSO ₂ CF ₃	
7.116	CI	58-60
7.117	CH₂CHCICOOC₂H₅	
7.118	CH₂CHCICON(C₂H₅)₂	
7.119	CH₂CHCICONHOH	
7.120	CH₂CHCICOOCH₂C ₆ H ₅	
7.121	CH₂CH(CH₃)COOH	
7.122	CH ₂ CH(CH ₃)COOC ₂ H ₅	
7.123	-COOCH ₂	
7.124	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
7.125	-соос(сн ₃) ₂ соосн ₂	
7.126	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
7.127	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	





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Comp.No.	R ₆	M.p.
7.128	OCH ₂ —	
7.129	CH=CH-CF ₃	102-103

Table 8: Compounds of formula Ih

$$CI \xrightarrow{CI} CH_3 SOCH_3$$

$$R_6 CH_3$$

$$CH_3$$

$$CH_3$$

Comp.No.	R ₆	M.p.
8.1	Н	98-100
8.2	NH ₂	164-165
8.3	NO ₂	164-165
8.4	Br	
8.5	1	
8.6	CN	· ·
8.7	OCH ₃	152-153
8.8	N(SO ₂ CH ₃) ₂	246-248
8.9	NHSO₂CH₃	153-154
8.10	OC ₃ H ₇ (iso)	
8.11	O-propargyl	151-152
8.12	OCH(CH₃)C≡CH	
3.13	O-phenyl	
3.14	O-2-pyridyl	
3.15	O-2-pyrimidinyl	
3.16	OCH ₂ COOC ₂ H ₅	165-166
3.17	OCH ₂ COOC ₅ H ₁₁ (n)	89-91
3.18	OCH₂COO-benzyl	-
3.19	OCH(CH ₃)COObenzyl (S)	
3.20	OCH(CH₃)COObenzyl (R)	
3.21	OCH(CH₃)COObenzyl (R,S)	

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8.22	SC ₃ H ₇ (iso)	
8.23	SH	
8.24	SCH₂COOCH₃	
8.25	SCH ₂ COOC ₂ H ₅	
8.26	SCH(CH ₃)COObenzyl (S)	
8.27	SCH(CH ₃)COObenzyl (R)	
8.28	SCH(CH₃)COObenzyl (R,S)	
8.29	SCH₂COObenzyl	
8.30	SO₂CI	160-163
8.31	SO₂CH₃	
8.32	SO₂NHCH₃	
8.33	СООН	78-86
8.34	COOCH₃	134-136
8.35	COOC ₃ H ₇ (iso)	116-119
8.36	COOC(CH ₃) ₂ COOH	78-86
8.37	COOC(CH ₃)₂COO-allyl	
8.38	COOC(CH ₃) ₂ COOCH ₃	
8.39	COOC(CH₃)₂COOethyl	oil
8.40	COOC(CH₃)₂CONH-allyl	168-172
8.41	CH₂CHClCOOethyl	87-89
8.42	CH₂CH=CH₂	118-120
8.43	CH₂CH₂CH₃	128-129
8.44	CH₂CH₂CF₃	
8.45	OCH(CH ₃)COOC ₂ H ₅ (R)	
8.46	OCH(CH ₃)COOC ₂ H ₅ (S)	
8.47	OCH(CH₃)COOC₂H₅(R,S)	
8.48	CH₂CHCICOOH	92-94
8.49	CH₂CHCICOOCH₃	95-96
8.50	CH₂CHClCOOC₃H₁(iso)	

		·
8.51	CH₂CHCICONHallyl	
8.52	CH₂C(CH₃)CICOOH	
8.53	CH₂C(CH₃)CICOOCH₃	oil
8.54	CH₂C(CH₃)CICOOEt	
8.55	CH₂C(CH₃)CICONHEt	
8.56	CH₂CH₂COOH	
8.57	CH₂CH₂COOCH₃	153-154
8.58	CH ₂ CH ₂ COOEt	
8.59	снсіснсісоон	
8.60	CHCICHCICOOCH₃	
8.61	CHCICHCICOOEt	·
8.62	CH₂CH(OCH₃)COOH	
8.63	CH₂CH(OCH₃)COOCH₃	
8.64	CH₂CH(OCH₃)COOEt	
8.65	CH₂CH(SCH₃)COOH	
8.66	CH₂CH(SCH₃)COOCH₃	
8.67	CH₂CH(SCH₃)COOEt	
8.68	CH=CHCOOH	·.
8.69	CH=CHCOOCH ₃	
8.70	CH=CHCOOEt	
8.71	CH=CCICOOH	
8.72	CH=CCICOOCH3	
8.73	COOEt	122-123
8.74	CONH ₂	
8.75	-C(0)OCH ₂	
8.76	CONHSO₂CH₃	
8.77	COOCH₂COOH	
8.78	COOCH ₂ COOCH ₃	





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8.79	COOCH(CH₃)COOH	
8.80	COOCH(CH ₃)COOCH ₃	
8.81	COOCH(CH₃)CH₂COOH	
8.82	COOCH(CH₃)CH₂COOCH₃	
8.83	COOC(CH ₃) ₂ CN	
8.84	COOCH₂CH₂OCH₃	
8.85	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
8.86	COOC(CH ₃) ₂ -C(O)O-CH ₂	
8.87	COOC(CH ₃) ₂ COOCH ₂ PHENYL	oil
8.88	COOCH₂C=CH	
8.89	COOC(CH ₃) ₂ COOCH ₂ C=CH	
8.90	COOCH(CH₃)C∈CH	
8.91	COOC(CH ₃) ₂ COCH ₃	
8.92	NHallyl	
8.93	N(COCH₃)allyl	
8.94	N(Et)SO ₂ CH ₃	
8.95	N(allyl)SO₂CH₃	
8.96	N(allyl)SO₂Et	
8.97	SO ₂ N(CH ₃) ₂	
8.98	SO ₂ NH ₂	
8.99	SO₂NHCOCH₃	
8.100	ОН	
8.101	OEt	
8.102	Oallyl	
8.103	OCH₂C≡CCH₃	
8.104	OCH(CH ₃)CH=CH ₂	
8.105	OCH ₂ CH ₂ OCH ₂ CH ₃	
8.106	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
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8.107	OCH ₂	
8.108	OCH₂CH₂NHCH₃	
8.109	OCH₂CH₂N(CH₃)COCH₃	
8.110	OCH ₂ CH ₂ COOH	
8.111	OC(CH₃)₂COOH	
8.112	OC(CH ₃)₂COOCH ₃	
8.113	OC(CH₃)₂COOEt	
8.114	OCH₂COOH	
8.115	OSO ₂ CH ₃	
8.116	OSO₂CF ₃	
8.117	CH₂CHCICOOC₂H₅	
8.118	CH₂CHCICON(C₂H₅)₂	
8.119	CH₂CHCICONHOH	
8.120	CH ₂ CHClCOOCH ₂ C ₆ H ₅	
8.121	CH₂CH(CH₃)COOH	
8.122	CH ₂ CH(CH ₃)COOC ₂ H ₅	
8.123	-COOCH ₂	
8.124	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
8.125	-COOC(CH ₃) ₂ COOCH ₂	
8.126	COOC(CH₃)₂CONHCH₂C≡CH	
8.127	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
3.128	OCH ₂ —	
3.129	CH=CH-CF ₃	





Table 9: Compounds of formula li

Comp.No.	R ₆	M.p.
9.1	Н	110-112
9.2	NH ₂	185-187
9.3	NO ₂	156-159
9.4	Br	
9.5	TI TI	167-170
9:6	CN	
9.7	OCH ₃	150-151
9.8	N(SO ₂ CH ₃) ₂	265-267
9.9	NHSO₂CH₃	185-187
9.10	OC ₃ H ₇ (iso)	
9.1.1	O-propargyl	126-127
9.12	OCH(CH₃)C≡CH	
9.13	O-phenyl	
9.14	O-2-pyridyl	
9.15	O-2-pyrimidinyl	
9.16	OCH ₂ COOC ₂ H ₅	140-141
9.17	OCH ₂ COOC ₅ H ₁₁ (n)	118-120
9.18	OCH ₂ COO-benzyl	
9.19	OCH(CH ₃)COObenzyl (S)	
9.20	OCH(CH₃)COObenzyl (R)	
9.21	OCH(CH ₃)COObenzyl (R,S)	

Comp.No.	R ₆	M.p.
9.22	SC ₃ H ₇ (iso)	93-95
9.23	SH	
9.24	SCH₂COOCH₃	
9.25	SCH₂COOC₂H₅	133-135
9.26	SCH(CH ₃)COObenzyl (S)	
9.27	SCH(CH₃)COObenzyl (R)	
9.28	SCH(CH ₃)COObenzyl (R,S)	
9.29	SCH₂COObenzyl	
9.30	SO₂CI	169-171
9.31	SO₂CH₃	100 177
9.32	SO₂NHCH₃	
9.33	СООН	201-208
.34	COOCH ₃	137-139
.35	COOC ₃ H ₇ (iso)	111-114
.36	COOC(CH ₃) ₂ COOH	
.37	COOC(CH ₃) ₂ COO-allyl	179-182
.38	COOC(CH ₃) ₂ COOCH ₃	
.39	COOC(CH ₃) ₂ COOethyl	a:l
40	COOC(CH ₃) ₂ CONH-allyl	oil
41	CH ₂ CHClCOOethyl	141-143
42	CH ₂ CH=CH ₂	oil
43	CH ₂ CH ₂ CH ₃	oil
44	CH ₂ CH ₂ CF ₃	107-109
45		
46	OCH(CH ₃)COOC ₂ H ₅ (R)	
47	OCH(CH ₃)COOC ₂ H ₅ (S)	
+ 7 	OCH(CH₃)COOC₂H₅(R,S)	
	CH₂CHCICOOH	185-187
19	CH₂CHCICOOCH₃	96-98





Comp.No.	R ₆	M.p.
9.50	CH₂CHCICOOC₃H₂(iso)	
9.51	CH₂CHCICONHallyI	
9.52	CH₂C(CH₃)CICOOH	
9.53	CH₂C(CH₃)CICOOCH₃	oil
9.54	CH₂C(CH₃)CICOOEt	
9.55	CH₂C(CH₃)CICONHEt	
9.56	CH ₂ CH ₂ COOH	
9.57	CH₂CH₂COOCH₃	152-153
9.58	CH₂CH₂COOEt	
9.59	CHCICHCICOOH .	
9.60	CHCICHCICOOCH3	
9.61	CHCICHCICOOEt	
9.62	CH₂CH(OCH₃)COOH	
9.63	CH₂CH(OCH₃)COOCH₃	
9.64	CH₂CH(OCH₃)COOEt	
9.65	CH₂CH(SCH₃)COOH	
9.66	CH ₂ CH(SCH ₃)COOCH ₃	
9.67	CH₂CH(SCH₃)COOEt	
9.68	CH=CHCOOH	
9.69	CH=CHCOOCH ₃	
9.70	CH=CHCOOEt	
9.71	CH=CCICOOH	
9.72	CH=CCICOOCH ₃	
9.73	COOEt	105-107
9.74	CONH ₂	
9.75	-C(O)OCH ₂	
9.76	CONHSO ₂ CH ₃	

Comp.No.	R ₆	M.p.
9.77	COOCH₂COOH	·
9.78	COOCH₂COOCH₃	
9.79	COOCH(CH3)COOH	
9.80	COOCH(CH ₃)COOCH ₃	
9.81	COOCH(CH ₃)CH ₂ COOH	
9.82	COOCH(CH ₃)CH ₂ COOCH ₃	
9.83	COOC(CH ₃) ₂ CN	
9.84	COOCH ₂ CH ₂ OCH ₃	
9.85	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
9.86	COOC(CH ₃) ₂ -C(O)O-CH ₂	
9.87	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
9.88	COOCH₂C≡CH	
9.89	COOC(CH ₃) ₂ COOCH ₂ C≡CH	
.90	COOCH(CH₃)C≡CH	
.91	COOC(CH ₃)₂COCH ₃	
.92	NHaliyi	
.93	N(COCH₃)allyl	
.94	N(Et)SO₂CH₃	
.95	N(allyI)SO ₂ CH ₃	
.96	N(allyl)SO ₂ Et	
.97	SO ₂ N(CH ₃) ₂	
.98	SO ₂ NH ₂	
.99	SO₂NHCOCH₃	
100	OH .	164-165
101	OEt	
102	Oallyl	
103	OCH₂C≡CCH₃	





Comp.No.	R ₆	M.p.
9.104	OCH(CH ₃)CH=CH ₂	<u> </u>
9.105	OCH ₂ CH ₂ OCH ₂ CH ₃	
9.106	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
9.107	OCH ₂	
9.108	OCH₂CH₂NHCH₃	
9.109	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
9.110	OCH ₂ CH ₂ COOH	
9.111	OC(CH ₃)₂COOH	
9.112	OC(CH ₃) ₂ COOCH ₃	
9.113	OC(CH₃)₂COOEt	
9.114	OCH₂COOH	
9.115	OSO ₂ CH ₃	
9.116	OSO ₂ CF ₃	
9.117	CH₂CHCICOOC₂H₅	
9.118	CH₂CHCICON(C₂H₅)₂	
9.119	CH₂CHCICONHOH	
9.120	CH₂CHCICOOCH₂C6H5	
9.121	CH₂CH(CH₃)COOH	
9.122	CH ₂ CH(CH ₃)COOC ₂ H ₅	
9.123	-COOCH ₂ —	
9.124	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
9.125	-cooc(ch ₃) ₂ соосн ₂ —	
0.126	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	·
0.127	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	

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Comp.No.	R ₆	M.p.
9.128	OCH ₂	





Table 10: Compounds of formula lj

$$CH_{3} \xrightarrow{CI} CH_{3} SCH_{3}$$

$$CH_{3} CH_{3}$$

$$CH_{3}$$

$$CH_{3}$$

$$CH_{3}$$

Comp.No.	R ₆	M.p.
10.1	Н	72-74
10.2	CN	
10.3	OCH₃	
10.4	NHSO₂CH₃	
10.5	OC ₃ H ₇ (iso)	
10.6	O-propargyl	
10.7	OCH(CH₃)C≡CH	
10.8	OCH ₂ COOCH ₂ CH ₃	
10.9	OCH ₂ CH ₂ OCH ₃	
10.10	OCH ₂ CH ₂ SCH ₂ CH ₃	
10.11	OCH₂COOCH₃	
10.12	OCH ₂ COOC ₅ H ₁₁ (n)	<u> </u>
10.13	OCH₂COO-benzyl	
10.14	OCH(CH ₃)COObenzyl	
10.15	SC ₃ H ₇ (iso)	
10.16	SCH₂COOCH₃	
10.17	SCH ₂ COOC ₂ H ₅	
10.18	SCH(CH ₃)COObenzyl	
10.19	SCH₂COObenzyl	
10.20	COOCH ₃	
10.21	COOC ₃ H ₇ (iso)	

Comp.No.	R ₆	M.p.
10.22	COOC(CH ₃) ₂ COOH	· ·
10.23	COOC(CH ₃) ₂ COO-allyl	
10.24	COOC(CH₃)₂COOCH₃	
10.25	COOC(CH ₃)₂COOethyl	
10.26	COOC(CH ₃)₂CONH-allyl	
10.27	CH₂CHClCOOethyl	
10.28	CH ₂ CH=CH ₂	
10.29	CH ₂ CH ₂ CH ₃	
10.30	CH ₂ CH ₂ CF ₃	-
10.31	OCH(CH ₃)COOC ₂ H ₅ (R)	
10.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
10.33	OCH(CH₃)COOC₂H₅(R,S)	
10.34	CH₂CHCICOOH	
10.35	CH₂CHCICOOCH₃	
10.36	CH₂CHCICOOC₃H₂(iso)	· ·
10.37	CH₂CHCICONHallyl	
10.38	CH₂C(CH₃)CICOOH	
10.39	CH₂C(CH₃)CICOOCH₃	
10.40	CH₂C(CH₃)CICOOEt	
10.41	CH₂C(CH₃)CICONHEt	
10.42	CH₂CH₂COOH	
10.43	CH ₂ CH ₂ COOCH ₃	
10.44	CH ₂ CH ₂ COOEt	
10.45	СНСІСНСІСООН	
10.46	CHCICHCICOOCH ₃	
10.47	CHCICHCICOOEt	
10.48	CH₂CH(OCH₃)COOH	
10.49	CH₂CH(OCH₃)COOCH₃	

Comp.No.	R ₆	M.p.
10.50	CH₂CH(OCH₃)COOEt	
10.51	CH₂CH(SCH₃)COOH	
10.52	CH₂CH(SCH₃)COOCH₃	
10.53	CH₂CH(SCH₃)COOEt	
10.54	CH=CHCOOH	
10.55	CH=CHCOOCH ₃	
10.56	CH=CHCOOEt	
10.57	CH=CCICOOH	
10.58	CH=CCICOOCH ₃	
10.59	COOEt	
10.60	CONH ₂	
10.61	-C(O)OCH ₂	
10.62	CONHSO₂CH₃	
10.63	СООСН₂СООН	
10.64	COOCH₂COOCH₃	
10.65	соосн(сн₃)соон	
10.66	COOCH(CH ₃)COOCH ₃	
10.67	COOCH(CH ₃)CH ₂ COOH	
10.68	COOCH(CH ₃)CH ₂ COOCH ₃	
10.69	COOC(CH₃)₂CN	
10.70	COOCH₂CH₂OCH₃	
10.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
10.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
10.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
10.74	COOCH₂C≡CH	
10.75	COOC(CH ₃) ₂ COOCH ₂ C=CH	

Comp.No.	R ₆	M.p.
10.76	COOCH(CH₃)C≡CH	
10.77	COOC(CH ₃)₂COCH ₃	
10.78	NHallyl	
10.79	N(COCH ₃)allyl	
10.80	N(Et)SO ₂ CH ₃	
10.81	N(allyl)SO ₂ CH ₃	
10.82	N(allyl)SO ₂ Et	
10.83	SO ₂ N(CH ₃) ₂	
10.84	SO ₂ NH ₂	
10.85	SO ₂ NHCOCH ₃	
10.86	ОН	
10.87	OEt	
10.88	Oallyl	
10.89	OCH ₂ C≡CCH ₃	
10.90	OCH(CH ₃)CH=CH ₂	
10.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
10.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
10.93	OCH ₂	
10.94	OCH₂CH₂NHCH₃	
10.95	OCH₂CH₂N(CH₃)COCH₃	
10.96	OCH ₂ CH ₂ COOH	
10.97	OC(CH₃)₂COOH	
10.98	OC(CH ₃) ₂ COOCH ₃	
10.99	OC(CH ₃)₂COOEt	
0.100	OCH₂COOH .	
0.101	OSO ₂ CH ₃	
0.102	OSO ₂ CF ₃	





Comp.No.	R ₆	M.p.
10.103	CH₂CHCICOOC₂H₅	w.p.
10.104	CH ₂ CHCICON(C ₂ H ₅) ₂	
10.105	CH₂CHCICONHOH	
10.106	CH2CHCICOOCH2C6H5	
10.107	CH₂CH(CH₃)COOH	
10.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
10.109	-соосн ₂ —	
10.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
10.111	-COOC(CH ₃) ₂ COOCH ₂ —	
10.112	COOC(CH ₃) ₂ CONHCH ₂ C=CH	
10.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
10.114	OCH ₂	

Table 11: Compounds of formula lk

$$\begin{array}{c|c} & CI & CH_3 \\ \hline & & \\$$

Comp.No.	R ₆	M.p.
11.1	Н	88-91
11.2	CN	
11.3	OCH ₃	
11.4	NHSO₂CH₃	
11.5	OC ₃ H ₇ (iso)	
11.6	O-propargyl	
11.7	OCH(CH₃)C≡CH	
11.8	OCH ₂ COOCH ₂ CH ₃	
11.9	OCH₂CH₂OCH₃	
11.10	OCH ₂ CH ₂ SCH ₂ CH ₃	· · · · · · · · · · · · · · · · · · ·
11.11	OCH ₂ COOCH ₃	
11.12	OCH ₂ COOC ₅ H ₁₁ (n)	
11.13	OCH₂COO-benzyl	
11.14	OCH(CH ₃)COObenzyl	
11.15	SC ₃ H ₇ (iso)	
11.16	SCH₂COOCH₃	
11.17	SCH ₂ COOC ₂ H ₅	
11.18	SCH(CH₃)COObenzyl	
11.19	SCH₂COObenzyl	
11.20	COOCH ₃	
11.21	COOC ₃ H ₇ (iso)	





Comp.No.	R ₆	M.p.
11.22	COOC(CH ₃) ₂ COOH	
11.23	COOC(CH ₃) ₂ COO-allyl	
11.24	COOC(CH₃)₂COOCH₃	
11.25	COOC(CH₃)₂COOethyl	
11.26	COOC(CH₃)₂CONH-allyl	
11.27	CH₂CHClCOOethyl	
11.28	CH ₂ CH=CH ₂	
11.29	CH₂CH₂CH₃	
11.30	CH ₂ CH ₂ CF ₃	
11.31	OCH(CH ₃)COOC₂H ₅ (R)	
11.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
11.33	OCH(CH₃)COOC₂H₅(R,S)	
11.34	CH₂CHCICOOH	
11.35	CH₂CHCICOOCH₃	
11.36	CH₂CHCICOOC₃H₂(iso)	
11.37	CH₂CHCICONHallyl	
11.38	CH₂C(CH₃)CICOOH	
11.39	CH₂C(CH₃)CICOOCH₃	
11.40	CH₂C(CH₃)CICOOEt	
11.41	CH₂C(CH₃)CICONHEt	
11.42	CH ₂ CH ₂ COOH	
11.43	CH₂CH₂COOCH₃	
11.44	CH₂CH₂COOEt	
11.45	СНСІСНСІСООН	
11.46	CHCICHCICOOCH ₃	
11.47	CHCICHCICOOEt	
11.48	CH₂CH(OCH₃)COOH	
11.49	CH₂CH(OCH₃)COOCH₃	

Comp.No.	R ₆	M.p.
11.50	CH₂CH(OCH₃)COOEt	М.р.
11.51	CH₂CH(SCH₃)COOH	
11.52	CH₂CH(SCH₃)COOCH₃	
11.53	CH₂CH(SCH₃)COOEt	
11.54	CH=CHCOOH	
11.55	CH=CHCOOCH ₃	
11.56	CH=CHCOOEt	
11.57	CH=CCICOOH	· ·
11.58	CH=CCICOOCH ₃	
11.59	COOEt	
11.60	CONH ₂	
11.61	-C(O)OCH ₂ O	
11.62	CONHSO₂CH₃	
11.63	COOCH₂COOH	
11.64	COOCH₂COOCH₃	
11.65	COOCH(CH ₃)COOH	
11.66	COOCH(CH ₃)COOCH ₃	
11.67	COOCH(CH ₃)CH ₂ COOH	
11.68	COOCH(CH₃)CH₂COOCH₃	
11.69	COOC(CH ₃)₂CN	
11.70	COOCH ₂ CH ₂ OCH ₃	
11.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
1.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
1.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
1.74	COOCH₂C≡CH	
1.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	





Comp.No.	R ₆	M.p.
11.76	COOCH(CH₃)C≡CH	
11.77	COOC(CH ₃) ₂ COCH ₃	
11.78	NHallyl	
11.79	N(COCH₃)allyl	
11.80	N(Et)SO ₂ CH ₃	
11.81	N(allyI)SO₂CH ₃	
11.82	N(allyl)SO₂Et	
11.83	SO ₂ N(CH ₃) ₂	
11.84	SO₂NH₂	
11.85	SO₂NHCOCH ₃ .	
11.86	ОН	
11.87	OEt	
11.88	Oallyl	
11.89	OCH₂C≡CCH₃	
11.90	OCH(CH ₃)CH=CH ₂	
11.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
11.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
11.93	OCH2 O	
11.94	OCH ₂ CH ₂ NHCH ₃	· · · · · · · · · · · · · · · · · · ·
11.95	OCH₂CH₂N(CH₃)COCH₃	
11.96	OCH ₂ CH ₂ COOH	
11.97	OC(CH ₃)₂COOH	
11.98	OC(CH ₃) ₂ COOCH ₃	
11.99	OC(CH ₃)₂COOEt	
11.100	OCH₂COOH	
11.101	OSO₂CH₃	
11.102	OSO₂CF₃	

Comp.No.	R ₆	M.p.
11.103	CH₂CHCICOOC₂H₅	
11.104	CH₂CHCICON(C₂H₅)₂	
11.105	CH₂CHCICONHOH	
11.106	CH₂CHCICOOCH₂C6H5	
11.107	CH₂CH(CH₃)COOH	
11.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
11.109	-COOCH ₂	
11.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
11.111	-cooc(ch ₃) ₂ соосн ₂ —	
11.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
11.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	· ·
11.114	OCH ₂ —	





Table 12: Compounds of formula Im

$$\begin{array}{c|c} CI & CH_3 \\ \hline \\ CH_3 & SO_2CH_3 \\ \hline \\ R_6 & CH_3 \end{array} \tag{Im}$$

Comp.No.	R ₆	M.p.
12.1	Н	91-92
12.2	CN	
12.3	OCH ₃	
12.4	NHSO ₂ CH ₃	
12.5	OC ₃ H ₇ (iso)	
12.6	O-propargyl	
12.7	OCH(CH₃)C≡CH	
12.8	OCH ₂ COOCH ₂ CH ₃	
12.9	OCH₂CH₂OCH₃	
12.10	OCH₂CH₂SCH₂CH₃	
12.11	OCH₂COOCH₃	
12.12	OCH₂COOC₅H₁₁ (n)	
12.13	OCH ₂ COO-benzyl	
12.14	OCH(CH ₃)COObenzyl	
12.15	SC ₃ H ₇ (iso)	
12.16	SCH₂COOCH ₃	
12.17	SCH₂COOC₂H₅	
12.18	SCH(CH₃)COObenzyl	
12.19	SCH₂COObenzyl	
12.20	COOCH ₃	
12.21	COOC ₃ H ₇ (iso)	

Comp.No.	R ₆	M.p.
12.22	COOC(CH ₃) ₂ COOH	
12.23	COOC(CH ₃) ₂ COO-allyl	
12.24	COOC(CH ₃) ₂ COOCH ₃	
12.25	COOC(CH ₃) ₂ COOethyl	
12.26	COOC(CH ₃) ₂ CONH-allyl	
12.27	CH₂CHClCOOethyl	
12.28	CH ₂ CH=CH ₂	
12.29	CH ₂ CH ₂ CH ₃	
12.30	CH ₂ CH ₂ CF ₃	
12.31	OCH(CH ₃)COOC ₂ H ₅ (R)	
12.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
12.33	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
12.34	CH₂CHCICOOH	
12.35	CH₂CHCICOOCH₃	
12.36	CH ₂ CHClCOOC ₃ H ₇ (iso)	
12.37	CH₂CHClCONHallyl	
12.38	CH₂C(CH₃)CICOOH	
12.39	CH₂C(CH₃)CICOOCH₃	
12.40	CH₂C(CH₃)CICOOEt	
12.41	CH₂C(CH₃)CICONHEt	
12.42	CH₂CH₂COOH	
12.43	CH₂CH₂COOCH₃	
12.44	CH ₂ CH ₂ COOEt	
12.45	СНСІСНСІСООН	
12.46	CHCICHCICOOCH3	
12.47	CHCICHCICOOEt	
12.48	CH₂CH(OCH₃)COOH	
12.49	CH₂CH(OCH₃)COOCH₃	





Comp.No.	R ₆	M.p.
12.50	CH₂CH(OCH₃)COOEt	
12.51	CH₂CH(SCH₃)COOH	
12.52	CH₂CH(SCH₃)COOCH₃	
12.53	CH₂CH(SCH₃)COOEt	
12.54	CH=CHCOOH	
12.55	CH=CHCOOCH ₃	
12.56	CH=CHCOOEt	
12.57	CH=CCICOOH	
12.58	CH=CCICOOCH3	
12.59	COOEt	
12.60	CONH ₂	
12.61	-C(O)OCH ₂	
12.62	CONHSO₂CH₃	
12.63	COOCH₂COOH	
12.64	COOCH₂COOCH₃	
12.65	COOCH(CH3)COOH	
12.66	COOCH(CH ₃)COOCH ₃	
12.67	COOCH(CH₃)CH₂COOH	
12.68	COOCH(CH ₃)CH ₂ COOCH ₃	
12.69	COOC(CH ₃) ₂ CN	
12.70	COOCH ₂ CH ₂ OCH ₃	
2.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
2.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
2.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
2.74	COOCH₂C≡CH	
2.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	

Comp.No.	R ₆	M.p.
12.76	COOCH(CH₃)C≡CH	
12.77	COOC(CH ₃) ₂ COCH ₃	
12.78	NHallyl	
12.79	N(COCH₃)allyl	
12.80	N(Et)SO₂CH ₃	
12.81	N(aliyI)SO ₂ CH ₃	
12.82	N(allyl)SO ₂ Et	
12.83	SO₂N(CH₃)₂	
12.84	SO ₂ NH ₂	
12.85	SO₂NHCOCH ₃	
12.86	ОН	
12.87	OEt	
12.88	Oallyl	
12.89	OCH₂C≡CCH₃	
12.90	OCH(CH ₃)CH=CH ₂	
12.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
12.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
12.93	OCH ₂	
12.94	OCH₂CH₂NHCH₃	
12.95	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
12.96	OCH ₂ CH ₂ COOH	
2.97	OC(CH₃)₂COOH	
12.98	OC(CH₃)₂COOCH₃	
2.99	OC(CH₃)₂COOEt	
2.100	OCH₂COOH	
2.101	OSO ₂ CH ₃	
2.102	OSO ₂ CF ₃	





Comp.No.	R ₆	M.p.
12.103	CH₂CHCICOOC₂H₅	
12.104	CH₂CHCICON(C₂H₅)₂	
12.105	CH₂CHCICONHOH	
12.106	CH₂CHCICOOCH₂C6H5	
12.107	CH₂CH(CH₃)COOH	
12.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
12.109	-COOCH ₂	
12.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
12.111	-соос(сн ₃) ₂ соосн ₂ —	
12.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
12.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
12.114	OCH ₂ —	

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Table 13: Compounds of formula In

$$Br \xrightarrow{CI} CH_3 SCH_3$$

$$N \xrightarrow{N} CH_3$$
(In)

Comp.No.	R ₆	M.p.
13.1	Н	65-70
13.2	CN	· · · · · · · · · · · · · · · · · · ·
13.3	OCH₃	
13.4	NHSO ₂ CH ₃	
13.5	OC ₃ H ₇ (iso)	
13.6	O-propargyl	
13.7	OCH(CH₃)C≡CH	
13.8	OCH ₂ COOCH ₂ CH ₃	
13.9	OCH₂CH₂OCH₃	
13.10	OCH ₂ CH ₂ SCH ₂ CH ₃	
13.11	OCH₂COOCH₃	
13.12	OCH ₂ COOC ₅ H ₁₁ (n)	
13.13	OCH₂COO-benzyl	
13.14	OCH(CH ₃)COObenzyl	
13.15	SC ₃ H ₇ (iso)	
13.16	SCH₂COOCH₃	
13.17	SCH ₂ COOC ₂ H ₅	
13.18	SCH(CH₃)COObenzyl	
13.19	SCH₂COObenzyl	
13.20	COOCH ₃	
3.21	COOC ₃ H ₇ (iso)	

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Comp.No.	R ₆	M.p.
13.22	COOC(CH ₃) ₂ COOH	
13.23	COOC(CH ₃) ₂ COO-allyl	
13.24	COOC(CH ₃) ₂ COOCH ₃	
13.25	COOC(CH ₃) ₂ COOethyl	
13.26	COOC(CH ₃)₂CONH-allyl	
13.27	CH₂CHClCOOethyl	
13.28	CH ₂ CH=CH ₂	
13.29	CH₂CH₂CH₃	
13.30	CH₂CH₂CF₃	
13.31	OCH(CH₃)COOC₂H₅(R)	
13.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
13.33	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
13.34	CH₂CHCICOOH	
13.35	CH₂CHCICOOCH₃	
13.36	CH ₂ CHClCOOC ₃ H ₇ (iso)	
13.37	CH₂CHCICONHallyI	· ·
13.38	CH₂C(CH₃)CICOOH	
13.39	CH₂C(CH₃)CICOOCH₃	
13.40	CH₂C(CH₃)CICOOEt	
13.41	CH₂C(CH₃)CICONHEt	
13.42	CH₂CH₂COOH	
13.43	CH₂CH₂COOCH₃	· ·
13.44	CH₂CH₂COOEt	
13.45	СНСІСНСІСООН	
13.46	CHCICHCICOOCH3	
13.47	CHCICHCICOOEt	
13.48	CH₂CH(OCH₃)COOH	
13.49	CH₂CH(OCH₃)COOCH₃	· .

Comp.No.	R ₆	M.p.
13.50	CH₂CH(OCH₃)COOEt	
13.51	CH₂CH(SCH₃)COOH	
13.52	CH₂CH(SCH₃)COOCH₃	
13.53	CH₂CH(SCH₃)COOEt	
13.54	СН=СНСООН	
13.55	CH=CHCOOCH ₃	
13.56	CH=CHCOOEt	
13.57	CH=CCICOOH	
13.58	CH=CCICOOCH ₃	-
13.59	COOEt	
13.60	CONH ₂	
13.61	-C(O)OCH ₂	
13.62	CONHSO₂CH₃	
13.63	COOCH₂COOH	
13.64	COOCH₂COOCH₃	
13.65	COOCH(CH3)COOH	
13.66	COOCH(CH ₃)COOCH ₃	
13.67	COOCH(CH₃)CH₂COOH	
13.68	COOCH(CH ₃)CH ₂ COOCH ₃	· ·
13.69	COOC(CH ₃)₂CN	
13.70	COOCH ₂ CH ₂ OCH ₃	
13.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
13.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
13.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
13.74	COOCH₂C≡CH	
13.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	





Comp.No.	R ₆	M.p.
13.76	COOCH(CH₃)C≡CH	
13.77	COOC(CH ₃) ₂ COCH ₃	
13.78	NHallyl	
13.79	N(COCH₃)allyl	
13.80	N(Et)SO₂CH₃	
13.81	N(allyl)SO ₂ CH ₃	
13.82	N(allyl)SO ₂ Et	
13.83	SO ₂ N(CH ₃) ₂	
13.84	SO ₂ NH ₂	
13.85	SO₂NHCOCH₃	
13.86	ОН	
13.87	OEt	
13.88	Oallyl	
13.89	OCH ₂ C≡CCH ₃	
13.90	OCH(CH ₃)CH=CH ₂	
13.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
13.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
13.93	OCH ₂ —	
13.94	OCH₂CH₂NHCH₃	
13.95	OCH₂CH₂N(CH₃)COCH₃	
13.96	OCH₂CH₂COOH	
13.97	OC(CH ₃) ₂ COOH	
13.98	OC(CH ₃) ₂ COOCH ₃	
13.99	OC(CH₃)₂COOEt	
13.100	OCH₂COOH	
3.101	OSO₂CH ₃	
3.102	OSO ₂ CF ₃	

Comp.No.	R ₆	M.p.
13.103	CH₂CHCICOOC₂H₅	ivi.p.
13.104	CH₂CHCICON(C₂H₅)₂	
13.105	CH ₂ CHCICONHOH	
13.106	CH ₂ CHCICOOCH ₂ C ₆ H ₅	
13.107	CH₂CH(CH₃)COOH	
13.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
13.109	-COOCH ₂	
13.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
13.111	-соос(сн ₃) ₂ соосн ₂	
13.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
13.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
3.114	OCH ₂	

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Table 14: Compounds of formula lo

$$Br \xrightarrow{CI} CH_3 SOCH_3$$

$$N \xrightarrow{N} CH_3$$
(Io)

Comp.No.	R_6	M.p.
14.1	Н	115-117
14.2	CN	
14.3	OCH ₃	
14.4	NHSO₂CH ₃	
14.5	OC ₃ H ₇ (iso)	
14.6	O-propargyl	
14.7	OCH(CH₃)C≡CH	
14.8	OCH ₂ COOCH ₂ CH ₃	
4.9	OCH ₂ CH ₂ OCH ₃	
4.10	OCH ₂ CH ₂ SCH ₂ CH ₃	
4.11	OCH ₂ COOCH ₃	
4.12	OCH ₂ COOC ₅ H ₁₁ (n)	
4.13	OCH ₂ COO-benzyl	
4.14	OCH(CH ₃)COObenzyl	
4.15	SC ₃ H ₇ (iso)	
4.16	SCH₂COOCH₃	
4.17	SCH₂COOC₂H₅	
4.18	SCH(CH₃)COObenzyl	
4.19	SCH₂COObenzyl	
4.20	COOCH ₃	
4.21	COOC ₃ H ₇ (iso)	

Comp.No.	R ₆	M.p.
14.22	COOC(CH ₃)₂COOH	-
14.23	COOC(CH ₃) ₂ COO-allyl	
14.24	COOC(CH ₃) ₂ COOCH ₃	
14.25	COOC(CH ₃) ₂ COOethyl	
14.26	COOC(CH ₃) ₂ CONH-allyl	
14.27	CH₂CHCICOOethyl	
14.28	CH ₂ CH=CH ₂	
14.29	CH₂CH₂CH₃	
14.30	CH₂CH₂CF₃	
14.31	OCH(CH ₃)COOC ₂ H ₅ (R)	
14.32	OCH(CH₃)COOC₂H₅(S)	·
14.33	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
14.34	CH₂CHCICOOH	
14.35	CH₂CHClCOOCH₃	
14.36	CH ₂ CHClCOOC ₃ H ₇ (iso)	
14.37	CH₂CHCICONHallyl	
14.38	CH₂C(CH₃)CICOOH	
14.39	CH₂C(CH₃)CICOOCH₃	
14.40	CH₂C(CH₃)CICOOEt	
14.41	CH₂C(CH₃)CICONHEt	
14.42	CH₂CH₂COOH	
14.43	CH₂CH₂COOCH₃	
4.44	CH ₂ CH ₂ COOEt	
4.45	CHCICHCICOOH	
4.46	CHCICHCICOOCH ₃	
4.47	CHCICHCICOOEt	· ·
4.48	CH₂CH(OCH₃)COOH	
4.49	CH ₂ CH(OCH ₃)COOCH ₃	





Comp.No.	R ₆	M.p.
14.50	CH₂CH(OCH₃)COOEt	
14.51	CH₂CH(SCH₃)COOH	
14.52	CH₂CH(SCH₃)COOCH₃	
14.53	CH₂CH(SCH₃)COOEt	
14.54	CH=CHCOOH	
14.55	CH=CHCOOCH₃	
14.56	CH=CHCOOEt	
14.57	CH=CCICOOH	
14.58	CH=CCICOOCH ₃	
14.59	COOEt	
14.60	CONH ₂	
14.61	-C(O)OCH ₂	
14.62	CONHSO₂CH ₃	
14.63	COOCH₂COOH	
14.64	COOCH₂COOCH₃	
14.65	COOCH(CH ₃)COOH	
14.66	COOCH(CH ₃)COOCH ₃	
14.67	COOCH(CH₃)CH₂COOH	
14.68	COOCH(CH₃)CH₂COOCH₃	
14.69	COOC(CH ₃)₂CN	
14.70	COOCH ₂ CH ₂ OCH ₃	
14.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
14.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
14.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
14.74	COOCH₂C≡CH	
14.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	

Comp.No.	R ₆	M.p.
14.76	COOCH(CH₃)C≡CH	·
14.77	COOC(CH ₃)₂COCH ₃	
14.78	NHallyl	
14.79	N(COCH₃)allyl	
14.80	N(Et)SO₂CH₃	
14.81	N(allyl)SO ₂ CH ₃	
14.82	N(allyl)SO₂Et	
14.83	SO ₂ N(CH ₃) ₂	
14.84	SO ₂ NH ₂	-
14.85	SO₂NHCOCH₃	
14.86	ОН	
14.87	OEt	
14.88	Oallyl	
14.89	OCH₂C≡C(CH₃)	
14.90	OCH(CH ₃)CH=CH ₂	
14.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
14.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
14.93	OCH ₂ —O	
4.94	OCH₂CH₂NHCH₃	
4.95	OCH₂CH₂N(CH₃)COCH₃	
4.96	OCH₂CH₂COOH	
4.97	OC(CH ₃) ₂ COOH	
4.98	OC(CH ₃) ₂ COOCH ₃	
4.99	OC(CH ₃) ₂ COOEt	
4.100	OCH₂COOH	
4.101	OSO ₂ CH ₃	
4.102	OSO ₂ CF ₃	





Comp.No.	R ₆	M.p.
14.103	CH₂CHCICOOC₂H₅	·
14.104	CH ₂ CHCICON(C ₂ H ₅) ₂	
14.105	CH₂CHCICONHOH	
14.106	CH₂CHCICOOCH₂C₅H₅	
14.107	CH₂CH(CH₃)COOH	
14.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	<u>-</u>
14.109	-COOCH ₂ —	
14.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
14.111	-соос(сн ₃) ₂ соосн ₂ —Д	
14.112	COOC(CH ₃) ₂ CONHCH ₂ C=CH	
14.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
14.114	OCH ₂	

Table 15: Compounds of formula Ip

$$\mathsf{Br} \xrightarrow{\mathsf{CI}} \overset{\mathsf{CH}_3}{\mathsf{N}-\mathsf{N}} \overset{\mathsf{SO}_2\mathsf{CH}_3}{\mathsf{CH}_3} \tag{Ip}$$

Comp.No.	R ₆	M.p.
15.1	Н	110-112
15.2	CN	
15.3	OCH₃	
15.4	NHSO₂CH₃	
15.5	OC ₃ H ₇ (iso)	
15.6	O-propargyl	
15.7	OCH(CH₃)C≡CH	
15.8	OCH ₂ COOCH ₂ CH ₃	
15.9	OCH₂CH₂OCH₃	
15.10	OCH ₂ CH ₂ SCH ₂ CH ₃	:
15.11	OCH₂COOCH₃	
15.12	OCH ₂ COOC ₅ H ₁₁ (n)	
15.13	OCH ₂ COO-benzyl	· · · · · · · · · · · · · · · · · · ·
15.14	OCH(CH ₃)COObenzyl	
15.15	SC ₃ H ₇ (iso)	
15.16	SCH₂COOCH₃	
15.17	SCH₂COOC₂H₅	
15.18	SCH(CH₃)COObenzyl	
15.19	SCH₂COObenzyl	
15.20	COOCH ₃	
15.21	COOC ₃ H ₇ (iso)	





Comp.No.	R ₆	M.p.
15.22	COOC(CH ₃)₂COOH	
15.23	COOC(CH ₃) ₂ COO-allyl	
15.24	COOC(CH ₃) ₂ COOCH ₃	
15.25	COOC(CH ₃) ₂ COOethyl	
15.26	COOC(CH ₃) ₂ CONH-allyl	
15.27	CH₂CHCICOOethyl	
15.28	CH ₂ CH=CH ₂	
15.29	CH₂CH₂CH₃	
15.30	CH₂CH₂CF₃	
15.31	OCH(CH₃)COOC₂H₅(R)	
15.32	OCH(CH ₃)COOC₂H ₅ (S)	
15.33	OCH(CH₃)COOC₂H₅(R,S)	
15.34	CH₂CHCICOOH	
15.35	CH₂CHCICOOCH₃	
15.36	CH ₂ CHClCOOC ₃ H ₇ (iso)	
15.37	CH ₂ CHClCONHallyl	
15.38	CH₂C(CH₃)CICOOH	
15.39	CH₂C(CH₃)CICOOCH₃	
15.40	CH₂C(CH₃)CICOOEt	
15.41	CH₂C(CH₃)CICONHEt	
15.42	CH₂CH₂COOH	
15.43	CH ₂ CH ₂ COOCH ₃	
15.44	CH ₂ CH ₂ COOEt	
15.45	СНСІСНСІСООН	
15.46	CHCICHCICOOCH3	
15.47	CHCICHCICOOEt	
15.48	CH₂CH(OCH₃)COOH	
15.49	CH ₂ CH(OCH ₃)COOCH ₃	

Comp.No.	R ₆	M.p.
15.50	CH₂CH(OCH₃)COOEt	- '
15.51	CH₂CH(SCH₃)COOH	
15.52	CH₂CH(SCH₃)COOCH₃	
15.53	CH₂CH(SCH₃)COOEt	
15.54	CH=CHCOOH	
15.55	CH=CHCOOCH ₃	
15.56	CH=CHCOOEt	
15.57	CH=CCICOOH	
15.58	CH=CCICOOCH ₃	
15.59	COOEt	
15.60	CONH ₂	
15.61		
	-C(O)OCH ₂	
15.62	CONHSO₂CH₃	
15.63	СООСН₂СООН	
15.64	COOCH₂COOCH₃	
15.65	COOCH(CH3)COOH	
15.66	COOCH(CH ₃)COOCH ₃	
15.67	COOCH(CH₃)CH₂COOH	
15.68	COOCH(CH₃)CH₂COOCH₃	
15.69	COOC(CH ₃) ₂ CN	
15.70	COOCH₂CH₂OCH₃	
5.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
5.72	COOC(CH ₃) ₂ -C(O)O-CH ₂ -	
F 70		
5.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
5.74	COOCH ₂ C≡CH	
5.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	

Comp.No.	R ₆	M.p.
15.76	COOCH(CH₃)C≡CH	
15.77	COOC(CH ₃) ₂ COCH ₃	
15.78	NHallyl	
15.79	N(COCH₃)allyl	
15.80	N(Et)SO ₂ CH ₃	
15.81	N(allyl)SO ₂ CH ₃	
15.82	N(allyl)SO ₂ Et	
15.83	SO ₂ N(CH ₃) ₂	
15.84	SO₂NH₂	
15.85	SO₂NHCOCH₃	
15.86	ОН	
15.87	OEt	
15.88	Oallyl	
15.89	OCH₂C≡CCH ₃	
15.90	OCH(CH ₃)CH=CH ₂	
15.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
15.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
15.93	OCH ₂	
15.94	OCH₂CH₂NHCH₃	
15.95	OCH₂CH₂N(CH₃)COCH₃	
15.96	OCH₂CH₂COOH	
15.97	OC(CH ₃)₂COOH	
15.98	OC(CH ₃)₂COOCH ₃	
15.99	OC(CH ₃) ₂ COOEt	
15.100	OCH₂COOH	
15.101	OSO₂CH₃	
15.102	OSO ₂ CF ₃	

Comp.No.	R ₆	M.p.
15.103	CH₂CHCICOOC₂H₅	•
15.104	CH ₂ CHCICON(C ₂ H ₅) ₂	
15.105	CH₂CHCICONHOH	
15.106	CH₂CHCICOOCH₂C ₆ H ₅	
15.107	CH₂CH(CH₃)COOH	
15.108	CH₂CH(CH₃)COOC₂H₅	
15.109	-COOCH ₂ —	
15.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
15.111	-COOC(CH ₃) ₂ COOCH ₂ —	
15.112	COOC(CH ₃) ₂ CONHCH ₂ C=CH	
15.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
15.114	OCH ₂	



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Table 16: Compounds of formula Iq

Comp.	R ₆	R ₄	R ₁	R ₂	R ₃	n	M.p.
No.							ivi.p.
16.1	OCH ₃	CI	Et	Me	Me	0	
16.2	OCH ₃	CI	Et	Me	Me	1	
16.3	OCH ₃	CI	Et	Me	Me	2	
16.4	OCH ₃	F	Et	Me	Me	0	
16.5	OCH₃	F	Et	Me	Me	1	
16.6	OCH ₃	F	Et	Me	Me	2	
16.7	Н	CI	Et	Me	Me	0	oil
16.8	Н	CI	Et	Me	Me	1	oil
16.9	Н	CI	Et	Me	Me	2	oil
16.10	Н	F	Et	Ме	Ме	0	
16.11	Н	F	Et	Ме	Ме	1	
16.12	Н	F	Et	Me	Ме	2	
16.13	O-propargyl	CI	Et	Me	Me	O	
16.14	O-propargyl	CI	Et	Me	Me	1	
16.15	O-propargyl	CI	Et	Me	Ме	2	
16.16	O-propargyl	F	Et	Me	Ме	0	
16.17	O-propargyl	F	Et	Me	Me	1	
16.18	O-propargyl	F	Et	Ме	Ме	2	
16.19	COOC ₃ H ₇ (iso)	CI	Et	Ме	Ме	0	
16.20	COOC ₃ H ₇ (iso)	CI	Et	Me	Ме	1	
16.21	COOC ₃ H ₇ (iso)	CI	Et	Me	Ме	. 2	

						÷.*	
Comp.	R ₆	R ₄	R ₁	R ₂	R ₃	n	M.p.
No.							
16.22	COOC ₃ H ₇ (iso)	F	Et	Ме	Me	0	
16.23	COOC ₃ H ₇ (iso)	F	Et	Ме	Ме	1	
16.24	COOC ₃ H ₇ (iso)	F	Et	Ме	Ме	2	
16.25	COOC₃H ₇ (iso)	Н	Et	Ме	Me	0	49-51
16.26	COOC ₃ H ₇ (iso)	Н	Et	Ме	Ме	1	110-112
16.27	COOC ₃ H ₇ (iso)	Н	Et	Ме	Ме	2	102-103
16.28	COOC(CH ₃) ₂ COOCH ₃	CI	Et	Ме	Me	0	
16.29	COOC(CH ₃) ₂ COOCH ₃	CI	Et	Ме	Me	1	
16.30	COOC(CH ₃) ₂ COOCH ₃	CI	· Et	Me	Ме	2	
16.31	COOC(CH ₃) ₂ COOCH ₃	F	Et	Me	Ме	0	
16.32	COOC(CH ₃) ₂ COOCH ₃	F	Et	Me	Me	1	
16.33	COOC(CH ₃) ₂ COOCH ₃	F	Et	Me	· Me	2	
16.34	COOC(CH ₃) ₂ COOCH ₃	Н	Et	Me	Me	0	
16.35	COOC(CH ₃) ₂ COOCH ₃	Н	Et	Me	Me	1	
16.36	COOC(CH ₃) ₂ COOCH ₃	Н	Et	Ме	Me	2	
16.37	SCH₂COOCH₃	CI	Et	Ме	Me	0	
16.38	SCH₂COOCH₃	CI	Et	Me	Me	1	
16.39	SCH₂COOCH₃	CI	Et	Ме	Ме	2	
16.40	SCH₂COOCH₃	F	Et	Me	Me .	0	
16.41	SCH₂COOCH₃	F	Et	Me	Ме	1	
16.42	SCH₂COOCH₃	F	Et	Ме	Me	2	
16.43	SCH₂COOCH₃	Н	Et	Ме	Me	0	
16.44	SCH₂COOCH₃	Н	Et	Ме	Ме	1	
16.45	SCH₂COOCH₃	Н	Et	Me	Me	2	<u>'</u>
6.46	Н	CI	Ме	Ме	Et	0	48-50
6.47	Н	CI	Ме	Me	Et	1	113-114
6.48	Н	CI	Me	Me	Et	2	101-103





Comp.	R ₆	R ₄	R ₁	R ₂	R ₃	n	M.p.
No.						"	wi.p.
16.49	Н	F	Me	Me	Et	0	oil
16.50	Н	F	Me	Me	Et	1	87-89
16.51	Н	F	Me	Me	Et	2	87-89
16.52	OCH ₃	CI	Me	Me	Et	0	67-69
16.53	OCH ₃	CI	Me	Me	Et	1	-
16.54	OCH₃	CI	Me	Me	Et	2	
16.55	OCH ₃	F	Me	Me	Et		
16.56	OCH ₃	F	Me	Me	Et	0	
16.57	OCH ₃	F	Me	Me	Et	1	
16.58	O-propargyl	CI	Me	Me		2	
16.59	O-propargyl	CI	Me		Et	0	
16.60	O-propargyl .	CI		Ме	Et	1	
16.61	O-propargyl	F	Me	Ме	Et	2	
16.62	O-propargyl		Me	Ме	Et	0	64-67
6.63		F	Ме	Ме	Et	1	
6.64	O-propargyl	F	Ме	Me	Et	2	103-104
	COOC ₃ H ₇ (iso)	CI	Ме	Me	Et	0	57-60
6.65	COOC ₃ H ₇ (iso)	CI	Ме	Me	Et	1	138-140
6.66	COOC ₃ H ₇ (iso)	CI	Ме	Ме	Et	2	70-72
6.67	COOC ₃ H ₇ (iso)	F	Ме	Me	Et	0	
6.68	COOC ₃ H ₇ (iso)	F	Ме	Me	Et	1	
6.69	COOC ₃ H ₇ (iso)	F	Ме	Ме	Et	2	
6.70	COOC ₃ H ₇ (iso)	Н	Ме	Me	Et	0	· · · · · ·
6.71	COOC ₃ H ₇ (iso)	Н	Ме	Me	Et	1	
6.72	COOC ₃ H ₇ (iso)	Н	Me	Me	Et	2	
6.73	COOC(CH ₃) ₂ COOCH ₃	CI	Ме	Me	Et	0	
6.74	COOC(CH ₃) ₂ COOCH ₃	CI	Me	Me	Et	1	
6.75	COOC(CH ₃) ₂ COOCH ₃	CI	Me	Me	Et	2	





Comp.	R ₆		15				
No.	••6	R₄	R₁	R ₂	R ₃	n	M.p.
							
16,76	COOC(CH ₃) ₂ COOCH ₃	F	Ме	Me	Et	0	
16.77	COOC(CH ₃) ₂ COOCH ₃	F	Me	Me	Et	1	
16.78	COOC(CH ₃) ₂ COOCH ₃	F	Me	Me	Et	2	
16.79	COOC(CH ₃) ₂ COOCH ₃	Н	Me	Me	Et	0	
16.80	COOC(CH ₃) ₂ COOCH ₃	Н	Me	Me	Et	1	
16.81	COOC(CH ₃) ₂ COOCH ₃	Н	Me	Me	Et	2	
16.82	SCH₂COOCH₃	CI	Me	Me	Et	0	
16.83	SCH₂COOCH₃	CI	Me	Me	Et	1	-
16.84	SCH₂COOCH₃	CI	Ме	Ме	Et	2	
16.85	SCH₂COOCH₃	F	Me	Me	Et	0	
16.86	SCH₂COOCH ₃	F	Me	Me	Et	1	
16.87	SCH₂COOCH₃	F	Me	Me	Et	2	
16.88	SCH₂COOCH₃	Н	Ме	Me	Et	0	
16.89	SCH₂COOCH₃	Н	Me	Me	Et	1	
16.90	SCH₂COOCH₃	Н	Me	Ме	Et	2	
16.91	Н	CI	Me	Ме	tert-butyl	0	oil
16.92	Н	CI	Ме	Ме	tert-butyl	1	93-94
16.93	Н	CI	Ме	Me	tert-butyl	2	93-94
16.94	Н	CI	Ме	Me	CHF ₂	0	solid
16.95	Н	CI	Me	Ме	CHF ₂	1	78-79
16.96	Н	CI	Ме	Ме	CHF ₂	2	113-115
16.97	Н	F	Ме	Ме	CHF ₂	0	
16.98	Н	F	Ме	Ме	CHF ₂	1	
16.99	Н .	F	Ме	Ме	CHF ₂	2	
16.100	OCH ₃	CI	Ме	Me	CHF ₂	0	
16.101	OCH ₃	CI	Ме	Me	CHF₂	1	
16.102	OCH ₃	CI	Ме	Me	CHF ₂	2	

Comp.	R ₆	R ₄	R ₁	R ₂	R ₃	n	M.p.
No.							
16.103	OCH₃	F	Me	Me	CHF ₂	0	
16.104	OCH ₃	F	Me	Me	CHF ₂	1	
16.105	OCH ₃	F	Me	Me	CHF ₂	2	
16.106	O-propargyl	CI	Ме	Ме	CHF ₂	0	
16.107	O-propargyl	CI	Me	Me	CHF ₂	1	
16.108	O-propargyl	CI	Me	Me	CHF ₂	2	
16.109	O-propargyl	F	Me	Me	CHF ₂	0	
16.110	O-propargyl	F	Me	Me	CHF ₂	1	
16.111	O-propargyl	F	Ме	Me	CHF ₂	2	<u> </u>
16.112	COOC ₃ H ₇ (iso)	CI	Me	Me	CHF ₂	0	
16.113	COOC ₃ H ₇ (iso)	CI	Ме	Me	CHF ₂	1	
16.114	COOC ₃ H ₇ (iso)	CI	Ме	Me	CHF₂	2	
16.115	COOC ₃ H ₇ (iso)	F	Me	Me	CHF₂	0	
16.116	COOC ₃ H ₇ (iso)	F	Ме	Me	CHF ₂	1	
16.117	COOC ₃ H ₇ (iso)	F	Ме	Me	CHF ₂	2	
16.118	COOC ₃ H ₇ (iso)	Н	Ме	Me	CHF ₂	0	
16.119	COOC ₃ H ₇ (iso)	Н	Ме	Ме	CHF₂	1	
16.120	COOC ₃ H ₇ (iso)	Н	Ме	Ме	CHF ₂	2	
16.121	COOC(CH ₃) ₂ COOCH ₃	CI	Me	Ме	CHF ₂	0	
16.122	COOC(CH ₃) ₂ COOCH ₃	CI	Me	Ме	CHF ₂	1	
16.123	COOC(CH ₃) ₂ COOCH ₃	CI	Me	Ме	CHF ₂	2	
16.124	COOC(CH ₃) ₂ COOCH ₃	F	Me	Ме	CHF ₂	0	
16.125	COOC(CH ₃) ₂ COOCH ₃	F	Ме	Ме	CHF ₂	1	
16.126	COOC(CH ₃) ₂ COOCH ₃	F	Ме	Me	CHF ₂	2	
16.127	COOC(CH ₃) ₂ COOCH ₃	Н	Ме	Ме	CHF ₂	0	
16.128	COOC(CH ₃) ₂ COOCH ₃	Н	Ме	Me	CHF ₂	1	
16.129	COOC(CH ₃) ₂ COOCH ₃	Н	Ме	Ме	CHF ₂	2	

Comp.	R ₆	R ₄	R ₁	R ₂	R ₃	In	M.p.
No.						"	w.p.
16.130	SCH₂COOCH₃	CI	Me	Me	CHF ₂	0	
16.131	SCH₂COOCH₃	CI	Me	Me	CHF ₂	1	
16.132	SCH₂COOCH₃	CI	Me	Me	CHF ₂	2	
16.133	SCH₂COOCH₃	F	Me	Me	CHF ₂	0	
16.134	SCH₂COOCH₃	F	Me	Me	CHF ₂	1	
16.135	SCH₂COOCH ₃	F	Me	Me	CHF ₂	2	
16.136	SCH₂COOCH ₃	Н	Me	Me	CHF ₂	0	
16.137	SCH₂COOCH₃	Н	Me	Me	CHF ₂	1	
16.138	SCH₂COOCH₃	Н	Me	Me	CHF ₂	2	
16.139	OCH ₃	CI	Me	Et	Me	0	
16.140	OCH ₃	CI	Me	Et	Me	1	-
6.141	OCH ₃	CI	Me	Et	Me	2	
6.142	OCH ₃	F	Me	Et	Me	0	
6.143	OCH ₃	F	Me	Et	Me	1	
6.144	OCH ₃	F	Me	Et	Me	2	
6.145	Н	CI	Me	Et	Me	0	oil
6.146	Н	CI	Me	Et	Me	1	oil
6.147	Н	CI	Me	Et	Me	2	72-76
6.148	Н	F	Me	Et	Me	0	72-70
6.149	Н	F	Me	Et	Me	1	
6.150	Н	F	Me	Et	Me	2	
6.151	O-propargyl	CI	Ме	Et	Me	0	
6.152	O-propargyl	CI	Me	Et	Me	1	· ·
6.153	O-propargyl	CI	Ме	Et	Me	2	
6.154	O-propargyl	F	Me	Et	Me	0	
3.155	O-propargyl	F	Me	Et	Me	1	
3.156	O-propargyl	F	Ме	Et	Me	2	



Comp.	R ₆	R ₄	R ₁	R ₂	R ₃		
No.					1.13	n	M.p.
16.157	COOC ₃ H ₇ (iso)	CI	Me	Et	Me	0	
16.158	COOC ₃ H ₇ (iso)	CI	Me	Et	Me	1	ļ
16.159	COOC ₃ H ₇ (iso)	CI	Me	Et	Me	2	
16.160	COOC ₃ H ₇ (iso)	F	Me	Et	Me	0	
16.161	COOC ₃ H ₇ (iso)	F	Me	Et	Me	1	
16.162	COOC ₃ H ₇ (iso)	F	Me	Et	Me	2	
16.163	COOC ₃ H ₇ (iso)	Н	Me	Et	Me	0	<u> </u>
16.164	COOC ₃ H ₇ (iso)	Н	Me	Et	Me	1	
16.165	COOC ₃ H ₇ (iso)	Н	Me	Et	Me	2	
16.166	COOC(CH ₃) ₂ COOCH ₃	CI	Me	Et	Me	0	
16.167	COOC(CH ₃) ₂ COOCH ₃	CI	Me	Et	Me	1	
16.168	COOC(CH ₃) ₂ COOCH ₃	CI	Me	Et	Me	2	
16.169	COOC(CH ₃) ₂ COOCH ₃	F	Me	Et	Me	0	
16.170	COOC(CH ₃) ₂ COOCH ₃	F	Me	Et	Me	1	
16.171	COOC(CH ₃) ₂ COOCH ₃	F	Me	Et	Me	2	
16.172	COOC(CH ₃) ₂ COOCH ₃	H	Me	Et	Me	0	
16.173	COOC(CH ₃) ₂ COOCH ₃	Н	Me	Et	Me	1	
6.174	COOC(CH ₃) ₂ COOCH ₃	H	Me	Et	Me	2	
6.175	CH₂CHCICOOH	CI	Me	Et	Me	0	
6.176	CH₂CHCICOOH	CI	Me	Et	Me	1	
6.177	CH₂CHCICOOH	CI	Me	Et	Me	2	
6.178	CH2CHCICOOH	F	Ме	Et	Me	0	
6.179	CH2CHCICOOH	F	Me	Et	Me	1	
6.180	CH2CHCICOOH	F	Me	Et	Me	2	
6.181	CH ₂ CHCICOOH	Н	Me	Et	Me	0	
6.182	CH₂CHCICOOH	Н	Me	Et	Me	1	
6.183	CH₂CHCICOOH	Н	Me	Et	Me	2	

Comp.	R ₆	R ₄	R ₁	R ₂	R ₃	n	M.p.
No.						"	Mi.p.
16.184	CH₂CHCICOOEt	CI	Me	Et	Me	0	ļ
16.185	CH₂CHCICOOEt	CI	Ме	Et	Me	1	
16.186	CH₂CHCICOOEt	CI	Me	Et	Me	2	
16.187	CH₂CHCICOOEt	F	Me	Et	Me	0	
16.188	CH₂CHCICOOEt	F	Me	Et	Me	1	
16.189	CH₂CHCICOOEt	F	Me	Et	Me	2	
16.190	CH₂CHCICOOEt	H	Me	Et	Me	0	
16.191	CH₂CHCICOOEt	Н	Me	Et	Me	1	
16.192	CH₂CHCICOOEt	Н	Me	Et	Me	2	
16.193	SCH₂COOCH₃	CI	Me	Et	Me	0	
16.194	SCH₂COOCH ₃	CI	Me	Et	Me	1	
16.195	SCH₂COOCH₃	CI	Me	Et	Me	. 2	
16.196	SCH₂COOCH₃	F	Me	Et	Me	0	
16.197	SCH₂COOCH₃	F	Me	Et	Me	1	,
16.198	SCH₂COOCH₃	F	Me	Et	Me	2	
16.199	SCH₂COOCH₃	Н	Me	Et	Me	0	·
16.200	SCH₂COOCH₃	Н	Me	Et	Me	1	
16.201	SCH₂COOCH₃	Н	Me	Et	Me	2	
16.202	CH₂CHCICOOH	CI	Ме	Me	Et	0	
16.203	CH₂CHCICOOH	CI	Ме	Me	Et	1	
6.204	CH₂CHCICOOH	CI	Me	Me	Et	2	
6.205	CH₂CHCICOOH	F	Ме	Me	Et	0	
6.206	CH₂CHCICOOH	F	Me	Me	Et		
6.207	CH₂CHCICOOH	F	Ме	Me	Et	2	
6.208	CH₂CHCICOOH	Н	Ме	Me	Et	0	
6.209	CH₂CHCICOOH	Н	Ме	Me	Et	1	
6.210	CH₂CHCICOOH	H	Ме	Me	Et	2	







Comp.	R ₆	R ₄	R ₁	R ₂	R ₃	ln	100
No.						"	M.p.
16.211	CH₂CHCICOOEt	CI	Me	Me	Et	0	
16.212	CH₂CHCICOOEt	CI	Me	Me	Et	1	-
16.213	CH₂CHCICOOEt	CI	Me	Me	Et	2	
16.214	CH₂CHCICOOEt	F	Me	Me	Et	0	
16.215	CH₂CHClCOOEt	F	Me	Me	Et	1	
16.216	CH ₂ CHCICOOEt	F	Me	Me	Et	2	
16.217	CH ₂ CHClCOOEt	Н	Me	Me	Et	0	
16.218	CH₂CHCICOOEt	H	Me	Me	Et	1	
16.219	CH₂CHCICOOEt	H	Me	Me	Et	2	ļ
16.220	CH2CHCICOOCH3	CI	Me	Me	Et	0	
16.221	CH₂CHCICOOCH₃	CI	Me	Me	Et	1	
16.222	CH₂CHCICOOCH₃	CI	Me	Me	Et		
16.223	CH₂CHCICOOCH₃	F	Me	Me	Et	0	
16.224	CH₂CHCICOOCH₃	F	Me	Me	Et	1	
6.225	CH₂CHCICOOCH₃	F	Me	Me	Et		
6.226	CH₂CHCICOOCH₃	H	Me	Me	Et	0	
6.227	CH₂CHCICOOCH₃	H	Me	Me	Et	1	
6.228	CH₂CHCICOOCH₃	H	Me	Me	Et		
6.229	OCH₂COOEt	F	Me	Me	ET	2	
6.230	OCH₂COOEt	F	Me	Me	Et	0	
6.231	OCH₂COOEt	F	Me	Me	Et	$\frac{1}{2}$	
6.232	Н	CI	Me	Me	CH ₂ CH ₂ CH ₃	2	
6.233	Н	CI	Ме	Me	CH ₂ CH ₂ CH ₃	0	
6.234	Н	CI	Me	Me	CH ₂ CH ₂ CH ₃	1	
6.235	Н	F	Me	Me	CH ₂ CH ₂ CH ₃	2	
6.236	Н	F	Me	Me	CH ₂ CH ₂ CH ₃	0	
5.237	Н	F		Me		1	
				1416	CH₂CH₂CH₃	2	





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Comp. No.	R ₆	R ₄	R ₁	R ₂	R ₃	n	M.p.
16.238	OCH ₃	CI	Me	Ме	CH ₂ CH ₂ CH ₃	0	
16.239	OCH ₃	CI	Me	Ме	CH₂CH₂CH₃	1	
16.240	OCH ₃	CI	Me	Me	CH ₂ CH ₂ CH ₃	2	
16.241	OCH ₃	F	Me	Me	CH ₂ CH ₂ CH ₃	0	
16.242	OCH ₃	F	Me	Ме	CH ₂ CH ₂ CH ₃	1	·
16.243	OCH ₃	F	Me	Me	CH₂CH₂CH₃	2	
16.244	O-propargyl	CI	Ме	Me	CH ₂ CH ₂ CH ₃	0	
16.245	O-propargyl	CI	Me	Me	CH ₂ CH ₂ CH ₃	1	
16.246	O-propargyl	CI	Me	Ме	CH₂CH₂CH₃	2	
16.247	O-propargyl	F	Me	Me	CH ₂ CH ₂ CH ₃	0	
16.248	O-propargyl	F	Me	Me	CH₂CH₂CH₃	1	
16.249	O-propargyl	F	Ме	Ме	CH ₂ CH ₂ CH ₃	2	
16.250	COOC ₃ H ₇ (iso)	CI	Ме	Ме	CH₂CH₂CH₃	0	
16.251	COOC ₃ H ₇ (iso)	CI	Ме	Ме	CH₂CH₂CH₃	1	
16.252	COOC ₃ H ₇ (iso)	CI	Me	Ме	CH₂CH₂CH₃	2	
16.253	COOC ₃ H ₇ (iso)	F	Me	Ме	CH₂CH₂CH₃	0	
16.254	COOC ₃ H ₇ (iso)	F	Me	Ме	CH₂CH₂CH₃	1	
16.255	COOC₃H ₇ (iso)	F	Ме	Ме	CH₂CH₂CH₃	2	
16.256	COOC ₃ H ₇ (iso)	Н	Me	Me	CH ₂ CH ₂ CH ₃	0	
16.257	COOC ₃ H ₇ (iso)	Н	Me	Ме	CH ₂ CH ₂ CH ₃	1	
16.258	COOC ₃ H ₇ (iso)	Н	Ме	Me	CH ₂ CH ₂ CH ₃	2	
16.259	COOC(CH ₃) ₂ COOCH ₃	CI	Ме	Ме	CH ₂ CH ₂ CH ₃	0	
16.260	COOC(CH₃)₂COOCH₃	CI	Me	Ме	CH₂CH₂CH₃	1	
16.261	COOC(CH ₃) ₂ COOCH ₃	CI	Ме	Me	CH₂CH₂CH₃	2	
16.262	COOC(CH ₃) ₂ COOCH ₃	F	Me	Ме	CH₂CH₂CH₃	0	
16.263	COOC(CH ₃)₂COOCH ₃	F	Me	Me	CH₂CH₂CH₃	1	
16.264	COOC(CH ₃) ₂ COOCH ₃	F	Ме	Ме	CH ₂ CH ₂ CH ₃	2	





Comp.	R ₆	R ₄	R ₁	R ₂	R ₃	n	M.p.
No.						1"	lwi.p.
16.265	COOC(CH ₃) ₂ COOCH ₃	Н	Ме	Me	CH₂CH₂CH₃	0	
16.266	COOC(CH ₃) ₂ COOCH ₃	Н	Me	Me	CH ₂ CH ₂ CH ₃	1	
16.267	COOC(CH ₃) ₂ COOCH ₃	Н	Me	Me	CH₂CH₂CH₃	2	
16.268	SCH₂COOCH₃	CI	Ме	Ме	CH₂CH₂CH₃	0	
16.269	SCH₂COOCH₃	CI	Me	Me	CH ₂ CH ₂ CH ₃	1	
16.270	SCH₂COOCH₃	CI	Ме	Me	CH₂CH₂CH₃	2	
16.271	SCH₂COOCH₃	F	Me	Me	CH ₂ CH ₂ CH ₃	0	
16.272	SCH₂COOCH₃	F	Me	Me	CH ₂ CH ₂ CH ₃	1	
16.273	SCH₂COOCH₃	F	Me	Me	CH ₂ CH ₂ CH ₃	2	
16.274	SCH₂COOCH ₃	Н	Me	Me	CH ₂ CH ₂ CH ₃	0	
16.275	SCH₂COOCH₃	H	Me	Me	CH ₂ CH ₂ CH ₃	1	
16.276	SCH₂COOCH₃	Н	Me	Me	CH₂CH₂CH₃	2	
16.277	CH₂CHCICOOH	CI	Me	Me	CH₂CH₂CH₃	0	
16.278	CH₂CHCICOOH	CI	Me	Me	CH ₂ CH ₂ CH ₃	1	
16.279	CH₂CHCICOOH	CI	Me	Me	CH₂CH₂CH₃	2	
16.280	CH₂CHCICOOH	F	Me	Me	CH ₂ CH ₂ CH ₃	0	
16.281	CH₂CHCICOOH	F	Me	Ме	CH ₂ CH ₂ CH ₃	1	
16.282	CH₂CHCICOOH	F	Me	Me	CH ₂ CH ₂ CH ₃	2	-
16.283	CH₂CHCICOOH	Н	Me	Me	CH ₂ CH ₂ CH ₃	0	
16.284	CH2CHCICOOH	Н	Ме	Me	CH₂CH₂CH₃	1	
16.285	CH₂CHCICOOH	Н	Ме	Me	CH ₂ CH ₂ CH ₃	2	
16.286	CH₂CHCICOOEt	CI	Ме	Me	CH ₂ CH ₂ CH ₃	0	
16.287	CH₂CHCICOOEt	CI	Ме	Me	CH₂CH₂CH₃	1	
16.288	CH₂CHCICOOEt	CI	Ме	Me	CH ₂ CH ₂ CH ₃	2	
6.289	CH ₂ CHCICOOEt	F	Ме	Me	CH ₂ CH ₂ CH ₃	0	
6.290	CH₂CHCICOOEt	F	Ме	Me	CH ₂ CH ₂ CH ₃	1	
6.291	CH ₂ CHCICOOEt	F	Ме	Ме	CH₂CH₂CH₃	2	





Comp.	R ₆	R ₄	R ₁	R ₂	R ₃	n	M.p.
No.					1.13		IVI.p.
16.292	CH₂CHCICOOEt	Н	Me	Me	CH₂CH₂CH₃	0	
16.293	CH₂CHCICOOEt	Н	Me	Me	CH ₂ CH ₂ CH ₃	1	
16.294	CH₂CHCICOOEt	Н	Me	Me	CH₂CH₂CH₃	2	
16.295	CH2CHCICOOCH3	CI	Ме	Me	CH₂CH₂CH₃	0	
16.296	CH2CHCICOOCH3	CI	Ме	Me	CH₂CH₂CH₃	1	
16.297	CH₂CHCICOOCH3	CI	Me	Me	CH ₂ CH ₂ CH ₃	2	
16.298	CH2CHCICOOCH3	F	Me	Me	CH ₂ CH ₂ CH ₃	0	
16.299	CH2CHCICOOCH3	F	Me	Me	CH₂CH₂CH₃	1	
16.300	CH2CHCICOOCH3	F	Me	Ме	CH₂CH₂CH₃	2	-
16.301	CH2CHCICOOCH3	Н	Me	Me	CH₂CH₂CH₃	0	
16.302	CH₂CHCICOOCH3	Н	Ме	Ме	CH₂CH₂CH₃	1	
16.303	CH₂CHCICOOCH3	Н	Ме	Ме	CH₂CH₂CH₃	2	
16.304	OCH ₃	F	Me	Ме	CH ₂ CF ₃	0	oil
16.305	OCH₃	F	Me	Me	CH₂CF₃	1	
16.306	OCH₃	F	Ме	Ме	CH ₂ CF ₃	2	
16.307	OCH ₃	CI	Ме	CHF ₂	Ме	0	
16.308	OCH₃	CI	Me	CHF ₂	Ме	1	
16.309	OCH₃	CI	Ме	CHF ₂	Me	2	
16.310	OCH₃	F	Ме	CHF ₂	Me	0	
16.311	OCH₃	F	Ме	CHF ₂	Ме	1	
16.312	OCH₃	F	Ме	CHF ₂	Ме	2	
16.313	Н	CI	Ме	CHF ₂	Ме	0	
6.314	Н	CI	Ме	CHF ₂	Me	1	
6.315	Н	CI	Ме	CHF ₂	Me	2	
6.316	Н .	F	Ме	CHF ₂	Me	0	
6.317	Н .	F	Ме	CHF ₂	Me	1	
6.318	Н	F	Me	CHF₂	Me	2	





Comp.	R ₆	R ₄	R,	R ₂	R ₃	n	M.p.
16.319	O-propargyl	CI	Me	CHF	₂ Me		
16.320		CI	Me			0	
16.321		CI				1	
16.322		F	Me		_	2	
16.323			Me			0	
16.324		F	Me		1	1	
		F	Ме	CHF ₂	Me	2	
16.325		CI	Me	CHF ₂	Ме	0	<u> </u>
16.326		CI	Me	CHF ₂	Ме	1	
16.327	COOC ₃ H ₇ (iso)	CI	Ме	CHF ₂	Me	2	
16.328	COOC ₃ H ₇ (iso)	F	Me	CHF ₂	Me	0	
16.329	COOC ₃ H ₇ (iso)	F	Ме	CHF ₂	Me	1	
16.330	COOC ₃ H ₇ (iso)	F	Me	CHF ₂	Me	2	!
16.331	COOC ₃ H ₇ (iso)	Н	Me	CHF ₂	Me	0	
16.332	COOC ₃ H ₇ (iso)	H	Me	CHF ₂	Me	1	
16.333	COOC ₃ H ₇ (iso)	H	Me	CHF ₂	Me	2	
16.334	COOC(CH ₃) ₂ COOCH ₃	CI	Me	CHF ₂	Me		
16.335	COOC(CH ₃) ₂ COOCH ₃	CI	Me	CHF ₂	Me	0	
16.336	COOC(CH ₃) ₂ COOCH ₃	CI	Me	CHF ₂		1	
16.337	COOC(CH ₃) ₂ COOCH ₃	F	Me		Ме	2	
16.338	COOC(CH ₃) ₂ COOCH ₃	F		CHF ₂	Me	0	
16.339	COOC(CH ₃) ₂ COOCH ₃	F	Me	CHF ₂	Me	1	
16.340			Ме	CHF₂	Me	2	
6.341	COOC(CH₃)₂COOCH₃	Н	Ме	CHF ₂	Me	0	
	COOC(CH ₃) ₂ COOCH ₃	Н	Me	CHF ₂	Ме	1	
6.342	COOC(CH ₃) ₂ COOCH ₃	Н	Me	CHF ₂	Me	2	
6.343	CH₂CHCICOOH	CI	Ме	CHF ₂	Me	0	
6.344	CH ₂ CHCICOOH	CI	Me	CHF ₂	Me	1	
6.345	CH ₂ CHCICOOH	CI	Me	CHF ₂	Me	2	

Comp.	R ₆	R ₄	R ₁	R ₂	R ₃	· .	
No.	·				13	n	M.p.
16.346	CH₂CHCICOOH	F	Ме	CHF:	Me	0	
16.347	CH₂CHCICOOH	F	Me	CHF ₂	Me	1	
16.348	CH₂CHCICOOH	F	Ме	CHF ₂	Me	2	
16.349	CH₂CHCICOOH	Н	Me	CHF ₂	Me	0	
16.350	CH₂CHCICOOH	Н	Me	CHF ₂	Me	1	
16.351	CH₂CHCICOOH	Н	Ме	CHF₂	Me	2	
16.352	CH₂CHCICOOEt	CI	Me	CHF ₂	Me	0	
16.353	CH2CHCICOOEt	Cl	Ме	CHF ₂	Me	1	·
16.354	CH₂CHCICOOEt	. CI	Me	CHF ₂	Ме	2	7
16.355	CH ₂ CHCICOOEt	F	Ме	CHF ₂	Me	0	
16.356	CH ₂ CHClCOOEt	F	Ме	CHF ₂	Me	1	
16.357	CH₂CHCICOOEt	F	Ме	CHF ₂	Ме	2	
16.358	CH₂CHCICOOEt	Н	Me	CHF ₂	Me	0	
16.359	CH₂CHCICOOEt	Н	Ме	CHF ₂	Ме	1	
16.360	CH₂CHCICOOEt	Н	Me	CHF ₂	Ме	2	7
16.361	SCH₂COOCH₃	CI	Me	CHF ₂	Ме	0	
16.362	SCH₂COOCH₃	CI	Ме	CHF ₂	Ме	1	
16.363	SCH₂COOCH₃	CI	Ме	CHF ₂	Ме	2	
16.364	SCH₂COOCH₃	F	Ме	CHF ₂	Ме	0	
16.365	SCH₂COOCH₃	F	Ме	CHF ₂	Ме	1	
16.366	SCH₂COOCH₃	F	Ме	CHF ₂	Ме	2	
16.367	SCH₂COOCH₃	Н	Ме	CHF ₂	Me	0	
16.368	SCH₂COOCH₃	Н	Ме	CHF ₂	Ме	1	
6.369	SCH₂COOCH₃	Н	Ме	CHF ₂	Ме	2	
6.370	OCH ₃	CI	Ме	CF ₃	Me	0	
6.371	OCH ₃	CI	Ме	CF ₃	Ме	1	
6.372	OCH₃	CI	Me	CF ₃	Ме	2	





Comp.	R ₆	R ₄	R ₁	R ₂	R ₃	n	M.p.
No.						"	w.p.
16.373	OCH ₃	F	Me	CF ₃	Me	0	
16.374	OCH ₃	F	Me	CF ₃	Me	1	
16.375	OCH ₃	F	Me	CF ₃	Me	2	
16.376	Н	CI	Me	CF ₃	Me	0	
16.377	Н	CI	Me	CF ₃	Me	1	
16.378	Н	CI	Me	CF ₃	Me	2	
16.379	Н	F	Me	CF₃	Me	0	
16.380	Н	F	Me	CF ₃	Me	1	
16.381	Н	F	Me	CF ₃	Me	2	
16.382	O-propargyl	CI	Me	CF₃	Me	0	
16.383	O-propargyl	CI	Me	CF ₃	Me	1	
16.384	O-propargyl	CI	Me	CF ₃	Me	2	
16.385	O-propargyl	F	Me	CF₃	Me	0	
16.386	O-propargyl	F	Ме	CF₃	Me	1	
16.387	O-propargyl	F	Ме	CF ₃	Me	2	
16.388	COOC ₃ H ₇ (iso)	CI	Me	CF₃	Me	0	
16.389	COOC ₃ H ₇ (iso)	CI	Me	CF ₃	Me	1	
16.390	COOC ₃ H ₇ (iso)	CI	Me	CF ₃	Me	2	
16.391	COOC ₃ H ₇ (iso)	F	Me	CF ₃	Me	0	
16.392	COOC ₃ H ₇ (iso)	F	Me	CF ₃	Me	1	
6.393	COOC ₃ H ₇ (iso)	F	Ме	CF ₃	Me	2	
6.394	COOC ₃ H ₇ (iso)	Н	Me	CF ₃	Me	0	
6.395	COOC ₃ H ₇ (iso)	Н	Ме	CF₃	Me	1	
6.396	COOC ₃ H ₇ (iso)	Н	Ме	CF₃	Me	2	
6.397	COOC(CH ₃) ₂ COOCH ₃	CI	Ме	CF ₃	Ме	0	
6.398	COOC(CH₃)₂COOCH₃	CI	Ме	CF ₃	Ме	1	
6.399	COOC(CH₃)₂COOCH₃	CI	Ме	CF ₃	Me	2	

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Comp.	R ₆	R ₄	R ₁	R ₂	R ₃	n	M.p.
No.							p.
16.400	COOC(CH ₃) ₂ COOCH ₃	F	Me	CF ₃	Me	0	
16.401	COOC(CH ₃) ₂ COOCH ₃	F	Me	CF ₃	Me	1	
16.402	COOC(CH ₃) ₂ COOCH ₃	F	Me	CF ₃	Me	2	
16.403	COOC(CH ₃) ₂ COOCH ₃	H	Me		Me	0	·
16.404	COOC(CH ₃) ₂ COOCH ₃	H	Me		Me	1	
16.405	COOC(CH ₃) ₂ COOCH ₃	Н	Me	CF ₃	Me	2	
16.406	CH₂CHCICOOH	CI	Me	CF ₃	Me		
16.407	CH₂CHCICOOH	CI	Me	CF ₃	Me	0	
16.408	CH₂CHCICOOH	CI	Me	CF ₃		1	
16.409	CH₂CHCICOOH	F	Me		Me	2	
16.410	CH₂CHCICOOH	F		CF ₃	Me	0	
16.411	CH₂CHCICOOH	F	Ме	CF ₃	Ме	1	
16.412	CH ₂ CHClCOOH		Ме	CF ₃	Ме	2	
16.413	CH ₂ CHClCOOH	H	Ме	CF ₃	Ме	0	
16.414	<u></u>	H	Ме	CF₃	Ме	1	
	CH₂CHCICOOH	H	Ме	CF ₃	Me	2	
6.415	CH₂CHCICOOEt	CI	Me	CF ₃	Ме	0	
6.416	CH ₂ CHClCOOEt	CI	Ме	CF ₃	Ме	1	
6.417	CH ₂ CHClCOOEt	CI	Me	CF ₃	Me	2	
6.418	CH₂CHCICOOEt	F	Ме	CF ₃	Me	0	
6.419	CH₂CHClCOOEt	F	Ме	CF ₃	Me	1	
6.420	CH ₂ CHClCOOEt	F	Ме	CF ₃	Ме	2	
6.421	CH₂CHCICOOEt	Н	Me	CF₃	Ме	0	
6.422	CH ₂ CHClCOOEt	Н	Me	CF ₃	Ме		
6.423	CH₂CHCICOOEt	Н	Ме	CF₃	Me	2	
6.424	SCH₂COOCH₃	CI	Me	CF ₃	Me	0	
6.425	SCH₂COOCH3	CI	Me	CF ₃	Me	1	
6.426	SCH₂COOCH₃	CI	Me	CF₃	Me	2	





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Comp. No.	R ₆	R ₄	R ₁	R ₂	R ₃	n	M.p.
16.427	SCH₂COOCH₃	F	Me	CF ₃	Ме	0	
16.428	SCH₂COOCH₃	F	Ме	CF ₃	Me	1	
16.429	SCH₂COOCH₃	F	Me	CF ₃	Me	2	<u> </u>
16.430	SCH₂COOCH₃	Н	Me	CF₃	Me	0	
16.431	SCH₂COOCH₃	Н	Me	CF ₃	Me	1	
16.432	SCH₂COOCH₃	Н	Me	CF ₃	Me	2	
16.433	Н	Н	Et	Ме	Me	0	35-38
16.434	NH ₂	CI	Ме	Me	Et	0	82-83
16.435	I	Н	Et	Ме	Me	0	78-80





Table 17: Compounds of formula Ir

Comp.	R_6	R ₄	R ₅	n	М.р.
17.1	Н	CH₃	CH₃	0	
17.2	Н	CH ₃	CH ₃	1	
17.3	Н	CH ₃	CH ₃	2	
17.4	Н	F	CF ₃	0	oil
17.5	Н	F	CF ₃	1	112-120
17.6	Н	F	. CF ₃	2	121-123
17.7	Н	CI	F	0	oil
17.8	Н	CI	F	1	99-101
17.9	Н	CI	F	2	85-87
17.10	Н	F	F	0	solid
17.11	Н	F	F	1	solid
17.12	Н	F	F	2	-
17.13	COOC(CH₃)₂COOH	Н	Н	0	
17.14	COOC(CH ₃)₂COOH	Н	Н	1	
17.15	COOC(CH ₃)₂COOH	Н	Н	2	139-143
17.16	COOC(CH ₃) ₂ COOCH ₃	Н	H	0	-
7.17	COOC(CH ₃) ₂ COOCH ₃	Н	Н	1	
7.18	COOC(CH ₃) ₂ COOCH ₃	Н	Н	2	oil
7.19	Н	Br	CI	0	84-88
7.20	Н	Br	CI	1	91-93

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Comp.	R ₆	R ₄	10		r
No.		114	R ₅	n	М.р.
17.21	Н	Br	CI	2	123-124

Table 18: Compounds of formula Is

$$R_{5} \xrightarrow{R_{4}} R_{1} \xrightarrow{S(O)} R_{2}$$

$$R_{6} \qquad (Is)$$

Comp. No.	R ₆	R ₅	R ₄	R ₁	R ₂	M.p.
18.1	Н	F	CI	CH ₃	SCH ₃	oil
18.2	Н	CI	F	CH ₃	SCH ₃	88-90
18.3	OCH ₃	CI	F	CH ₃	SCH₃	98-100
18.4	Н	CI	CI	CH ₃	SCH₃	oil
18.5	OCH ₃	CI	F	CH₃	SOCH ₃	171-173
18.6	NO ₂	CI	F	CH ₃	SOCH₃	solid
18.7	Н	CI	F	CH ₃	SOCH ₃	solid
18.8	Н	CF ₃	F	CH ₃	SCH ₃	solid





Table 19: Compounds of formula It

Comp.No.	R ₆	M.p.
19.1	Н	
19.2	CN	
19.3	OCH ₃	
19.4	NHSO₂CH₃	
19.5	OC ₃ H ₇ (iso)	
19.6	O-propargyl	
19.7	OCH(CH₃)C≡CH	
19.8	OCH ₂ COOCH ₂ CH ₃	
19.9	OCH ₂ CH ₂ OCH ₃	
19.10	OCH ₂ CH ₂ SCH ₂ CH ₃	
19.11	OCH ₂ COOCH ₃	
19.12	OCH ₂ COOC ₅ H ₁₁ (n)	
19.13	OCH₂COO-benzyl	
19.14	OCH(CH ₃)COObenzyl	
19.15	SC ₃ H ₇ (iso)	
19.16	SCH₂COOCH₃	
9.17	SCH ₂ COOC ₂ H ₅	
9.18	SCH(CH ₃)COObenzyl	
9.19	SCH₂COObenzyl	
9.20	COOCH ₃	
9.21	COOC ₃ H ₇ (iso)	

Comp.No.	R ₆	M.p.
19.22	COOC(CH ₃) ₂ COOH	
19.23	COOC(CH ₃)₂COO-allyl	
19.24	COOC(CH ₃)₂COOCH ₃	
19.25	COOC(CH ₃)₂COOethyl	
19.26	COOC(CH ₃) ₂ CONH-allyi	
19.27	CH₂CHClCOOethyl	
19.28	CH ₂ CH=CH ₂	
19.29	CH₂CH₂CH₃	
19.30	CH ₂ CH ₂ CF ₃	
19.31	OCH(CH ₃)COOC ₂ H ₅ (R)	
19.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
19.33	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
19.34	CH₂CHCICOOH	
19.35	CH₂CHCICOOCH₃	
19.36	CH₂CHCICOOC₃H₁(iso)	
19.37	CH ₂ CHCICONHallyl	
19.38	CH₂C(CH₃)CICOOH	
19.39	CH₂C(CH₃)CICOOCH₃	
19.40	CH₂C(CH₃)CICOOEt	
19.41	CH₂C(CH₃)CICONHEt	
19.42	CH ₂ CH ₂ COOH	
19.43	CH ₂ CH ₂ COOCH ₃	
19.44	CH ₂ CH ₂ COOEt	
19.45	CHCICHCICOOH	
19.46	CHCICHCICOOCH3	
19.47	CHCICHCICOOEt	
19.48	CH₂CH(OCH₃)COOH	·
19.49	CH₂CH(OCH₃)COOCH₃	





Comp.No.	R ₆	M.p.
19.50	CH₂CH(OCH₃)COOEt	
19.51	CH₂CH(SCH₃)COOH	
19.52	CH₂CH(SCH₃)COOCH₃	
19.53	CH₂CH(SCH₃)COOEt	
19.54	CH=CHCOOH	
19.55	CH=CHCOOCH ₃	
19.56	CH=CHCOOEt	
19.57	CH=CCICOOH	
19.58	CH=CCICOOCH ₃	
19.59	COOEt	
19.60	CONH₂	
19.61	-C(O)OCH ₂	
19.62	CONHSO₂CH₃	
19.63	COOCH₂COOH	
19.64	COOCH₂COOCH₃	
19.65	COOCH(CH3)COOH	
19.66	COOCH(CH ₃)COOCH ₃	
19.67	COOCH(CH₃)CH₂COOH	
19.68	COOCH(CH₃)CH₂COOCH₃	· ·
19.69	COOC(CH ₃) ₂ CN	
19.70	COOCH₂CH₂OCH₃	
19.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
19.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	-
9.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
9.74	COOCH₂C≡CH	
9.75	COOC(CH ₃) ₂ COOCH ₂ C=CH	

Comp.No.	R_6	M.p.
19.76	COOCH(CH₃)C≡CH	
19.77	COOC(CH ₃) ₂ COCH ₃	
19.78	NHallyl	
19.79	N(COCH₃)aliyl	
19.80	N(Et)SO ₂ CH ₃	
19.81	N(allyl)SO ₂ CH ₃	
19.82	N(allyl)SO ₂ Et	
19.83	SO ₂ N(CH ₃) ₂	
19.84	SO₂NH₂	-
19.85	SO₂NHCOCH ₃	
19.86	ОН	
19.87	OEt	
19.88	Oallyl	
19.89	OCH ₂ C≡CCH ₃	
19.90	OCH(CH ₃)CH=CH ₂	
19.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
19.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
19.93	0	
	OCH ₂	
19.94	OCH₂CH₂NHCH₃	
19.95	OCH₂CH₂N(CH₃)COCH₃	
19.96	OCH₂CH₂COOH	
9.97	OC(CH ₃)₂COOH	
9.98	OC(CH ₃)₂COOCH ₃	
9.99	OC(CH ₃)₂COOEt	
9.100	OCH₂COOH ·	
9.101	OSO ₂ CH ₃	
9.102	OSO ₂ CF ₃	





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Comp.No.	R ₆	M.p.
19.103	CH₂CHCICOOC₂H₅	
19.104	CH ₂ CHCICON(C ₂ H ₅) ₂	
19.105	CH2CHCICONHOH	
19.106	CH ₂ CHCICOOCH ₂ C ₆ H ₅	
19.107	CH₂CH(CH₃)COOH	
19.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
19.109	-COOСН ₂ —	
19.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
19.111	-COOC(CH ₃) ₂ COOCH ₂ —	
19.112	COOC(CH ₃)₂CONHCH₂C=CH	
19.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
19.114	OCH ₂ —	





Table 20: Compounds of formula lu

Comp.No.	R ₆	M.p.
20.1	Н	·
20.2	CN	
20.3	OCH ₃	
20.4	NHSO₂CH₃	
20.5	OC ₃ H ₇ (iso)	
20.6	O-propargyl	
20.7	OCH(CH₃)C≡CH	
20.8	OCH₂COOCH₂CH₃	· · · · · · · · · · · · · · · · · · ·
20.9	OCH₂CH₂OCH₃	
20.10	OCH ₂ CH ₂ SCH ₂ CH ₃	
20.11	OCH ₂ COOCH ₃	
20.12	OCH ₂ COOC ₅ H ₁₁ (n)	
20.13	OCH₂COO-benzyl	
20.14	OCH(CH₃)COObenzyl	-
20.15	SC ₃ H ₇ (iso)	
20.16	SCH₂COOCH₃	
20.17	SCH₂COOC₂H₅	
20.18	SCH(CH₃)COObenzyl	
20.19	SCH₂COObenzyl	
20.20	COOCH₃	





Comp.No.	R ₆	M.p.
20.21	COOC ₃ H ₇ (iso)	
20.22	COOC(CH₃)₂COOH	
20.23	COOC(CH ₃) ₂ COO-allyl	
20.24	COOC(CH₃)₂COOCH₃	
20.25	COOC(CH ₃) ₂ COOethyl	
20.26	COOC(CH₃)₂CONH-allyl	
20.27	CH₂CHClCOOethyl	
20.28	CH ₂ CH=CH ₂	
20.29	CH ₂ CH ₂ CH ₃	
20.30	CH ₂ CH ₂ CF ₃	
20.31	OCH(CH ₃)COOC ₂ H ₅ (R)	
20.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
20.33	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
20.34	CH₂CHCICOOH	
20.35	CH₂CHCICOOCH₃	
20.36	CH ₂ CHClCOOC ₃ H ₇ (iso)	
20.37	CH₂CHCICONHallyl	
20.38	CH₂C(CH₃)CICOOH	
20.39	CH ₂ C(CH ₃)CICOOCH ₃	
20.40	CH₂C(CH₃)CICOOEt	
20.41	CH₂C(CH₃)CICONHEt	
20.42	CH ₂ CH ₂ COOH	
20.43	CH ₂ CH ₂ COOCH ₃	
20.44	CH ₂ CH ₂ COOEt	
20.45	СНСІСНСІСООН	
20.46	CHCICHCICOOCH ₃	· ·
20.47	CHCICHCICOOEt	
20.48	CH₂CH(OCH₃)COOH	

Comp.No.	R ₆	M.p.
20.49	CH₂CH(OCH₃)COOCH₃	
20.50	CH ₂ CH(OCH ₃)COOEt	
20.51	CH₂CH(SCH₃)COOH	
20.52	CH₂CH(SCH₃)COOCH₃	
20.53	CH₂CH(SCH₃)COOEt	
20.54	CH=CHCOOH	
20.55	CH=CHCOOCH ₃	
20.56	CH=CHCOOEt	
20.57	CH=CCICOOH	
20.58	CH=CCICOOCH ₃	
20.59	COOEt	
20.60	CONH ₂	
20.61	-C(O)OCH ₂	· · · · · · · · · · · · · · · · · · ·
20.62	CONHSO₂CH₃	
20.63	COOCH₂COOH	
20.64	COOCH₂COOCH₃	
20.65	COOCH(CH3)COOH	
20.66	COOCH(CH ₃)COOCH ₃	
20.67	COOCH(CH₃)CH₂COOH	
20.68	COOCH(CH₃)CH₂COOCH₃	
20.69	COOC(CH ₃)₂CN	
20.70	COOCH₂CH₂OCH₃	
20.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
20.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
20.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
20.74	COOCH₂C≡CH	

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Comp.No.	R ₆	M.p.
20.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	
20.76	COOCH(CH₃)C≡CH	
20.77	COOC(CH ₃) ₂ COCH ₃	
20.78	NHallyl	
20.79	N(COCH₃)allyl	
20.80	N(Et)SO ₂ CH ₃	
20.81	N(allyl)SO₂CH₃	
20.82	N(allyl)SO ₂ Et	
20.83	SO₂N(CH ₃)₂	
20.84	SO₂NH₂	
20.85	SO₂NHCOCH₃	
20.86	ОН	
20.87	OEt	
20.88	Oallyl	
0.89	OCH ₂ C≡CCH ₃	
0.90	OCH(CH ₃)CH=CH ₂	
0.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
0.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
0.93		
	OCH ₂	
0.94	OCH ₂ CH ₂ NHCH ₃	
0.95	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
0.96	OCH ₂ CH ₂ COOH	
0.97	OC(CH₃)₂COOH	
0.98	OC(CH ₃) ₂ COOCH ₃	
0.99	OC(CH ₃) ₂ COOEt	
.100	OCH₂COOH	
.101	OSO₂CH₃	

Comp.No.	R ₆	M.p.
20.102	OSO₂CF ₃	
20.103	CH₂CHCICOOC₂H₅	
20.104	CH₂CHCICON(C₂H₅)₂	
20.105	CH₂CHCICONHOH	
20.106	CH₂CHCICOOCH₂C ₆ H ₅	
20.107	CH₂CH(CH₃)COOH	
20.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
20.109	-COOCH ₂ —	
20.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
20.111	-cooc(ch ₃) ₂ соосн ₂ —Д	
20.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
20.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
20.114	OCH ₂	





Table 21: Compounds of formula lv

Comp.No.	R ₆	M.p.
21.1	Н	
21.2	CN	
21.3	OCH ₃	
21.4	NHSO ₂ CH ₃	
21.5	OC ₃ H ₇ (iso)	
21.6	O-propargyl	· .
21.7	OCH(CH₃)C≡CH	
21.8	OCH ₂ COOCH ₂ CH ₃	
21.9	OCH₂CH₂OCH₃	
21.10	OCH ₂ CH ₂ SCH ₂ CH ₃	
21.11	OCH₂COOCH₃	
21.12	OCH ₂ COOC ₅ H ₁₁ (n)	
21.13	OCH₂COO-benzyl	
21.14	OCH(CH ₃)COObenzyl	
21.15	SC ₃ H ₇ (iso)	
21.16	SCH₂COOCH₃	
21.17	SCH ₂ COOC ₂ H ₅	
21.18	SCH(CH₃)COObenzyl	
21.19	SCH₂COObenzyl	
21.20	COOCH ₃	
1.21	COOC ₃ H ₇ (iso)	

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Comp.No.	R ₆	M.p.
21.22	COOC(CH ₃)₂COOH	
21.23	COOC(CH ₃)₂COO-allyl	
21.24	COOC(CH ₃) ₂ COOCH ₃	
21.25	COOC(CH ₃) ₂ COOethyl	
21.26	COOC(CH₃)₂CONH-aliyl	
21.27	CH₂CHClCOOethyl	
21.28	CH ₂ CH=CH ₂	
21.29	CH ₂ CH ₂ CH ₃	
21.30	CH ₂ CH ₂ CF ₃	
21.31	OCH(CH₃)COOC₂H₅(R)	
21.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
21.33	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
21.34	CH₂CHCICOOH	
21.35	CH₂CHCICOOCH₃	
21.36	CH₂CHCICOOC₃H₁(iso)	
21.37	CH₂CHCICONHallyl	
21.38	CH₂C(CH₃)CICOOH	
1.39	CH₂C(CH₃)CICOOCH₃	·
1.40	CH₂C(CH₃)CICOOEt	
1.41	CH₂C(CH₃)CICONHEt	· · · · · · · · · · · · · · · · · · ·
1.42	CH₂CH₂COOH	
1.43	CH ₂ CH ₂ COOCH ₃	
1.44	CH ₂ CH ₂ COOEt	
1.45	CHCICHCICOOH	
1.46	CHCICHCICOOCH3	
1.47	CHCICHCICOOEt	
1.48	CH ₂ CH(OCH ₃)COOH	
1.49	CH₂CH(OCH₃)COOCH₃	

Comp.No.	R ₆	M.p.
21.50	CH₂CH(OCH₃)COOEt	
21.51	CH₂CH(SCH₃)COOH	
21.52	CH₂CH(SCH₃)COOCH₃	
21.53	CH₂CH(SCH₃)COOEt	
21.54	CH=CHCOOH	
21.55	CH=CHCOOCH ₃	
21.56	CH=CHCOOEt	
21.57	CH=CCICOOH	
21.58	CH=CCICOOCH ₃	
21.59	COOEt	
21.60	CONH ₂	· · · · · · · · · · · · · · · · · · ·
21.61	-C(O)OCH ₂	·
21.62	CONHSO₂CH₃	
21.63	COOCH₂COOH	
21.64	COOCH₂COOCH₃	
21.65	COOCH(CH₃)COOH	
21.66	COOCH(CH ₃)COOCH ₃	
21.67	COOCH(CH₃)CH₂COOH	
21.68	COOCH(CH₃)CH₂COOCH₃	
21.69	COOC(CH₃)₂CN	
21.70	COOCH₂CH₂OCH₃	
21.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
21.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
21.73	COOC(CH ₃)₂COOCH₂PHENYL	
21.74	COOCH ₂ C≡CH	
21.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	

Comp.No.	R ₆	M.p.
21.76	COOCH(CH₃)C≡CH	·
21.77	COOC(CH ₃)₂COCH ₃	
21.78	NHallyl	
21.79	N(COCH ₃)allyl	
21.80	N(Et)SO ₂ CH ₃	
21.81	N(allyl)SO ₂ CH ₃	
21.82	N(allyl)SO₂Et	
21.83	SO ₂ N(CH ₃) ₂	
21.84	SO ₂ NH ₂	·
21.85	SO₂NHCOCH₃	
21.86	ОН	
21.87	OEt	
21.88	Oallyl	
21.89	OCH₂C≡CCH ₃	
21.90	OCH(CH ₃)CH=CH ₂	
21.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
21.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
21.93	OCH ₂	
21.94	OCH ₂ CH ₂ NHCH ₃	· ·
21.95	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
21.96	OCH ₂ CH ₂ COOH	· .
21.97	OC(CH ₃)₂COOH	
21.98	OC(CH ₃) ₂ COOCH ₃	
21.99	OC(CH ₃)₂COOEt	
21.100	OCH₂COOH	
21.101	OSO₂CH₃	
21.102	OSO₂CF₃	





Comp.No.	R ₆	M.p.
21.103	CH₂CHCICOOC₂H₅	
21.104	CH ₂ CHCICON(C ₂ H ₅) ₂	
21.105	CH₂CHCICONHOH	
21.106	CH₂CHCICOOCH₂C ₆ H ₅	
21.107	CH₂CH(CH₃)COOH	
21.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
21.109	-соосн ₂ —	
21.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
21.111	-cooc(ch ₃) ₂ соосн ₂ —Д	
21.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
21.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
21.114	OCH ₂ —	

Table 22: Compounds of formula lw

$$\operatorname{Br} \xrightarrow{\operatorname{F} \operatorname{CH}_3} \operatorname{SCH}_3$$

$$\operatorname{CH}_3$$

Comp.No.	R ₆	M.p.
22.1	Н	
22.2	CN	
22.3	OCH ₃	
22.4	NHSO₂CH ₃	
22.5	OC ₃ H ₇ (iso)	
22.6	O-propargyl	
22.7	OCH(CH₃)C≡CH	
22.8	OCH ₂ COOCH ₂ CH ₃	
22.9	OCH ₂ CH ₂ OCH ₃	
22.10	OCH ₂ CH ₂ SCH ₂ CH ₃	
22.11	OCH ₂ COOCH ₃	
22.12	OCH ₂ COOC ₅ H ₁₁ (n)	
22.13	OCH₂COO-benzyl	
22.14	OCH(CH ₃)COObenzyl	
22.15	SC ₃ H ₇ (iso)	
22.16	SCH₂COOCH₃	
22.17	SCH ₂ COOC ₂ H ₅	
22.18	SCH(CH ₃)COObenzyl	
22.19	SCH₂COObenzyl	
22.20	COOCH ₃	
22.21	COOC₃H ₇ (iso)	

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Comp.No.	R ₆	M.p.
22.22	COOC(CH ₃) ₂ COOH	
22.23	COOC(CH ₃) ₂ COO-allyl	
22.24	COOC(CH ₃)₂COOCH ₃	
22.25	COOC(CH ₃) ₂ COOethyl	
22.26	COOC(CH ₃) ₂ CONH-allyl	
22.27	CH₂CHClCOOethyl	
22.28	CH ₂ CH=CH ₂	
22.29	CH ₂ CH ₂ CH ₃	
22.30	CH ₂ CH ₂ CF ₃	
22.31	OCH(CH ₃)COOC ₂ H ₅ (R)	
22.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
22.33	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
22.34	CH₂CHCICOOH	
22.35	CH ₂ CHClCOOCH ₃	
22.36	CH ₂ CHClCOOC ₃ H ₇ (iso)	
22.37	CH₂CHClCONHallyI	
22.38	CH ₂ C(CH ₃)CICOOH	
22.39	CH ₂ C(CH ₃)CICOOCH ₃	
22.40	CH₂C(CH₃)CICOOEt	
22.41	CH₂C(CH₃)CICONHEt	· ·
22.42	CH ₂ CH ₂ COOH	
22.43	CH ₂ CH ₂ COOCH ₃	
22.44	CH ₂ CH ₂ COOEt	
22.45	CHCICHCICOOH	
22.46	CHCICHCICOOCH ₃	
22.47	CHCICHCICOOEt	
22.48	CH ₂ CH(OCH ₃)COOCH ₃	

Comp.No.	R ₆	M.p.
22.50	CH₂CH(OCH₃)COOEt	
22.51	CH₂CH(SCH₃)COOH	
22.52	CH₂CH(SCH₃)COOCH₃	
22.53	CH₂CH(SCH₃)COOEt	
22.54	CH=CHCOOH	
22.55	CH=CHCOOCH ₃	· ·
22.56	CH=CHCOOEt	
22.57	CH=CCICOOH	
22.58	CH=CCICOOCH ₃	
22.59	COOEt .	
22.60	CONH ₂	
22.61	-C(O)OCH ₂	
22.62	CONHSO₂CH ₃	· · · · · · · · · · · · · · · · · · ·
22.63	COOCH₂COOH	
22.64	COOCH₂COOCH₃	
22.65	COOCH(CH ₃)COOH	
22.66	COOCH(CH ₃)COOCH ₃	
22.67	COOCH(CH ₃)CH ₂ COOH	
22.68	COOCH(CH ₃)CH ₂ COOCH ₃	
22.69	COOC(CH ₃)₂CN	
22.70	COOCH₂CH₂OCH₃	
22.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
22.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
22.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
22.74	COOCH₂C≡CH	
22.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	





Comp.No.	R ₆	M.p.
22.76	COOCH(CH ₃)C≡CH	
22.77	COOC(CH ₃) ₂ COCH ₃	
22.78	NHallyl	
22.79	N(COCH₃)allyl	
22.80	N(Et)SO ₂ CH ₃	
22.81	N(allyl)SO ₂ CH ₃	
22.82	N(allyl)SO₂Et	
22.83	SO ₂ N(CH ₃) ₂	
22.84	SO ₂ NH ₂	
22.85	SO₂NHCOCH₃	
22.86	ОН	
22.87	OEt	
22.88	Oallyl	
22.89	OCH ₂ C≡CCH ₃	
22.90	OCH(CH ₃)CH=CH ₂	
22.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
22.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	· ·
22.93	OCH ₂	
22.94	OCH₂CH₂NHCH₃	
22.95	OCH₂CH₂N(CH₃)COCH₃	
2.96	OCH₂CH₂COOH	
2.97	OC(CH₃)₂COOH	
2.98	OC(CH ₃)₂COOCH ₃	
2.99	OC(CH ₃) ₂ COOEt	
2.100	OCH₂COOH	
2.101	OSO ₂ CH ₃	
2.102	OSO₂CF ₃	

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Comp.No.	R ₆	M.p.
22.103	CH₂CHCICOOC₂H₅	
22.104	CH ₂ CHCICON(C ₂ H ₅) ₂	
22.105	CH₂CHCICONHOH	
22.106	CH2CHCICOOCH2C6H5	
22.107	CH₂CH(CH₃)COOH	
22.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
22.109	-COOCH ₂	
22.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
22.111	-COOC(CH ₃) ₂ COOCH ₂ —	
22.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
22.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
22.114	OCH ₂	





Table 23: Compounds of formula lx

Comp.No.	R ₆	M.p.
23.1	Н	
23.2	CN	
23.3	OCH₃	
23.4	NHSO ₂ CH ₃	
23.5	OC ₃ H ₇ (iso)	
23.6	O-propargyl	
23.7	OCH(CH₃)C≡CH	
23.8	OCH ₂ COOCH ₂ CH ₃	
23.9	OCH₂CH₂OCH₃	
23.10	OCH ₂ CH ₂ SCH ₂ CH ₃	
23.11	OCH₂COOCH₃	
23.12	OCH ₂ COOC ₅ H ₁₁ (n)	
23.13	OCH₂COO-benzyl	
23.14	OCH(CH ₃)COObenzyl	
23.15	SC ₃ H ₇ (iso)	
23.16	SCH₂COOCH₃	
23.17	SCH ₂ COOC ₂ H ₅	
23.18	SCH(CH ₃)COObenzyl	
23.19	SCH₂COObenzyl	
23.20	COOCH ₃	
23.21	COOC ₃ H ₇ (iso)	

Comp.No.	R ₆	M.p.
23.22	COOC(CH₃)₂COOH	· ·
23.23	COOC(CH ₃) ₂ COO-allyl	
23.24	COOC(CH ₃) ₂ COOCH ₃	
23.25	COOC(CH ₃) ₂ COOethyl	
23.26	COOC(CH₃)₂CONH-allyl	
23.27	CH₂CHCICOOethyl	
23.28	CH₂CH=CH₂	
23.29	CH₂CH₂CH₃	
23.30	CH ₂ CH ₂ CF ₃	
23.31	OCH(CH₃)COOC₂H₅(R)	
23.32	OCH(CH ₃)COOC₂H ₅ (S)	
23.33	OCH(CH ₃)COOC₂H ₅ (R,S)	
23.34	CH₂CHCICOOH	
23.35	CH₂CHCICOOCH ₃	
23.36	CH ₂ CHClCOOC ₃ H ₇ (iso)	
23.37	CH₂CHCICONHallyI	
23.38	CH₂C(CH₃)CICOOH	
23.39	CH₂C(CH₃)CICOOCH₃	
23.40	CH₂C(CH₃)CICOOEt	
23.41	CH₂C(CH₃)CICONHEt	
23.42	CH₂CH₂COOH	
23.43	CH₂CH₂COOCH₃	
23.44	CH ₂ CH ₂ COOEt	
23.45	CHCICHCICOOH	
23.46	CHCICHCICOOCH3	
23.47	CHCICHCICOOEt	
23.48	CH ₂ CH(OCH ₃)COOH	
23.49	CH₂CH(OCH₃)COOCH₃	

Comp.No.	R ₆	M.p.
23.50	CH₂CH(OCH₃)COOEt	
23.51	CH₂CH(SCH₃)COOH	
23.52	CH₂CH(SCH₃)COOCH₃	
23.53	CH ₂ CH(SCH ₃)COOEt	
23.54	CH=CHCOOH	
23.55	CH=CHCOOCH ₃	
23.56	CH=CHCOOEt	
23.57	CH=CCICOOH	
23.58	CH=CCICOOCH ₃	
23.59	COOEt	
23.60	CONH ₂	
23.61	-C(O)OCH₂ △	
23.62	CONHSO₂CH₃	
23.63	COOCH₂COOH	
23.64	COOCH₂COOCH₃	
23.65	COOCH(CH ₃)COOH	
23.66	COOCH(CH ₃)COOCH ₃	
23.67	COOCH(CH₃)CH₂COOH	
23.68	COOCH(CH₃)CH₂COOCH₃	
23.69	COOC(CH ₃) ₂ CN	
23.70	COOCH₂CH₂OCH₃	
23.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
23.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
23.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
23.74	COOCH₂C≡CH	
3.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	

Comp.No.	R ₆	M.p.
23.76	COOCH(CH₃)C≡CH	
23.77	COOC(CH ₃) ₂ COCH ₃	
23.78	NHallyl	
23.79	N(COCH ₃)allyl	
23.80	N(Et)SO₂CH₃	
23.81	N(allyl)SO ₂ CH ₃	
23.82	N(allyl)SO₂Et	
23.83	SO₂N(CH ₃)₂	
23.84	SO₂NH₂	-
23.85	SO₂NHCOCH₃	
23.86	ОН	·
23.87	OEt	
23.88	Oallyl	
23.89	OCH ₂ C≡CCH ₃	
23.90	OCH(CH₃)CH=CH₂	
23.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
23.92	OCH₂CH₂OCH₂CH₂OCH₃	
23.93	OCH ₂	
23.94	OCH₂CH₂NHCH₃	
23.95	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
23.96	OCH ₂ CH ₂ COOH	
23.97	OC(CH ₃) ₂ COOH	
23.98	OC(CH ₃) ₂ COOCH ₃	
23.99	OC(CH ₃) ₂ COOEt	
23.100	OCH₂COOH	
23.101	OSO₂CH₃	
23.102	OSO ₂ CF ₃	







Comp.No.	R ₆	M.p.
23.103	CH₂CHCICOOC₂H₅	•
23.104	CH ₂ CHCICON(C ₂ H ₅) ₂	
23.105	CH₂CHCICONHOH	
23.106	CH₂CHCICOOCH₂C₅H₅	
23.107	CH₂CH(CH₃)COOH	
23.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
23.109	-COOCH ₂ —	
23.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
23.111	-COOC(CH ₃) ₂ COOCH ₂ —	
23.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
23.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
23.114	OCH ₂	

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Table 24: Compounds of formula ly

$$\mathsf{Br} \overset{\mathsf{F}}{\underset{\mathsf{R}_{6}}{\bigvee}} \mathsf{CH}_{3} \\ \mathsf{CH}_{3} \\ \mathsf{CH}_{3} \\ \mathsf{CH}_{3} \\ \mathsf{CH}_{3}$$

Comp.No.	R ₆	M.p.
24.1	Н	
24.2	CN	
24.3	OCH ₃	
24.4	NHSO₂CH₃	
24.5	OC ₃ H ₇ (iso)	
24.6	O-propargyl	
24.7	OCH(CH₃)C≡CH	
24.8	OCH ₂ COOCH ₂ CH ₃	
24.9	OCH ₂ CH ₂ OCH ₃	
24.10	OCH ₂ CH ₂ SCH ₂ CH ₃	
24.11	OCH₂COOCH₃	
24.12	OCH ₂ COOC ₅ H ₁₁ (n)	
24.13	OCH₂COO-benzyl	
24.14	OCH(CH ₃)COObenzyl	
24.15	SC ₃ H ₇ (iso)	
24.16	SCH₂COOCH₃	
24.17	SCH ₂ COOC ₂ H ₅	
24.18	SCH(CH₃)COObenzyl	
24.19	SCH₂COObenzyl	
24.20	COOCH ₃	
24.21	COOC ₃ H ₇ (iso)	

Comp.No.	R ₆	M.p.
24.22	COOC(CH ₃)₂COOH	
24.23	COOC(CH ₃) ₂ COO-allyl	
24.24	COOC(CH ₃) ₂ COOCH ₃	
24.25	COOC(CH ₃) ₂ COOethyl	
24.26	COOC(CH ₃) ₂ CONH-allyl	
24.27	CH₂CHClCOOethyl	
24.28	CH ₂ CH=CH ₂	
24.29	CH₂CH₂CH₃	
24.30	CH₂CH₂CF₃	
24.31	OCH(CH ₃)COOC ₂ H ₅ (R)	
24.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
24.33	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
24.34	CH₂CHCICOOH	
24.35	CH₂CHCICOOCH₃	
24.36	CH ₂ CHCICOOC ₃ H ₇ (iso)	
24.37	CH₂CHClCONHallyl	
24.38	CH₂C(CH₃)CICOOH	
24.39	CH₂C(CH₃)CICOOCH₃	
24.40	CH₂C(CH₃)CICOOEt	
24.41	CH₂C(CH₃)CICONHEt	
24.42	CH₂CH₂COOH	
24.43	CH₂CH₂COOCH₃	
24.44	CH₂CH₂COOEt	
24.45	CHCICHCICOOH	
24.46	CHCICHCICOOCH3	
4.47	CHCICHCICOOEt	
4.48	CH₂CH(OCH₃)COOH	
4.49	CH₂CH(OCH₃)COOCH₃	

Comp.No.	R ₆	M.p.
24.50	CH₂CH(OCH₃)COOEt	
24.51	CH₂CH(SCH₃)COOH	
24.52	CH₂CH(SCH₃)COOCH₃	
24.53	CH₂CH(SCH₃)COOEt	
24.54	CH=CHCOOH	
24.55	CH=CHCOOCH ₃	
24.56	CH=CHCOOEt	·
24.57	CH=CCICOOH	
24.58	CH=CCICOOCH ₃	
24.59	COOEt	
24.60	CONH ₂	
24.61	-C(O)OCH ₂ O	
24.62	CONHSO₂CH ₃	
24.63	COOCH₂COOH	
24.64	COOCH₂COOCH₃	
24.65	COOCH(CH3)COOH	· · · · · · · · · · · · · · · · · · ·
24.66	COOCH(CH3)COOCH3	
24.67	COOCH(CH₃)CH₂COOH	
24.68	COOCH(CH₃)CH₂COOCH₃	
24.69	COOC(CH ₃)₂CN	
24.70	COOCH ₂ CH ₂ OCH ₃	
24.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
24.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
24.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
24.74	COOCH₂C≡CH	
24.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	





Comp.No.	R ₆	M.p.
24.76	COOCH(CH₃)C≡CH	
24.77	COOC(CH ₃)₂COCH ₃	
24.78	NHallyl	
24.79	N(COCH₃)allyl	
24.80	N(Et)SO ₂ CH ₃	
24.81	N(allyl)SO ₂ CH ₃	
24.82	N(allyl)SO₂Et	
24.83	SO ₂ N(CH ₃) ₂	
24.84	SO ₂ NH ₂	
24.85	SO₂NHCOCH₃	
24.86	ОН	
24.87	OEt	
24.88	Oallyl	
24.89	OCH₂C≡CCH₃	
24.90	OCH(CH ₃)CH=CH ₂	
24.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
24.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
24.93	OCH ₂ —O	
24.94	OCH₂CH₂NHCH₃	
24.95	OCH₂CH₂N(CH₃)COCH₃	
24.96	OCH ₂ CH ₂ COOH	
24.97	OC(CH₃)₂COOH	
24.98	OC(CH ₃)₂COOCH ₃	
24.99	OC(CH ₃)₂COOEt	
24.100	OCH₂COOH	
24.101	OSO₂CH₃	
4.102	OSO ₂ CF ₃	

Comp.No.	R ₆	M.p.
24.103	CH₂CHCICOOC₂H₅	
24.104	CH ₂ CHCICON(C ₂ H ₅) ₂	
24.105	CH₂CHCICONHOH	
24.106	CH₂CHCICOOCH₂C ₆ H ₅	
24.107	CH₂CH(CH₃)COOH	
24.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
24.109	-COOCH ₂	
24.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
24.111	-COOC(CH ₃) ₂ COOCH ₂	· ·
24.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
24.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
24.114	OCH ₂	





Table 25: Compounds of formula Iz

$$\begin{array}{c|c} & \text{CI} & \text{CH}_3 \\ & & \text{SCH}_3 \\ & & \text{CH}_3 \end{array} \tag{Iz}$$

Comp.No.	R ₆	M.p.
25.1	Н	·
25.2	CN	
25.3	OCH₃	
25.4	NHSO₂CH₃	
25.5	OC ₃ H ₇ (iso)	
25.6	O-propargyl	
25.7	OCH(CH₃)C≡CH	
25.8	OCH ₂ COOCH ₂ CH ₃	
25.9	OCH₂CH₂OCH₃	
25.10	OCH₂CH₂SCH₂CH₃	
25.11	OCH₂COOCH₃	
25.12	OCH ₂ COOC ₅ H ₁₁ (n)	
25.13	OCH₂COO-benzyl	
25.14	OCH(CH ₃)COObenzyl	
25.15	SC ₃ H ₇ (iso)	
25.16	SCH₂COOCH₃	
25.17	SCH ₂ COOC ₂ H ₅	
25.18	SCH(CH₃)COObenzyl	
25.19	SCH₂COObenzyl	<u>-</u>
25.20	COOCH ₃	
25.21	COOC ₃ H ₇ (iso)	

Comp.No.	R ₆	M.p.
25.22	COOC(CH ₃) ₂ COOH	
25.23	COOC(CH ₃) ₂ COO-allyl	
25.24	COOC(CH ₃) ₂ COOCH ₃	
25.25	COOC(CH ₃) ₂ COOethyl	
25.26	COOC(CH ₃) ₂ CONH-allyl	
25.27	CH₂CHCICOOethyl	
25.28	CH ₂ CH=CH ₂	
25.29	CH₂CH₂CH₃	
25.30	CH₂CH₂CF₃	· · ·
25.31	OCH(CH ₃)COOC ₂ H ₅ (R)	· · · · · · · · · · · · · · · · · · ·
25.32	OCH(CH ₃)COOC₂H ₅ (S)	
25.33	OCH(CH₃)COOC₂H₅(R,S)	
25.34	CH₂CHCICOOH	
25.35	CH₂CHCICOOCH₃	
25.36	CH ₂ CHClCOOC ₃ H ₇ (iso)	
25.37	CH₂CHCICONHallyI	
25.38	CH₂C(CH₃)CICOOH	
25.39	CH₂C(CH₃)CICOOCH₃	
25.40	CH₂C(CH₃)CICOOEt	
25.41	CH₂C(CH₃)CICONHEt	
25.42	CH₂CH₂COOH	
25.43	CH ₂ CH ₂ COOCH ₃	
25.44	CH ₂ CH ₂ COOEt	
25.45	CHCICHCICOOH	
25.46	CHCICHCICOOCH ₃	
25.47	CHCICHCICOOEt	
25.48	CH₂CH(OCH₃)COOH	
25.49	CH₂CH(OCH₃)COOCH₃	





Comp.No.	R ₆	M.p.
25.50	CH ₂ CH(OCH ₃)COOEt	
25.51	CH₂CH(SCH₃)COOH	
25.52	CH₂CH(SCH₃)COOCH₃	
25.53	CH₂CH(SCH₃)COOEt	
25.54	CH=CHCOOH	
25.55	CH=CHCOOCH ₃	
25.56	CH=CHCOOEt	
25.57	CH=CCICOOH	
25.58	CH=CCICOOCH ₃	
25.59	COOEt	
25.60	CONH ₂	
25.61	-C(O)OCH ₂	
25.62	CONHSO₂CH ₃	
25.63	COOCH₂COOH	
25.64	COOCH₂COOCH₃	
25.65	COOCH(CH3)COOH	
25.66	COOCH(CH ₃)COOCH ₃	· ·
25.67	COOCH(CH₃)CH₂COOH	
25.68	COOCH(CH₃)CH₂COOCH₃	
25.69	COOC(CH ₃)₂CN	
25.70	COOCH₂CH₂OCH₃	
25.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
25.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
25.73	COOC(CH ₃)₂COOCH₂PHENYL	
25.74	COOCH₂C≡CH	
25.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	

Comp.No.	R ₆	M.p.
25.76	COOCH(CH₃)C≡CH	
25.77	COOC(CH ₃) ₂ COCH ₃	
25.78	NHaliy!	
25.79	N(COCH₃)allyl	
25.80	N(Et)SO ₂ CH ₃	
25.81	N(allyl)SO ₂ CH ₃	
25.82	N(allyI)SO₂Et	
25.83	SO ₂ N(CH ₃) ₂	
25.84	SO ₂ NH ₂	
25.85	SO₂NHCOCH₃	
25.86	OH ·	
25.87	OEt	
25.88	Oallyl	
25.89	OCH₂C≡CCH ₃	
25.90	OCH(CH ₃)CH=CH ₂	
25.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
25.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	<u>-</u>
25.93	OCH ₂	
25.94	OCH ₂ CH ₂ NHCH ₃	
25.95	OCH₂CH₂N(CH₃)COCH₃	
25.96	OCH₂CH₂COOH	
25.97	OC(CH ₃)₂COOH	
25.98	OC(CH ₃) ₂ COOCH ₃	
25.99	OC(CH ₃)₂COOEt	
25.100	OCH₂COOH	
25.101	OSO₂CH₃	
25.102	OSO ₂ CF ₃	





Comp.No.	R ₆	M.p.
25.103	CH₂CHCICOOC₂H₅	
25.104	CH ₂ CHCICON(C ₂ H ₅) ₂	
25.105	CH₂CHCICONHOH	
25.106	CH₂CHCICOOCH₂C6H5	
25.107	CH₂CH(CH₃)COOH	
25.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
25.109	-COOCH ₂	
25.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
25.111	-cooc(ch ₃) ₂ соосн ₂ —	
25.112	COOC(CH ₃)₂CONHCH₂C≡CH	
25.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	-
25.114	OCH ₂ —	

Table 26: Compounds of formula Izz

$$NC \xrightarrow{CI} CH_3 SOCH_3$$

$$N-N CH_3$$

$$CH_3$$

Comp.No.	R ₆	M.p.
26.1	Н	
26.2	CN	
26.3	OCH ₃	
26.4	NHSO ₂ CH ₃	
26.5	OC ₃ H ₇ (iso)	
26.6	O-propargyl	
26.7	OCH(CH₃)C≡CH	
26.8	OCH₂COOCH₂CH₃	
26.9	OCH ₂ CH ₂ OCH ₃	
26.10	OCH ₂ CH ₂ SCH ₂ CH ₃	
26.11	OCH₂COOCH₃	
26.12	OCH ₂ COOC ₅ H ₁₁ (n)	
26.13	OCH₂COO-benzyl	
26.14	OCH(CH₃)COObenzyl	
26.15	SC ₃ H ₇ (iso)	
26.16	SCH₂COOCH₃	
26.17	SCH ₂ COOC ₂ H ₅	
26.18	SCH(CH ₃)COObenzyl	
26.19	SCH₂COObenzyl	
26.20	COOCH ₃	
26.21	COOC ₃ H ₇ (iso)	

Comp.No.	R ₆	M.p.
26.22	COOC(CH₃)₂COOH	
26.23	COOC(CH ₃)₂COO-allyl	
26.24	COOC(CH ₃)₂COOCH ₃	
26.25	COOC(CH ₃)₂COOethyl	
26.26	COOC(CH ₃) ₂ CONH-allyl	
26.27	CH₂CHClCOOethyl	
26.28	CH ₂ CH=CH ₂	
26.29	CH₂CH₂CH₃	
26.30	CH ₂ CH ₂ CF ₃	
26.31	OCH(CH ₃)COOC₂H ₅ (R)	
26.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
26.33	OCH(CH₃)COOC₂H₅(R,S)	
26.34	CH₂CHCICOOH	
26.35	CH₂CHCICOOCH₃	
26.36	CH ₂ CHCICOOC ₃ H ₇ (iso)	
26.37	CH ₂ CHCICONHallyl	
26.38	CH₂C(CH₃)CICOOH	
26.39	CH ₂ C(CH ₃)CICOOCH ₃	
26.40	CH ₂ C(CH ₃)CICOOEt	
26.41	CH₂C(CH₃)CICONHEt	
26.42	CH₂CH₂COOH	
26.43	CH ₂ CH ₂ COOCH ₃	
26.44	CH ₂ CH ₂ COOEt	
26.45	СНСІСНСІСООН	
26.46	CHCICHCICOOCH3	
26.47	CHCICHCICOOEt	
26.48	CH₂CH(OCH₃)COOH	
26.49	CH ₂ CH(OCH ₃)COOCH ₃	

Comp.No.	R ₆	M.p.
26.50	CH₂CH(OCH₃)COOEt	
26.51	CH₂CH(SCH₃)COOH	
26.52	CH₂CH(SCH₃)COOCH₃	
26.53	CH₂CH(SCH₃)COOEt	
26.54	СН=СНСООН	
26.55	CH=CHCOOCH ₃	
26.56	CH=CHCOOEt	
26.57	CH=CCICOOH	
26.58	CH=CCICOOCH ₃	
26.59	COOEt	
26.60	CONH ₂	
26.61	-C(O)OCH ₂	
26.62	CONHSO₂CH ₃	
26.63	COOCH₂COOH	
26.64	COOCH₂COOCH₃	
26.65	COOCH(CH ₃)COOH	
26.66	COOCH(CH ₃)COOCH ₃	
26.67	COOCH(CH₃)CH₂COOH	
26.68	COOCH(CH₃)CH₂COOCH₃	· .
26.69	COOC(CH ₃) ₂ CN	
26.70	COOCH₂CH₂OCH₃	
26.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
26.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
26.73	COOC(CH ₃)₂COOCH₂PHENYL	
26.74	COOCH ₂ C≡CH	
6.75	COOC(CH ₃) ₂ COOCH ₂ C=CH	





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Comp.No.	R ₆	M.p.
26.76	COOCH(CH ₃)C≡CH	
26.77	COOC(CH ₃) ₂ COCH ₃	
26.78	NHallyl	
26.79	N(COCH₃)allyl	
26.80	N(Et)SO ₂ CH ₃	
26.81	N(allyl)SO ₂ CH ₃	
26.82	N(allyl)SO₂Et	
26.83	SO ₂ N(CH ₃) ₂	
26.84	SO ₂ NH ₂	
26.85	SO₂NHCOCH₃	
26.86	ОН	
26.87	OEt	
26.88	Oallyl	
26.89	OCH ₂ C≡CCH ₃	
26.90	OCH(CH ₃)CH=CH ₂	
26.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
26.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
26.93	OCH ₂	
26.94	OCH₂CH₂NHCH₃	
26.95	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
26.96	OCH₂CH₂COOH	
6.97	OC(CH ₃)₂COOH	
6.98	OC(CH ₃) ₂ COOCH ₃	
6.99	OC(CH ₃)₂COOEt	
6.100	OCH₂COOH	
6.101	OSO ₂ CH ₃	
6.102	OSO₂CF ₃	

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Comp.No.	R ₆	·
		M.p.
26.103	CH2CHCICOOC2H5	
26.104	CH₂CHCICON(C₂H₅)₂	
26.105	CH₂CHCICONHOH	
26.106	CH ₂ CHCICOOCH ₂ C ₆ H ₅	
26.107	CH₂CH(CH₃)COOH	
26.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
26.109	-COOCH ₂ —	
26.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	·
26.111	-cooc(ch ₃) ₂ соосн ₂ —	
26.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
26.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
26.114	OCH ₂ —	

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Table 27: Compounds of formula lyy

$$NC \xrightarrow{CI} CH_3 SO_2CH_3$$
 $N-N CH_3$
(lyy)

Comp.No.	R_6	M.p.
27.1	Н	
27.2	CN	
27.3	OCH₃	
27.4	NHSO₂CH ₃	
27.5	OC ₃ H ₇ (iso)	
27.6	O-propargyl	
27.7	OCH(CH₃)C≡CH	
27.8	OCH ₂ COOCH ₂ CH ₃	
27.9	OCH₂CH₂OCH₃	
27.10	OCH ₂ CH ₂ SCH ₂ CH ₃	
27.11	OCH₂COOCH₃	
27.12	OCH₂COOC₅H ₁₁ (n)	
27.13	OCH ₂ COO-benzyl	
27.14	OCH(CH₃)COObenzyl	
27.15	SC ₃ H ₇ (iso)	
27.16	SCH₂COOCH₃	
27.17	SCH ₂ COOC ₂ H ₅	
27.18	SCH(CH ₃)COObenzyl	
27.19	SCH₂COObenzyl	
27.20	COOCH ₃	
27.21	COOC ₃ H ₇ (iso)	

Comp.No.	R ₆	M.p.
27.22	COOC(CH₃)₂COOH	
27.23	COOC(CH ₃)₂COO-allyl	
27.24	COOC(CH ₃) ₂ COOCH ₃	
27.25	COOC(CH ₃)₂COOethyl	
27.26	COOC(CH ₃) ₂ CONH-allyl	
27.27	CH₂CHClCOOethyl	
27.28	CH ₂ CH=CH ₂	
27.29	CH ₂ CH ₂ CH ₃	
27.30	CH ₂ CH ₂ CF ₃	
27.31	OCH(CH ₃)COOC ₂ H ₅ (R)	
27.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
27.33	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
27.34	CH₂CHCICOOH	
27.35	CH₂CHCICOOCH₃	
27.36	CH₂CHCICOOC₃H₂(iso)	
27.37	CH₂CHCICONHallyl	
27.38	CH₂C(CH₃)CICOOH	
27.39	CH₂C(CH₃)CICOOCH₃	
27.40	CH₂C(CH₃)CICOOEt	
27.41	CH₂C(CH₃)CICONHEt	
27.42	CH ₂ CH ₂ COOH	
27.43	CH ₂ CH ₂ COOCH ₃	
27.44	CH ₂ CH ₂ COOEt	
27.45	СНСІСНСІСООН	
27.46	CHCICHCICOOCH3	
27.47	CHCICHCICOOEt	
27.48	CH₂CH(OCH₃)COOH	
27.49	CH ₂ CH(OCH ₃)COOCH ₃	





Comp.No.	R ₆	M.p.
27.50	CH₂CH(OCH₃)COOEt	· ·
27.51	CH₂CH(SCH₃)COOH	
27.52	CH₂CH(SCH₃)COOCH₃	
27.53	CH₂CH(SCH₃)COOEt	
27.54	CH=CHCOOH	
27.55	CH=CHCOOCH ₃	
27.56	CH=CHCOOEt	
27.57	CH=CCICOOH	
27.58	CH=CCICOOCH ₃	
27.59	COOEt	
27.60	CONH ₂	
27.61	-C(O)OCH ₂	
27.62	CONHSO₂CH₃	
27.63	COOCH₂COOH	
27.64	COOCH₂COOCH₃	
27.65	COOCH(CH3)COOH	
27.66	COOCH(CH ₃)COOCH ₃	
27.67	COOCH(CH₃)CH₂COOH	
27.68	COOCH(CH ₃)CH ₂ COOCH ₃	
27.69	COOC(CH ₃)₂CN	
27.70	COOCH₂CH₂OCH₃	
27.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
27.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
27.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
27.74	COOCH₂C≡CH	
27.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	

Comp.No.	R ₆	M.p.
27.76	COOCH(CH₃)C≡CH	
27.77	COOC(CH ₃) ₂ COCH ₃	
27.78	NHallyl	
27.79	N(COCH ₃)allyl	
27.80	N(Et)SO ₂ CH ₃	
27.81	N(allyl)SO ₂ CH ₃	
27.82	N(allyl)SO₂Et	
27.83	SO ₂ N(CH ₃) ₂	
27.84	SO₂NH₂	
27.85	SO₂NHCOCH₃	
27.86	ОН	
27.87	OEt	
27.88	Oallyl	
27.89	OCH ₂ C=CCH ₃	
27.90	OCH(CH ₃)CH=CH ₂	
27.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
27.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
27.93	OCH ₂	
27.94	OCH ₂ CH ₂ NHCH ₃	
27.95	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
27.96	OCH ₂ CH ₂ COOH	
27.97	OC(CH ₃) ₂ COOH	
27.98	OC(CH ₃) ₂ COOCH ₃	
27.99	OC(CH ₃)₂COOEt	
27.100 ⁻	OCH₂COOH	
27.101	OSO₂CH₃	
27.102	OSO₂CF₃	







Comp.No.	R ₆	M.p.
27.103	CH₂CHCICOOC₂H₅	
27.104	CH₂CHCICON(C₂H₅)₂	
27.105	CH₂CHCICONHOH	
27.106	CH₂CHCICOOCH₂C ₆ H ₅	
27.107	CH₂CH(CH₃)COOH	
27.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
27.109	-COOCH ₂	
27.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
27.111	-COOC(CH ₃) ₂ COOCH ₂ —	
27.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
27.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
27.114	OCH ₂ —	

Table 28: Compounds of formula lww

Comp.No.	R ₆	M.p.
28.1	Н	
28.2	CN	
28.3	OCH₃	
28.4	NHSO₂CH₃	
28.5	OC₃H ₇ (iso)	
28.6	O-propargyl	
28.7	OCH(CH₃)C≡CH	
28.8	OCH ₂ COOCH ₂ CH ₃	
28.9	OCH ₂ CH ₂ OCH ₃	
28.10	OCH ₂ CH ₂ SCH ₂ CH ₃	
28.11	OCH₂COOCH₃	
28.12	OCH ₂ COOC ₅ H ₁₁ (n)	
28.13	OCH₂COO-benzyl	
28.14	OCH(CH ₃)COObenzyl	
28.15	SC ₃ H ₇ (iso)	
28.16	SCH₂COOCH₃	
28.17	SCH₂COOC₂H₅	
28.18	SCH(CH ₃)COObenzyl	
28.19	SCH₂COObenzyl	
28.20	COOCH ₃	
28.21	COOC ₃ H ₇ (iso)	

(8)

l D	
R ₆	M.p.
COOC(CH ₃) ₂ COO-allyl	
COOC(CH ₃) ₂ COOCH ₃	
COOC(CH₃)₂COOethyl	
COOC(CH ₃)₂CONH-allyl	
CH₂CHClCOOethyl	
CH ₂ CH=CH ₂	
CH ₂ CH ₂ CH ₃	
CH₂CH₂CF₃	
OCH(CH ₃)COOC ₂ H ₅ (R)	
OCH(CH ₃)COOC ₂ H ₅ (S)	
OCH(CH₃)COOC₂H₅(R,S)	
CH₂CHCICOOH	
CH₂CHCICOOCH₃	
CH₂CHCICOOC₃H₂(iso)	
CH₂CHCICONHallyl	
CH₂C(CH₃)CICOOH	
CH₂C(CH₃)CICOOCH₃	
CH₂C(CH₃)CICOOEt	
CH₂C(CH₃)CICONHEt	
CH₂CH₂COOH	
CH₂CH₂COOCH₃	
CH₂CH₂COOEt	
СНСІСНСІСООН	
CHCICHCICOOCH3	
CHCICHCICOOEt	
CH₂CH(OCH₃)COOH	
CH₂CH(OCH₃)COOCH₃	
	COOC(CH ₃) ₂ COOH COOC(CH ₃) ₂ COO-allyl COOC(CH ₃) ₂ COOCH ₃ COOC(CH ₃) ₂ COOethyl COOC(CH ₃) ₂ CONH-allyl COOC(CH ₃) ₂ CONH-allyl CH ₂ CHCICOOethyl CH ₂ CH=CH ₂ CH ₂ CH ₂ CH ₃ CH ₂ CH ₂ CF ₃ OCH(CH ₃)COOC ₂ H ₅ (R) OCH(CH ₃)COOC ₂ H ₅ (R) OCH(CH ₃)COOC ₂ H ₅ (R,S) CH ₂ CHCICOOH CH ₂ CHCICOOCH ₃ CH ₂ C(CH ₃)CICOOCH CH ₂ CH ₂ COOCH CHCICHCICOOCH CHCICHCICOOCH CHCICHCICOOCH

Comp.No.	R ₆	M.p.
28.50	CH₂CH(OCH₃)COOEt	<u> </u>
28.51	CH₂CH(SCH₃)COOH	
28.52	CH₂CH(SCH₃)COOCH₃	
28.53	CH₂CH(SCH₃)COOEt	
28.54	CH=CHCOOH	
28.55	CH=CHCOOCH₃	
28.56	CH=CHCOOEt	
28.57	CH=CCICOOH .	
28.58	CH=CCICOOCH ₃	
28.59	COOEt	
28.60	CONH ₂	
28.61	-C(O)OCH ₂	
28.62	CONHSO₂CH ₃	
28.63	COOCH₂COOH	
28.64	COOCH₂COOCH₃	
28.65	COOCH(CH3)COOH	
28.66	COOCH(CH3)COOCH3	
28.67	COOCH(CH₃)CH₂COOH	
28.68	COOCH(CH ₃)CH ₂ COOCH ₃	
28.69	COOC(CH ₃) ₂ CN	
28.70	COOCH ₂ CH ₂ OCH ₃	
28.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
28.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
28.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
28.74	COOCH₂C≡CH	
28.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	





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Comp.No.	R ₆	M.p.
28.76	COOCH(CH₃)C≡CH	
28.77	COOC(CH ₃) ₂ COCH ₃	
28.78	NHallyl	
28.79	N(COCH₃)allyl	
28.80	N(Et)SO ₂ CH ₃	
28.81	N(allyl)SO ₂ CH ₃	
28.82	N(allyl)SO₂Et	
28.83	SO₂N(CH₃)₂	
28.84	SO ₂ NH ₂	
28.85	SO₂NHCOCH ₃	
28.86	ОН	
28.87	OEt	
28.88	Oallyl	· ·
28.89	OCH ₂ C≡CCH ₃	
28.90	OCH(CH ₃)CH=CH ₂	
28.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
28.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
28.93	OCH ₂	
28.94	OCH₂CH₂NHCH₃	
8.95	OCH₂CH₂N(CH₃)COCH₃	
8.96	OCH ₂ CH ₂ COOH	
8.97	OC(CH ₃)₂COOH	
8.98	OC(CH ₃)₂COOCH ₃	
8.99	OC(CH ₃) ₂ COOEt	
8.100	OCH₂COOH	
3.101	OSO ₂ CH ₃	
3.102	OSO ₂ CF ₃	

Comp.No.	R ₆	M.p.
28.103	CH₂CHCICOOC₂H₅	
28.104	CH₂CHCICON(C₂H₅)₂	
28.105	CH₂CHCICONHOH	
28.106	CH₂CHCICOOCH₂C6H5	
28.107	CH₂CH(CH₃)COOH	
28.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
28.109	-соосн ₂ —	
28.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
28.111	-COOC(CH ₃) ₂ COOCH ₂ —	
28.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
28.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
28.114	OCH ₂	0





Table 29: Compounds of formula Ivv

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Comp.No.	R ₆	M.p.
29.1	Н	
29.2	CN	
29.3	OCH ₃	
29.4	NHSO₂CH ₃	
29.5	OC ₃ H ₇ (iso)	
29.6	O-propargyl	
29.7	OCH(CH₃)C≡CH	
29.8	OCH ₂ COOCH ₂ CH ₃	
29.9	OCH₂CH₂OCH₃	
29.10	OCH₂CH₂SCH₂CH₃	
29.11	OCH₂COOCH₃	
29.12	OCH ₂ COOC ₅ H ₁₁ (n)	
29.13	OCH₂COO-benzyl	
29.14	OCH(CH ₃)COObenzyl	
29.15	SC ₃ H ₇ (iso)	
29.16	SCH₂COOCH₃	
29.17	SCH ₂ COOC ₂ H ₅	
29.18	SCH(CH ₃)COObenzyl	
9.19	SCH₂COObenzyl	
9.20	COOCH ₃	
9.21	COOC ₃ H ₇ (iso)	

Comp.No.	R ₆	M.p.
29.22	COOC(CH ₃) ₂ COOH	
29.23	COOC(CH ₃) ₂ COO-allyl	
29.24	COOC(CH ₃) ₂ COOCH ₃	
29.25	COOC(CH ₃) ₂ COOethyl	
29.26	COOC(CH ₃) ₂ CONH-allyl	
29.27	CH₂CHCICOOethyl	
29.28	CH₂CH=CH₂	
29.29	CH₂CH₂CH₃	
29.30	CH ₂ CH ₂ CF ₃	
29.31	OCH(CH ₃)COOC ₂ H ₅ (R)	
29.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
29.33	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
29.34	CH₂CHCICOOH	
29.35	CH₂CHCICOOCH₃	
29.36	CH ₂ CHClCOOC ₃ H ₇ (iso)	
29.37	CH₂CHCICONHallyl	
29.38	CH ₂ C(CH ₃)CICOOH	·
29.39	CH₂C(CH₃)CICOOCH₃	
29.40	CH₂C(CH₃)CICOOEt	
29.41	CH₂C(CH₃)CICONHEt	
29.42	CH₂CH₂COOH	
29.43	CH ₂ CH ₂ COOCH ₃	
29.44	CH₂CH₂COOEt	
29.45	СНСІСНСІСООН	
29.46	CHCICHCICOOCH ₃	
9.47	CHCICHCICOOEt	
9.48	CH₂CH(OCH₃)COOH	
9.49	CH₂CH(OCH₃)COOCH₃	





Comp.No.	R ₆	M.p.
29.50	CH₂CH(OCH₃)COOEt	
29.51	CH₂CH(SCH₃)COOH	
29.52	CH₂CH(SCH₃)COOCH₃	
29.53	CH₂CH(SCH₃)COOEt	
29.54	CH=CHCOOH	
29.55	CH=CHCOOCH₃	
29.56	CH=CHCOOEt	
29.57	CH=CCICOOH	
29.58	CH=CCICOOCH₃	
29.59	COOEt	
29.60	CONH ₂	
29.61	-C(O)OCH ₂	
29.62	CONHSO₂CH₃	
29.63	COOCH₂COOH	
29.64	COOCH₂COOCH₃	
29.65	СООСН(СН3)СООН	
29.66	COOCH(CH ₃)COOCH ₃	
29.67	COOCH(CH ₃)CH ₂ COOH	
29.68	COOCH(CH ₃)CH ₂ COOCH ₃	· ·
29.69	COOC(CH ₃)₂CN	
29.70	COOCH ₂ CH ₂ OCH ₃	
29.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
29.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
9.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
9.74	COOCH₂C≡CH	
9.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	

Comp.No.	R ₆	M.p.
29.76	COOCH(CH₃)C≡CH	
29.77	COOC(CH ₃) ₂ COCH ₃	
29.78	NHaliyi	
29.79	N(COCH ₃)allyl	
29.80	N(Et)SO ₂ CH ₃	
29.81	N(allyl)SO ₂ CH ₃	
29.82	N(allyl)SO ₂ Et	
29.83	SO ₂ N(CH ₃) ₂	
29.84	SO ₂ NH ₂	-
29.85	SO₂NHCOCH₃	
29.86	ОН	
29.87	OEt	
29.88	Oallyl	
29.89	OCH₂C≡CCH₃	
29.90	OCH(CH ₃)CH=CH ₂	
29.91	OCH ₂ CH ₂ OCH ₂ CH ₃	
29.92	OCH ₂ CH ₂ OCH ₂ CH ₂ OCH ₃	
29.93	OCH ₂	
29.94	OCH₂CH₂NHCH₃	
29.95	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
29.96	OCH ₂ CH ₂ COOH	
29.97	OC(CH₃)₂COOH	
29.98	OC(CH ₃) ₂ COOCH ₃	
29.99	OC(CH ₃) ₂ COOEt	
29.100	OCH ₂ COOH	
29.101	OSO ₂ CH ₃	
29.102	OSO ₂ CF ₃	





Comp.No.	R ₆	M.p.
29.103	CH₂CHCICOOC₂H₅	
29.104	CH ₂ CHCICON(C ₂ H ₅) ₂	
29.105	CH ₂ CHCICONHOH	
29.106	CH₂CHCICOOCH₂C₅H₅	
29.107	CH₂CH(CH₃)COOH	
29.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
29.109	-COOCH ₂	
29.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
29.111	-cooc(ch ₃) ₂ соосн ₂ —	
29.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
29.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
29.114	OCH ₂ —	

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Table 30: Compounds of formula luu

Comp.No.	R ₆	M.p.
30.1	Н	
30.2	CN	· · · · · · · · · · · · · · · · · · ·
30.3	OCH ₃	
30.4	NHSO₂CH₃	
30.5	OC ₃ H ₇ (iso)	
30.6	O-propargyl	
30.7	OCH(CH ₃)C≡CH	
30.8	OCH₂COOCH₂CH₃	
30.9	OCH₂CH₂OCH₃	
30.10	OCH₂CH₂SCH₂CH₃	
30.11	OCH₂COOCH₃	
30.12	OCH ₂ COOC ₅ H ₁₁ (n)	
30.13	OCH₂COO-benzyl	
30.14	OCH(CH ₃)COObenzyl	
30.15	SC ₃ H ₇ (iso)	
30.16	SCH₂COOCH₃	
30.17	SCH₂COOC₂H₅	
30.18	SCH(CH ₃)COObenzyl	
30.19	SCH₂COObenzyl	
30.20	COOCH ₃	
0.21	COOC ₃ H ₇ (iso)	

30.22	COOC(CH ₃)₂COOH	
30.23	COOC(CH ₃)₂COO-allyl	
30.24	COOC(CH ₃)₂COOCH ₃	
30.25	COOC(CH ₃)₂COOethyl	
30.26	COOC(CH ₃) ₂ CONH-allyl	
30.27	CH₂CHCICOOethyl	
30.28	CH ₂ CH=CH ₂	
30.29	CH₂CH₂CH₃	
30.30	CH ₂ CH ₂ CF ₃	
30.31	OCH(CH ₃)COOC ₂ H ₅ (R)	
30.32	OCH(CH ₃)COOC ₂ H ₅ (S)	
30.33	OCH(CH ₃)COOC ₂ H ₅ (R,S)	
30.34	CH₂CHCICOOH	
30.35	CH₂CHCICOOCH₃	
30.36	CH₂CHCICOOC₃H₂(iso)	
30.37	CH₂CHCICONHallyI	
30.38	CH₂C(CH₃)CICOOH	
30.39	CH₂C(CH₃)CICOOCH₃	
30.40	CH₂C(CH₃)CICOOEt	
30.41	CH₂C(CH₃)CICONHEt	· ·
30.42	CH₂CH₂COOH	
30.43	CH ₂ CH ₂ COOCH ₃	
30.44	CH ₂ CH ₂ COOEt	
30.45	СНСІСНСІСООН	
30.46	CHCICHCICOOCH3	
30.47	CHCICHCICOOEt	
30.48	CH₂CH(OCH₃)COOH	
30.49	CH₂CH(OCH₃)COOCH₃	
30.50	CH₂CH(OCH₃)COOEt	

20.51	lou ou	<u>. </u>
30.51	CH₂CH(SCH₃)COOH	
30.52	CH₂CH(SCH₃)COOCH₃	
30.53	CH₂CH(SCH₃)COOEt	
30.54	CH=CHCOOH	
30.55	CH=CHCOOCH₃	
30.56	CH=CHCOOEt	
30.57	CH=CCICOOH	
30.58	CH=CCICOOCH ₃	
30.59	COOEt	
30.60	CONH ₂	
30.61	0	
	-C(O)OCH ₂	
30.62	CONHSO₂CH₃	
30.63		
30.64	COOCH₂COOH	
30.65	COOCH₂COOCH₃	
	COOCH(CH3)COOH	
30.66	COOCH(CH3)COOCH3	
30.67	COOCH(CH₃)CH₂COOH	
30.68	COOCH(CH ₃)CH ₂ COOCH ₃	
30.69	COOC(CH ₃)₂CN	
30.70	COOCH₂CH₂OCH₃	
30.71	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OCH ₃	
30.72	COOC(CH ₃) ₂ -C(O)O-CH ₂	
30.73	COOC(CH ₃) ₂ COOCH ₂ PHENYL	
30.74	COOCH₂C≡CH	
30.75	COOC(CH ₃) ₂ COOCH ₂ C≡CH	
30.76		
30.77		
	COOCH(CH ₃)C≡CH COOC(CH ₃) ₂ COCH ₃	

30.78	NHalla	<u></u>
	NHallyl	
30.79	N(COCH₃)allyl	
30.80	N(Et)SO ₂ CH ₃	
30.81	N(aliyi)SO₂CH₃	
30.82	N(allyl)SO₂Et	
30.83	SO ₂ N(CH ₃) ₂	
30.84	SO ₂ NH ₂	
30.85	SO₂NHCOCH ₃	
30.86	ОН	
30.87	OEt	
30.88	Oaliyi	
30.89	OCH₂C≡CCH₃	
30.90	OCH(CH ₃)CH=CH ₂	
30.91	OCH₂CH₂OCH₂CH₃	
30.92	OCH₂CH₂OCH₂CH₃OCH₃	
30.93	OCH ₂	
30.94	OCH₂CH₂NHCH₃	
30.95	OCH ₂ CH ₂ N(CH ₃)COCH ₃	
30.96	OCH ₂ CH ₂ COOH	
30.97	OC(CH ₃) ₂ COOH	
30.98	OC(CH ₃) ₂ COOCH ₃	
30.99	OC(CH ₃) ₂ COOEt	_
30.100	OCH₂COOH	
30.101	OSO₂CH₃	
30.102	OSO ₂ CF ₃	· ·
30.103	CH₂CHCICOOC₂H₅	-
30.104	CH ₂ CHCICON(C ₂ H ₅) ₂	
30.105	CH₂CHCICONHOH	-

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30.106	CH₂CHCICOOCH₂C6H5	
30.107	CH₂CH(CH₃)COOH	
30.108	CH ₂ CH(CH ₃)COOC ₂ H ₅	
30.109	-COOСН ₂ —	
30.110	COOC(CH ₃) ₂ COOCH ₂ CH ₂ OC ₂ H ₅	
30.111	-COOC(CH ₃) ₂ COOCH ₂	
30.112	COOC(CH ₃) ₂ CONHCH ₂ C≡CH	
30.113	COOC(CH ₃) ₂ CON(CH ₂ CH ₃) ₂	
30.114	OCH ₂ —	



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Formulation Examples for active ingredients of formula I (throughout, percentages are by weight)

F1. Emulsifiable concentrates	a)	b)	c)	d)
a compound of Tables 1-30 calcium dodecylbenzenesulfonate castor oil polyglycol ether	5 % 6 %	10 ዩ 8 ዩ	25 % 6 %	50 % 8 %
(36 mol of ethylene oxide) octylphenol polyglycol ether	4 8	- 4 %	4 %	4 %
(7-8 mol of ethylene oxide) cyclohexanone	_	-	10 %	2 % 20 %
aromatic hydrocarbon mixture C ₉ -C ₁₂	85 %	78 %	55 ¥	16 %

Emulsions of any desired concentration can be obtained from such concentrates by dilution with water.

F2. Solutions	a)	b)	c)	d)
a compound of Tables 1-30 1-methoxy-3-(3-methoxy-	5 %	10 %	50 %	90 %
propoxy)-propane	-	20 %	20 %	_
polyethylene glycol (mol.wt.400)	20 %	10 %	_	_
N-methyl-2-pyrrolidone	-	-	30 %	10 %
aromatic hydrocarbon mixture	75 %	60 %	_	_
C_9-C_{12}				

The solutions are suitable for application in the form of micro-drops.

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F3. Wettable powders	a)		b)		c)		d)	
a compound of Tables 1-30	5	용	25	ક્ર	50	જ	80	9
sodium lignosulfonate	4	ቄ	_		3	ક	_	
sodium laurylsulfate	2	윰	3	용	_		4	ક
sodium diisobutylnaphthalene-							-	•
sulfonate	_		6	ક	5	용	6	용
octylphenol polyglycol ether	-		1	ક્ર	2	ક	_	
(7-8 mol of ethylene oxide)								
highly dispersed silicic acid	1	ક	3	용	5	윧	10	용
kaolin	88	윰	62	ક	35	ક	_	

The active ingredient is thoroughly mixed with the adjuvants and the mixture is thoroughly ground in a suitable mill, affording wettable powders which can be diluted with water to give suspensions of any desired concentration.

F4. Coated granules	a)		b)		c)		
a compound of Tables 1-30	0.1	ક	5	8	15	ક	
highly dispersed silicic acid	0.9	ક	2	ક	2	ક	
inorganic carrier	99.0	용	93	용	83	- 2 k	•
(diameter 0.1 - 1 mm)						Ū	
e.g. CaCO ₃ or SiO ₂							

The active ingredient is dissolved in methylene chloride, the solution is sprayed onto the carrier, and the solvent is subsequently evaporated off in vacuo.

F5. Coated granules	a)		b)		C)	
a compound of Tables 1-30	0.1	9	5	용	15	용
polyethylene glycol (mol.wt. 200)	1.0	કૃ	2	ઝ	3	윶
highly dispersed silicic acid	0.9	&	1	ક	2	용
inorganic carrier	98.0	9ઠ	92	ક	80	
(diameter 0.1 - 1 mm)						Ů
e.g. CaCO ₃ or SiO ₂						

The finely ground active ingredient is uniformly applied, in a mixer, to the carrier moistened with polyethylene glycol. Non-dusty coated granules are obtained in this manner.

F6. Extruder granules	a)		b)		c)		d)	
a compound of Tables 1-30	0.1	용	3	용	5	g.	15	કૃ
sodium lignosulfonate	1.5	ક	2	ક	3	윰	4	윧
carboxymethylcellulose	1.4	ક	2	ક	2	ቄ	2	કૃ
kaolin	97.0	ዔ	93	ዓ	90	9-	70	Q.

The active ingredient is mixed and ground with the adjuvants and the mixture is moistened with water. The mixture is extruded and then dried in a stream of air.

F7. Dusts	a)		b)		c)	
a compound of Tables 1-30	0.1	ક	1	용	5	ક
talcum	39.9	ક	49	용	35	융
kaolin	60.0	ક	50	ક	60	용

Ready-to-use dusts are obtained by mixing the active ingredient with the carriers and grinding the mixture in a suitable mill.

F8. Suspension concentrates	a)	b)	c)	đ)
a compound of Tables 1-30	3 %	10 %	25 %	50 %
ethylene glycol	5 %	5 %	5 %	5 %
nonylphenol polyglycol ether (15 mol of ethylene oxide)	-	1 %	2 %	-
sodium lignosulfonate	3 %	3 %	4 %	5 %
carboxymethylcellulose	1 %	1 %	1 %	1 %
37% aqueous formaldehyde solution	0.2 %	0.2 %	_	
silicone oil emulsion	0.8 %	0.8 %	0.8 %	0.8 %
water	87 %	79 %	62 %	38 %

The finely ground active ingredient is intimately mixed with the adjuvants, giving a suspension concentrate from which suspensions of any desired concentration can be obtained by dilution with water.

Biological Examples

Example B1: Pre-emergence herbicidal action

Monocotyledonous and dicotyledonous test plants are sown in standard soil in plastic pots. Immediately after sowing, an aqueous suspension or emulsion of the test compounds prepared from a 25 % wettable powder or emulsifiable concentrate (Example F3, b) or F1, c)) is applied by spraying at a rate of application corresponding to 2000 g of active ingredient/hectare (500 l water/ha). The test plants are then cultivated in a greenhouse under optimum conditions. After 3 weeks the test is evaluated in accordance with a scale of nine ratings (1 = total damage, 9 = no action). Ratings of from 1 to 4 (especially from 1 to 3) indicate good to very good herbicidal action.

Test plants: Avena, Setaria, Sinapis, Stellaria

The compounds according to the invention exhibit good herbicidal action.

Examples of the good herbicidal action are given in Table B1.

Table B1: Pre-emergence action:

Test plant: Compound No.	Avena	Setaria	Sinapis	Stellaria
1.001	1	1	1	
1.007	1	1	2	I
1.010	2	1	2	1
2.007	2	1		1
2.035	2	1 .	2	1
2.037	2	1	2	1
3.001	1	1	1	1
•	•	1	i	1 .

Test plant: Compound No.	Avena	Setaria	Sinapis	Stellaria	
3.010	3	ī	1		
3.011	2	1	1	I	
3.035	2	2	1	1	
7.001	5	1	1	1	
8.001	5	1	1	3	
8.009	3	1	4	3	
9.007	3	1	2	2	
	<i>-</i>	1	2	1	

The same results are obtained when compounds of formula I are formulated in accordance with Examples F2 and F4 to F8.

Example B2: Post-emergence herbicidal action (contact herbicide)

In a greenhouse, monocotyledonous and dicotyledonous test plants are raised in plastic pots containing standard soil and at the 4- to 6-leaf stage are sprayed with an aqueous suspension or emulsion of the test compounds of formula I prepared from a 25 % wettable powder or emulsifiable concentrate (Example F3, b) or F1, c)) at a rate of application corresponding to 2000 g of active ingredient/ha (500 l water/ha). The test plants are then grown on in the greenhouse under optimum conditions. After about 18 days the test is evaluated in accordance with a scale of nine ratings (1 = total damage, 9 = no action). Ratings of from 1 to 4 (especially from 1 to 3) indicate good to very good herbicidal action.

In this test too, the compounds of formula I according to the invention exhibit good herbicidal action.

Examples of the good herbicidal activity of the compounds of formula I are given in Table B2.

Table B2: Post-emergence action:

Test plant: Compound No.	Avena	Setaria	Sinapis	Stellaria	
1.001	1	1	1	1	
1.007	1	1	1	1	
1.010	2	1	1	1	
1.042	1	1	1	1	
1.044	5	2	1	1	
2.001	1	1	1	1	
2.007	2	1	1	1	
2.037	1	1	1	1	
2.038	1	1	1	1	
2.039	1	1	1	1	
2.041	5	2	1	1	
3.001	1	1	1	1	
3.010	2	1	1	1	
3.011	1	1	1	1	
3.035	1	· 1	1	1	
3.037	1	1	1	1 .	
3.039	1	1	1	1	
3.041	6	2	1	1	
7.001	6	2	2	2	
7.034	2	1	1	1	
7.040	6	2	1	1	
8.001	5	1	1	1	
8.009	2	1	1	1	
8.035	1	1	1 *	1	
8.073	1	1	1	1	
9.007	2	1	1	3	
9.041	6	3	1	1	
9.073	2	2	1	1	

The same results are obtained when compounds of formula I are formulated in accordance with Examples F2 and F4 to F8.

What is claimed is:

1. A compound of formula I

$$R_{5} \xrightarrow{R_{4}} R_{1} \xrightarrow{S(0)} R_{2}$$

$$R_{6} \xrightarrow{R_{1}} S(0) R_{2}$$

$$R_{3} \xrightarrow{R_{1}} (I)$$

wherein

 R_1 is C_1 - C_4 alkyl;

 R_2 is C_1 - C_4 alkyl, C_1 - C_4 haloalkyl, C_3 - or C_4 -alkenyl, C_3 - or C_4 -haloalkenyl or C_3 - or C_4 -alkynyl;

n is 0, 1 or 2;

R₃ is hydrogen, C₁-C₄alkyl, C₁-C₄haloalkyl, C₃- or C₄-alkenyl, C₃- or C₄-haloalkenyl, C₃- or C₄-alkynyl, -CH₂COOH, -CH₂COO-C₁-C₄alkyl or -CH₂CN;

R₄ is hydrogen, fluorine, chlorine or bromine;

R₅ is hydrogen, halogen, methyl, trifluoromethyl, cyano, nitro, amino or C₁-C₄halo-alkoxy;

(R₆) is hydrogen, halogen, cyano, NHR₁₀, NR₁₀R₁₁ or SO₂Cl;

R₁₀ and R₁₁ are each independently of the other C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₃-C₆cycloalkyl, C₁-C₈haloalkyl, C₃-C₈haloalkenyl, C₁-C₄alkylcarbonyl, C₁-C₄haloalkylcarbonyl, C₁-C₄haloalkylsulfonyl, benzoyl, benzoyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen; or

 (R_6) is OR_{20} ;

R₂₀ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl,

 $C_1-C_4 alkoxy-C_1-C_4 alkyl, \ C_1-C_4 alkylamino-C_1-C_4 alkyl, \ di-C_1-C_4 alkyl-amino-C_1-C_4 alkyl, \ C_1-C_4 alkoxy-C_1-C_4 alkyl, \ C_1-C_4 alkyl-thio-C_1-C_4 alkyl, \ phenyl, benzyl, pyridyl, pyrimidinyl, pyrazinyl, pyridazinyl, those$

aromatic and heteroaromatic rings being unsubstituted or mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; or

 R_{20} is C_1 - C_8 alkyl- $COXR_{21}$ or $CH(C_6H_5)COXR_{21}$;

X is oxygen, sulfur or NR₂₂;

R₂₁ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl monoto tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, or benzyl monoto tri-substituted by C₁-C₄alkyl or by halogen; and

R₂₂ is hydrogen, C₁-C₈alkyl or C₃-C₈alkenyl; or

 (R_6) is $S(O)_m R_{30}$;

m is 0, 1 or 2;

R₃₀ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, or C₁-C₄alkyl-COVR₃₁;

V is oxygen, sulfur or NR₃₂;

R₃₁ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl monoto tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl monoto tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen; and

R₃₂ is hydrogen, C₁-C₈alkyl or C₃-C₈alkenyl; or

 R_6 COR₄₀;

is hydrogen, chlorine, C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_3 - C_8 haloalkenyl, C_3 - C_6 cycloalkyl, C_1 - C_4 alkoxy- C_1 - C_4 alkyl, C_1 - C_4 alkyl, phenyl, phenyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; or

 R_6 is $COYR_{50}$;

Y is oxygen, sulfur, NR₅₁ or NOR₅₄;

 R_{50} is hydrogen, C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl,

$$-CH_2-CH-CH_2$$
, C_3-C_8 haloalkenyl, C_3-C_6 cycloalkyl, $-CH_2$,

 C_1 - C_4 alkoxy- C_1 - C_4 alkyl, C_1 - C_4 alkylthio- C_1 - C_4 alkyl, phenyl, phenyl mono- to trisubstituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, benzyl, benzyl mono- to

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tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, C_1 - C_4 alkyl- $COZR_{52}$, C_3 - C_6 cycloalkyl- $COZR_{52}$, C_1 - C_4 alkyl-CO- C_1 - C_4 alkyl or C_1 - C_4 cyanoalkyl;

Z is oxygen, sulfur, NR₅₃ or NOR₅₅;

R₅₂ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl,

$$O$$
 CH_2 — CH_2 — CH_2 , C_3 - C_8 haloalkenyl, C_3 - C_6 cycloalkyl, $-CH_2$,

 $\label{eq:continuous_continuous$

 R_{51} and R_{53} are each independently of the other C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_1 - C_4 alkylcarbonyl, C_1 - C_4 haloalkylcarbonyl, C_1 - C_4 alkylsulfonyl, benzoyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen;

 R_{54} and R_{55} are each independently of the other C_1 - C_4 alkyl; or

$$R_6$$
 is $-- SO_2$ N R_{56} ;

R₅₆ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl or C₃-C₈alkynyl;

R₅₇ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl or C₁-C₄alkylcarbonyl; or

 $R_6 \quad \text{is C_1-C_8alkyl-B, C_1-C_8haloalkyl-B, C_2-C_8alkenyl-B, C_2-C_8alkynyl-B, C_2-C_8haloalkyl-B, C_1-C_4alkyl-B or C_1-C_4alkyl-B; and C_1-C_4alkyl-B.}$

B is hydrogen, -COZR₅₂, cyano or C₁-C₄alkyl-C(O)-,

or a salt or stereoisomer of a compound of formula I.

- 2. A compound of formula I according to claim 1, wherein
- R_1 is C_1 - C_4 alkyl;

 R_2 is C_1 - C_4 alkyl, C_1 - C_4 haloalkyl, C_3 - or C_4 -alkenyl or C_3 - or C_4 -alkynyl;

n is 0, 1 or 2;

 R_3 is hydrogen, C_1 - C_4 alkyl, C_1 - C_4 haloalkyl, C_3 - or C_4 -alkenyl or C_3 - or C_4 -alkynyl;

R₄ is hydrogen, fluorine or chlorine;

R₅ is hydrogen, halogen, methyl. reifluoromethyl, cyano, nitro, amino or C₁-C₄haloalkoxy;

R₆ is hydrogen, halogen, C₁-C₅alkyl, C₁-C₅haloalkyl, C₂-C₅alkenyl, C₂-C₅haloalkenyl,

C₂-C₅alkynyl, C₂-C₅haloalkynyl, cyano, NHR₁₀ or NR₁₀R₁₁;

 R_{10} and R_{11} are each independently of the other C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_1 - C_4 alkylcarbonyl, C_1 - C_4 haloalkylsulfonyl, benzoyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; or

 R_6 is OR_{20} ;

R₂₀ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, benzyl, pyridyl, pyrimidinyl, pyrazinyl, pyridazinyl, those aromatic and heteroaromatic rings being unsubstituted or mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, or C₁-C₈alkyl-COXR₂₁;

X is oxygen, sulfur or NR₂₂;

 $R_{21} \quad \text{is hydrogen, C_1-C_8alkyl, C_3-C_8alkynyl, C_1-C_8haloalkyl, C_3-C_6cycloalkyl, C_1-C_4alkoxy-C_1-C_4alkyl, C_1-C_4alkylthio-C_1-C_4alkyl, phenyl, phenyl mono- to tri-substituted by C_1-C_4alkyl, C_1-C_4haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1-C_4alkyl or by halogen; and$

R₂₂ is hydrogen, C₁-C₈alkyl or C₃-C₈alkenyl; or

 R_6 is $S(O)_m R_{30}$;

m is 0, 1 or 2;

 $R_{30} \quad \text{is hydrogen, chlorine, C_1-C_8alkyl, C_3-C_8alkenyl, C_3-$C_8alkynyl, C_1-$C_8haloalkyl, C_3-$C_6cycloalkyl, C_1-$C_4alkyl, C_1-C_4alkyl, C_1-C_4alkyl, C_1-$C_4alkyl, phenyl, phenyl mono- to tri-substituted by C_1-C_4alkyl, C_1-C_4haloalkyl or by halogen, benzyl, benzyl mono- to tri-substituted by C_1-C_4alkyl, C_1-C_4haloalkyl or by halogen, or C_1-C_4alkyl-$COVR_{31}$;}$

V is oxygen, sulfur or NR₃₂;

R₃₁ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen; and

R₃₂ is hydrogen, C₁-C₈alkyl or C₃-C₈alkenyl; or

R₆ is COR₄₀;

 $R_{40} \quad \text{is hydrogen, C_1-C_8alkyl, C_3-C_8alkenyl, C_3-C_8alkynyl, C_1-C_8haloalkyl, C_3-C_6cycloalkyl, C_1-C_4alkoxy-C_1-C_4alkyl, C_1-C_4alkylthio-C_1-C_4alkyl, phenyl, phenyl mono- to tri-substituted by C_1-C_4alkyl, C_1-C_4haloalkyl or by halogen, C_1-C_4alkyl, C_2-C_4alkyl, C_3-C_4alkyl, C_4-C_4alkyl, C_4-C_4alkyl,$

benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; or

 R_6 is COYR₅₀;

Y is oxygen, sulfur, NR₅₁ or NOR₅₄;

R₅₀ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, or C₁-C₄alkyl-COZR₅₂;

Z is oxygen, sulfur, NR₅₃ or NOR₅₅;

R₅₂ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen;

 R_{51} and R_{53} are each independently of the other C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_1 - C_4 alkylcarbonyl, C_1 - C_4 haloalkylcarbonyl, C_1 - C_4 alkylsulfonyl, benzoyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; and

 R_{54} and R_{55} are each independently of the other C_1 - C_4 alkyl; or

 R_6 is C_1 - C_4 alkyl $COZR_{52}$, C_1 - C_4 haloalkyl $COZR_{52}$, C_2 - C_4 alkenyl $COZR_{52}$, C_2 - C_4 alkynyl $COZR_{52}$ or C_2 - C_4 haloalkenyl $COZR_{52}$, or a salt or stereoisomer of a compound of formula I.

3. A compound of formula I according to claim 2, wherein

 R_1 is C_1 - C_4 alkyl;

R₂ is C_1 - C_4 alkyl, C_1 - C_4 haloalkyl, C_3 - or C_4 alkenyl or C_3 - or C_4 alkynyl;

n is 0, 1 or 2;

 R_3 is C_1 - C_4 alkyl, C_1 - C_4 haloalkyl, C_3 - or C_4 -alkenyl or C_3 - or C_4 -alkynyl;

R₄ is hydrogen, fluorine or chlorine;

 R_5 is hydrogen, halogen, methyl, trifluoromethyl, cyano, nitro, amino or C_1 - C_4 haloalkoxy;

R₆ is hydrogen, halogen, cyano, NHR₁₀ or NR₁₀R₁₁;

 R_{10} and R_{11} are each independently of the other C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_1 - C_4 alkylcarbonyl, C_1 - C_4 haloalkylcarbonyl, C_1 - C_4 alkylsulfonyl,

 C_1 - C_4 haloalkylsulfonyl, benzoyl, benzoyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; or

 R_6 is OR_{20} ;

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R₂₀ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, benzyl, pyridyl, pyrimidinyl, pyrazinyl, pyridazinyl, those aromatic and heteroaromatic rings being unsubstituted or mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, or C₁-C₈alkyl-COXR₂₁;

X is oxygen, sulfur or NR₂₂;

R₂₁ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, benzyl mono- to tri-substituted by C₁-C₄alkyl or by halogen; and

R₂₂ is hydrogen, C₁-C₈alkyl or C₃-C₈alkenyl; or

 R_6 is $S(O)_m R_{30}$;

m is 0, 1 or 2;

R₃₀ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, or C₁-C₄alkyl-COVR₃₁;

V is oxygen, sulfur or NR₃₂;

R₃₁ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen; and

R₃₂ is hydrogen, C₁-C₈alkyl or C₃-C₈alkenyl; or

 R_6 is COR_{40} ;

is hydrogen, C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_3 - C_6 cycloalkyl, C_1 - C_4 alkoxy- C_1 - C_4 alkyl, C_1 - C_4 alkylthio- C_1 - C_4 alkyl, phenyl, phenyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; or

 R_6 is $COYR_{50}$;

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- Y is oxygen, sulfur, NR₅₁ or NOR₅₄;
- R₅₀ is hydrogen, C₁-C₈alkyl, C₃-C₈alkenyl, C₃-C₈alkynyl, C₁-C₈haloalkyl, C₃-C₆cycloalkyl, C₁-C₄alkoxy-C₁-C₄alkyl, C₁-C₄alkylthio-C₁-C₄alkyl, phenyl, phenyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, benzyl, benzyl mono- to tri-substituted by C₁-C₄alkyl, C₁-C₄haloalkyl or by halogen, or C₁-C₄alkyl-COZR₅₂;
- Z is oxygen, sulfur, NR₅₃ or NOR₅₅;
- $R_{52} \quad \text{is hydrogen, C_1-C_8alkyl, C_3-C_8alkynyl, C_1-C_8haloalkyl, C_3-C_6cycloalkyl, C_1-C_4alkyl, C_1-C_4alkyl, C_1-C_4alkyl, C_1-C_4alkyl, C_1-C_4alkyl, C_1-C_4alkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1-C_4alkyl, C_1-C_4haloalkyl or by halogen; }$
- R_{51} and R_{53} are each independently of the other C_1 - C_8 alkyl, C_3 - C_8 alkenyl, C_3 - C_8 alkynyl, C_1 - C_8 haloalkyl, C_1 - C_4 alkylcarbonyl, C_1 - C_4 haloalkylcarbonyl, C_1 - C_4 alkylsulfonyl, benzoyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen, benzyl, or benzyl mono- to tri-substituted by C_1 - C_4 alkyl, C_1 - C_4 haloalkyl or by halogen; and

 R_{54} and R_{55} are each independently of the other C_1 - C_4 alkyl; or

 R_6 is C_1 - C_4 alkylCOZR₅₂, C_1 - C_4 haloalkylCOZR₅₂, C_2 - C_4 alkenylCOZR₅₂, C_2 - C_4 alkynylCOZR₅₂ or C_2 - C_4 haloalkenylCOZR₅₂, or a salt or stereoisomer of a compound of formula I.

- 4. A compound according to claim 1, wherein R_5 is chlorine, bromine, methyl, trifluoromethyl or cyano.
- 5. A compound according to claim 1, wherein R_6 is hydrogen, halogen, OR_{20} , $S(O)_mR_{30}$ or $COYR_{50}$.
- 6. A compound according to claim 1, wherein n is 0 or 2.
- 7. A compound according to claim 1, wherein R₁ is methyl.
- 8. A compound according to claim 1, wherein R₂ is methyl.
- 9. A compound according to claim 1, wherein R₃ is methyl or ethyl.

- 10. A compound according to claim 9, wherein R₃ is methyl.
- 11. A compound according to claim 1, wherein R₄ is fluorine.
- 12. A compound according to claim 1, wherein R₄ is hydrogen.
- 13. A compound according to claim 1, wherein R₄ is chlorine.
- 14. A compound according to claim 1, wherein R_4 is chlorine; and R_6 is OR_{20} wherein R_{20} is as defined in claim 1.
- 15. A compound according to claim 1, wherein R_4 is fluorine; and R_6 is OR_{20} wherein R_{20} is as defined in claim 1.
- 16. A compound according to claim 1, wherein R_4 is chlorine; and R_6 is $S(O)_m R_{30}$ wherein R_{30} and m are as defined in claim 1.
- 17. A compound according to claim 1, wherein R_4 is fluorine; and R_6 is $S(O)_m R_{30}$ wherein R_{30} and m are as defined in claim 1.
- 18. A compound according to claim 1, wherein R_4 is chlorine; and R_6 is COR_{40} , $COYR_{50}$, C_1 - C_4 alkyl $COZR_{52}$, C_1 - C_4 haloalkyl $COZR_{52}$, C_2 - C_4 alkenyl $COZR_{52}$, C_2 - C_4 alkynyl- $COZR_{52}$ or C_2 - C_4 haloalkenyl $COZR_{52}$, wherein R_{40} , R_{50} , R_{52} , Y and Z are as defined in claim 1.
- 19. A compound according to claim 1, wherein R_4 is fluorine; and R_6 is COR_{40} , $COYR_{50}$, C_1 - C_4 alkylCOZR₅₂, C_1 - C_4 haloalkylCOZR₅₂, C_2 - C_4 alkenylCOZR₅₂, C_2 - C_4 alkynyl-COZR₅₂ or C_2 - C_4 haloalkenylCOZR₅₂, wherein R_{40} , R_{50} , R_{52} , Y and Z are as defined in claim 1.
- 20. A compound according to claim 1, wherein R₅ is chlorine; and R₆ is -COYR₅₀.
- 21. A compound according to claim 1, wherein R_5 is chlorine; and R_6 is C_1 - C_4 alkyl-B or C_1 - C_4 haloalkyl-B.
- 22. A process for the preparation of a compound of formula I

$$R_{4}$$

$$R_{1}$$

$$S(O)_{n}R_{2}$$

$$R_{3}$$

$$(I),$$

wherein R_1 to R_6 and n are as defined in claim 1, which process comprises cyclising a compound of formula III

wherein R₁, R₂ and R₄ to R₆ are as defined,

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a) with hydrazine optionally in the presence of a suitable solvent to form a compound of formula IIa

$$R_{5}$$
 R_{6}
 R_{4}
 R_{1}
 R_{1}
 $S-R_{2}$
 $N-NH$
(IIa)

and then reacting that compound in the presence of a compound of formula Xa containing a corresponding C_1 - C_4 alkyl, C_1 - C_4 haloalkyl, C_3 - or C_4 -alkenyl or C_3 - or C_4 -alkynyl group

$$R_3-L_1$$
 (Xa),

the radical R_3 in the compounds of formula Xa being as defined in claim 1 and L_1 being a leaving group, to form a compound of formula I

$$R_{5} \xrightarrow{R_{4}} R_{1} \xrightarrow{R_{1}} S(O)_{n} R_{2}$$

$$R_{5} \xrightarrow{R_{6}} R_{3}$$

$$(I)$$

wherein n is 0, and then oxidising that compound; or

b) with a compound of formula XI

$$NH_2-NH-R_3$$
 (XI),

wherein R_3 is as defined, optionally in the presence of a suitable solvent, to form a compound of formula I

$$R_{5} \xrightarrow{R_{4}} R_{1} \xrightarrow{S(O)_{n}} R_{2}$$

$$R_{5} \xrightarrow{R_{6}} R_{6}$$

$$(I),$$

wherein R_1 to R_6 are as defined, and n is 0, and then oxidising that compound.

23. A process for the preparation of a compound of formula II

wherein R_1 , R_2 , R_4 to R_6 and n are as defined in claim 1, which process comprises halogenating a compound of formula IV

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optionally in the presence of a solvent and a base, to form a compound of formula XII

$$R_4$$
 R_4 R_5 R_6 R_1 R_1 R_1 R_2 R_3 R_4 R_5 R_6 R_1 R_1 R_2 R_3 R_4 R_5 R_5 R_6 R_1 R_2 R_3 R_4 R_5 R_5

 R_1 and R_4 to R_6 in the compounds of formulae IV and XII being as defined and Hal being halogen, and cyclising that compound of formula XII with a compound of formula XIII

$$H_2N-NH-C(S)S-R_2$$
 (XIII),

wherein R_2 is as defined, optionally in the presence of a solvent and a base, to form a compound of formula XIV

$$R_4$$
 R_1 S SR_2 $N-N$ SR_2 R_6

and then subjecting that compound to a ring contraction (n=0) thermally or by acid catalysis, and then oxidising that compound (n=1 or 2).

24. A compound of formula II

$$R_{5} \xrightarrow{R_{4}} R_{1} \xrightarrow{S(O)_{n} R_{2}}$$

$$N \xrightarrow{NH} S(O)_{n} R_{2}$$

$$(II)$$

wherein R₁, R₂, R₄ to R₆ and n are as defined in claim 1.

25. A compound of formula XV

$$R_5$$
 R_4
 R_1
 R_3
 R_4
 R_1
 R_3
 R_4
 R_4
 R_1
 R_3
 R_4
 R_4
 R_4
 R_4
 R_4
 R_5
 R_6

wherein R₁ and R₃ to R₆ are as defined in claim 1.

- 26. A herbicidal and plant growth-inhibiting composition comprising one or more compounds of formula I according to claim 1.
- 27. A composition according to claim 26, comprising from 0.1 to 95 % of a compound of formula I according to claim 1.
- 28. A method of controlling undesired plant growth, which comprises applying an effective amount of a compound of formula I, according to claim 1, or of a composition comprising such a compound, to the plants or to the locus thereof.
- 29. A method according to claim 28, which comprises the application of a compound of formula I in an amount of from 0.001 to 2 kg per hectare.
- 30. A method of inhibiting plant growth, which comprises applying an effective amount of a compound of formula I, according to claim 1, or of a composition comprising such a compound, to the plants or to the locus thereof.

31. The use of a composition according to claim 26 in the selective control of weeds in crops of useful plants.

INTERNATIONAL SEARCH REPORT

Inter and Application No PCT/EP 96/02417

		PCI	/EP 96/0241/	
A. CLASS IPC 6	SIFICATION OF SUBJECT MATTER C07D231/18 A01N43/56 C07D40	5/12		
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B. FIELD	S SEARCHED			
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later tri	an the priority date claimed actual completion of the international search	** document member of the same patent family Date of mailing of the international search report		
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NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016		Fink, D		

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